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Brazing of Si₃N₄ to Metals with Al Filler (Report I)†

- Si₃ N₄ /Si₃ N₄ and Si₃ N₄ /Ti or Nb Joints -

Masaaki NAKA*, Masao KUBO** and Ikuo OKAMOTO***

Abstract

The wettability of molten aluminum on Si_3N_4 was investigated by a sessile drop technique in a vacuum. The joining of Si_3N_4 to Si_3N_4 , and Si_3N_4 to Si_3N_4 t

The work of adhesion of aluminum on Si_3N_4 increases significantly with increasing temperature, and aluminum definitly wets Si_3N_4 . The joining strength of Si_3N_4/Si_3N_4 joint with Al filler exhibits the maximum of 210MPa at elevated temperature of 373K, and decreases with further increasing testing temperature. The strength of Si_3N_4/Ti or Nb joint with Al filler is about half the strength of Si_3N_4/Si_3N_4 joint. The stress arisen from the thermal difference between Si_3N_4 and metal provides the detrimental effect on the strength of the Si_3N_4 /metal joint. At the joining layer of Si_3N_4/Ti or Nb joint the intermediate phase of $TiAl_3$ or $NbAl_3$ was formed.

KEY WORDS: (Ceamic-Metal Joining) (Joining) (Brazing) (Ceramics) (Silicon Nitride) (Aluminum) (Filler Metal) (Titanium) (Niobium)

1. Introduction

Brazing is extensively used in the fabrication of ceramic/metal joints by using filler metals. Several filler metals for joining ceramics to metals have to contain the reactive elements such as titanium.

On the other hand, McDonald and Eberhart¹⁾ have reported that the work of adhesion of molten metal on alumina is related to the free energy of oxide formation, and the high reactivity of molten aluminum against alumina. Naka et al.²⁾ have also indicated that aluminum among aluminum, copper and silver well wets silicon nitride, and the wettability of aluminum on Si₃N₄ increases markedly with increasing temperature. These results suggest the utility of aluminum as the brazing filler metal for silicon nitride.

Aluminum possesses other superior property; aluminum is a soft metal which relaxes the stress arisen from the difference of thermal expansion between ceramic and metal during joining.

In this paper the wettability of aluminum was evaluated by measuring the contact angle on Si_3N_4 in a vacuum, and the brazing of Si_3N_4 to metals such as Nb and Ti was performed using Al filler metal. Furthermore, the joining mechanism was investigated by measuring the joining strength of Si_3N_4/Si_3N_4 and $Si_3N_4/metal$ joints, and observing the microstructure of the joint.

2. Experimental

The pressureless sintered Si_3N_4 containing a few percent of Al_2O_3 and Y_2O_3 was used. The purity of aluminum was 99.99 mass%. The metals were Ti, Nb, Cu and Ag in a high purity. The wettability of molten aluminum, copper and silver to Si_3N_4 was evaluated by measuring the contact angle between the peripherral surface of sessile drop and horizontal surface of Si_3N_4 in a vacuum below 1.33 mPa as shown in Fig. 1. The heating rate up to desired temperature was $1.7 \, \text{K/s}$, and the sessile drop apparatus is shown in Fig. 2.

 Si_3N_4 of 15 mm in diameter and 3 mm in thickness, and Nb (or Ti) of 6 mm in diameter and 3 mm in thickness was used for a lap joint. First, the metallizing of

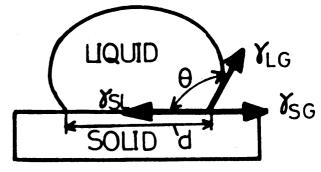


Fig. 1 Schematic of contact angle of liquid drop on Si₃ N₄.

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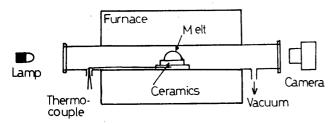


Fig. 2 Sessile drop apparatus.

 $\rm Si_3\,N_4$ with aluminum was conducted at 1373 K for 3.6 ks in 1.33 mPa, and then the lap joint with Al thinned down to 0.1 mm thickness was made in the joining condition of joining temperatures of 1273 K (for Nb) and 1073 K (for Ti) and brazing time of 300s under a load of 10 g.

The joining strength of the lap joint was determined under shear fracture loading at a cross head speed of 1.7 $\times~10^{-2}\,\text{mm\,s}^{-1}$. The microstructure of $\text{Si}_3\text{N}_4/\text{metal}$ was investigated by means of scanning electron microscope and EDX microanalyser.

3. Results and Discussion

3.1 Wettability of molten aluminum on Si₃ N₄

Figure 3 shows the temperature dependence of the equilibrium contact angle of molten aluminum, copper and silver on Si_3N_4 . The contact angle of molten metal was taken at 3.6ks as equilibrium value since the angle

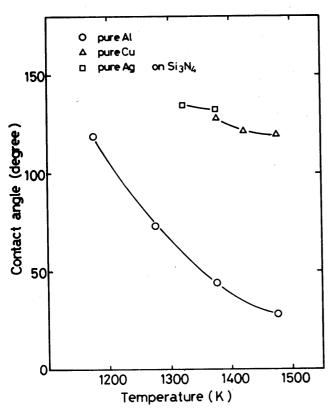


Fig. 3 Temperature dependence of equilibrium contact angle of Al, Cu and Ag on Si₃ N₄.

reached the final contact value. The angle was the lowest for aluminum with the highest temperature dependence, indicating the best wettability among the three metals.

In general, at a contact angle between 0 and $\pi/2$, the shape of the melt is convex and is said to be wetting at contact angles between $\pi/2$ and π the melt begins to swell and is said to be in the non-wetting state³⁾. The wetting of aluminum on Si_3N_4 begins at temperatures above 1240K. The angles of silver are slightly higher than that of copper in Fig. 3.

The work of adhesion, Wad, often becomes a measure of adhesion intensity, Wad is the work required to separate a unit area of solid-liquid interface into two faces and is defined by Young-Duprè equation,

$$W_{ad} = \gamma_{LG} (1 + \cos \theta_{\infty})$$

where γ_{LG} is the liquid energy. Using the data and their temperature coefficient $\alpha = d\gamma_{LG}/dT$ available from the literature⁵⁾ and the measured θ_{∞} for three metals, the work of adhesion, W_{ad} , of the three metals against Si_3N_4 calculated at various temperatures are shown in Fig. 4. Compared with copper and silver, aluminum greatly increases the work of adhesion, Silver shows a lower value than that of copper.

The work of adhesion, W_{ad} of metals against $Al_2O_3^{(1)}$ or cubic $ZrO_2^{(4)}$ is related to the formation energy of

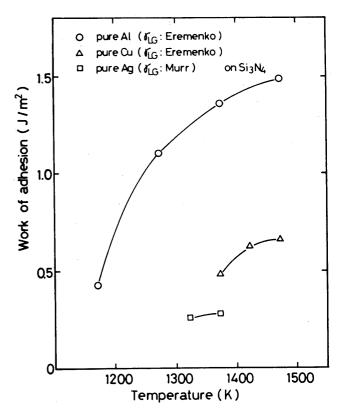


Fig. 4 Temperature dependence of work of adhesion, Wad, of Al, Cu and Ag on Si₃ N₄.

oxide, and the wetting of oxide ceramics with metals is controlled by the ease with which oxidation of the metals occurs. On the other hand, the wetting of silicon nitride ceramics may be related to the ease of nitride formation of the metals.

In order to identify the intermediate phase at the interface of Al/Si_3N_4 joint, the microstructure of the sessile drop made at 1473K for 3.6ks was observed using a scanning electron microscope and a transmission electron microscope. Figure 5 shows the interface of Al/Si_3N_4 sessile drop, and the intermediate phase between alumi-

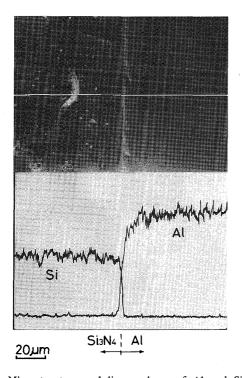


Fig. 5 Microstructure and line analyses of Al and Si in Al/Si₃ N₄ sessile drop cooled down from 1473 K.

num and Si_3N_4 is not clear and very thin. Figure 6 shows the microstructure of the interface using a transmission electron microscope. The intermediate phase is formed at the interface of aluminum and Si_3N_4 . The intermediate phase is composed of fine grains in diameter of 10 nm and grows preferentially in the grain boundaries of Si_3N_4 . The electron and X-ray diffraction patterns of the intermediate phase are identified as R15-A1N type sialon. In a part of the interface the amorphous phase of Al_2O_3 was also observed. The results of Figs. 5 to 6 indicate that aluminum definitely wets Si_3N_4 and is applicable to the brazing filler for Si_3N_4 .

3.2 Joining strength of Si₃N₄/Si₃N₄ and Si₃N₄/Ti or Nb joints

Figure 7 represents the testing temperature dependence of the strength of $\mathrm{Si}_3\mathrm{N}_4/\mathrm{Si}_3\mathrm{N}_4$ joints brazed at $1373\mathrm{K}$ for 3.6ks using Al filler. The strength of $\mathrm{Si}_3\mathrm{N}_4/\mathrm{Si}_3\mathrm{N}_4$ joint shows the maximum value of $210\mathrm{MPa}$ at $373\mathrm{K}$. The release of stress arising from the difference between the thermal expansion of $\mathrm{Si}_3\mathrm{N}_4$ and fillers in the joint accounts for the increase in strength of the $\mathrm{Si}_3\mathrm{N}_4$ joint at the testing temperature. Furthermore, the strength of the $\mathrm{Si}_3\mathrm{N}_4/\mathrm{Si}_3\mathrm{N}_4$ joint falls with increasing testing temperature.

The change in the fracture structures of $\mathrm{Si_3N_4/Si_3N_4}$ joints brazed at 1373K for 3.6ks using aluminum with testing temperature is shown in Fig. 8. The fracture mixed with the brittle fracture structure of $\mathrm{Si_3N_4}$ and ductile fracture structure of the filler below 373K, and the general ductile fracture structure of the filler above 573K are shown in Fig. 8. The decrease in strength of aluminum filler accounts for the decrease in strength of the $\mathrm{Si_3N_4}$ joint.

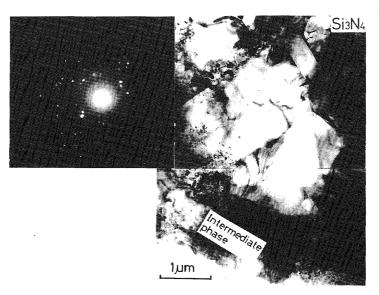


Fig. 6 Microstructure of Al/Si₃ N₄ drop.

The $\mathrm{Si_3}\,N_4/Nb$ or Ti joint is made using Al filler. $\mathrm{Si_3}\,N_4$ was metallized with Al filler at 1373K for 3.6ks. The metallized Al was thinned down to 0.1 mm in thickness, and then the $\mathrm{Si_3}\,N_4$ brazed with Nb or Ti by the metallized Al filler.

Figure 7 also shows the change in fracture strength of $\mathrm{Si}_3\,N_4/\mathrm{Nb}$ or Ti joint with testing temperature. The strength, 100 MPa, of $\mathrm{Si}_3\,N_4/\mathrm{Ti}$ or Nb joint is half the strength of $\mathrm{Si}_3\,N_4/\mathrm{Si}_3\,N_4$ joint at room temperature. The

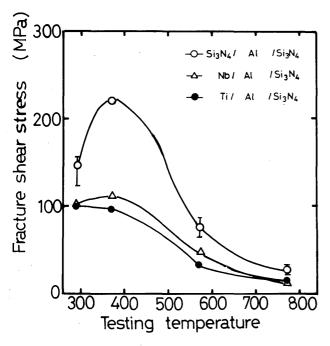


Fig. 7 Testing temperature dependence of fracture shear strength of $Si_3 N_4/Si_3 N_4$ joint brazed at 1373 K for 3.6 ks using Al filler.

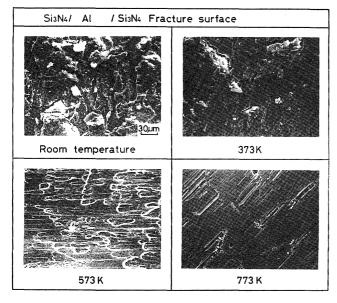


Fig. 8 Change in fracture structure of Si_3N_4/Si_3N_4 joint brazed at 1373 K for 3.6 ks using Al filler with testing temperature.

stress arisen from the difference of thermal expansion coefficient between $\mathrm{Si}_3\,\mathrm{N}_4$ and metal is attributable to the decrease in the strength of the $\mathrm{Si}_3\,\mathrm{N}_4$ joint. The strength of $\mathrm{Si}_3\,\mathrm{N}_4/\mathrm{Nb}$ joint is a little higher than that of $\mathrm{Si}_3\,\mathrm{N}_4/\mathrm{Ti}$ joint, and the strength of both joints decreases with increasing testing temperature. As shown in the change in fracture structures of $\mathrm{Si}_3\,\mathrm{N}_4/\mathrm{Nb}$ or Ti joints with testing temperature in Figs. 9 and 10, the joints represent the ductile fracture surfaces of Al fillers at all testing temperatures. Teh decrease in the strength of Al filler is attributable to the decrease in the strength of $\mathrm{Si}_3\,\mathrm{N}_4/\mathrm{metal}$ joint.

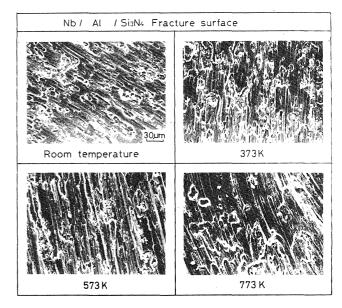


Fig. 9 Change in fracture structure of Si₃ N₄/Ti joint using Al filler with testing temperature.

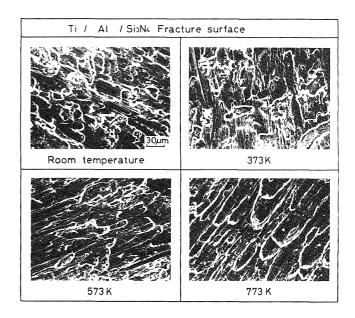


Fig. 10 Change in fracture structure of Si₃ N₄/Nb joint using Al filler with testing temperature.

3.3 Joining microstructure of Si₃ N₄/Ti or Nb joint

The microstructural analyses were performed using energy dispersive X-ray (EDX) microanalysis and scanning electron microscopy.

The microstructure and line analyses for Ti, Al and Si in $Si_3N_4/Al/Ti$ joint brazed with Al filler at 1073 K for 300s are shown in Fig. 11. In the joining layer, the granular intermediate phase $TiAl_3$ is formed as shown in Fig. 12 using EDX microanarises, and further the titanium matrix is melted down to the interface of Si_3N_4 by the dissolution action of aluminum.

Figure 13 shows the microstructure and line analyses

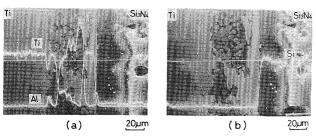


Fig. 11 Microstructure and line analyses for Ti and Al in $Si_3 N_4/Nb$ joint brazed at 1073 K for 300 s.

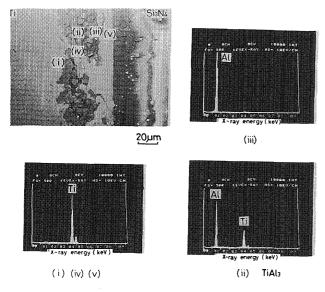


Fig. 12 Microstructure and spot analyses of Si₃ N₄/Ti joint brazed at 1073 K for 300s.

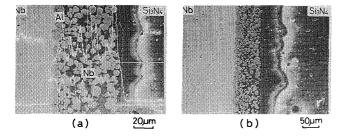
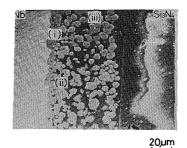


Fig. 13 Microstructure and line analyses for Nb and Al in $Si_3 N_4/Nb$ joint brazed at 1273 K 300 s.



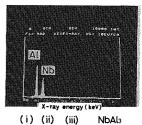


Fig. 14 Microstructure and spot analyses of Si₃ N₄/Nb joint brazed at 1273 K for 300s.

for Nb and Al in $Si_3N_4/Al/Nb$ joint brazed at 1273K for 300s. The granular intermediate phases are identified as NbAl₃ in Fig. 14 using EDX microanalysis. Since Al filler reacts with Nb during brazing of Si_3N_4 to Nb, the thickness of joining layer becomes 60μ m that is thicker than the thickness of 10μ m in Si_3N_4/Si_3N_4 joint. Compared with Si_3N_4/Ti joint the microstructure of Si_3N_4/Nb joint is uniform since the intermediate phase of NbAl₃ distributes homogeneously in Al matrix. This is attributable to the higher strength of Si_3N_4/Nb joint.

4. Conclusion

The wettability of molten aluminum, copper and silver was evaluated by a sessile drop technique. Further, the brazing of $\mathrm{Si}_3\mathrm{N}_4$ to $\mathrm{Si}_3\mathrm{N}_4$, and $\mathrm{Si}_3\mathrm{N}_4$ to Nb or Ti was conducted using Al filler in a vacuum. The joining strength and the microstructure at interface of $\mathrm{Si}_3\mathrm{N}_4$ joints were investigated by shear fracture loading, transmission electron microscopy and EDX microanalysis, respectively.

Aluminum among the three metals shows the lowest equilibrium contact angle with the highest temperature dependence of the angle, and the work of adhesion of aluminum definitely increases with increasing temperature. The formation of AlN sialon at the interface of Si_3N_4 and Al accelarates the wetting of molten aluminum. These results indicate the molten aluminum is applicable for the brazing filler of Si_3N_4 .

The joining strength of Si_3N_4/Si_3N_4 joint brazed with Al filler is 210MP at room temperature, and shows the maximum at elevated temperature. The decrease in strength of Al filler is attributable to the decrease in the strength of Si_3N_4/Si_3N_4 joint.

The joining strength of Si_3N_4/Nb or Ti joint brazed with Al filler is about half the joining strength of Si_3N_4/Si_3N_4 joint.

The stress arisen from the difference of thermal expansion coefficient between Si_3N_4 and metal accounts for the decrease in joining strength of Si_3N_4 /metal joint. The homogeneous distribution of intermediate phase of $NbAl_3$

in Al matrix is attributable to the higher joining strength of $\mathrm{Si}_3\,N_4/Nb$ joint.

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