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Optical Property of Photonic Crystals Infiltrated with Various Liquids and Liquid Crystals

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Abstract: Upon infiltration of various liquids in the opal and its polymer replicas as photonic crystal with a three-dimensional periodic structure of optical wavelength order, the color changed remarkably and a sharp reflection peak and an absorption dip change drastically in wavelength. These shifts are strongly dependent on the sort of infiltrated liquids and can be explained by the difference of refractive index of the liquid. On the contrary from these spectral shifts refractive index of the infiltrated liquid is easily evaluated. Utilizing these spectral shift the refractive index of the liquid and in the case of mixed liquid the concentration of each component of liquid can be evaluated. Theoretical calculation supported this interpretation. Photonic crystals infiltrated with liquid crystals and the effect of applied field are also studied and the mechanism of the novel optical property has been clarified.

INTRODUCTION

Photonic crystals, which have a three-dimensionally ordered structure with a periodicity of optical wavelength, have attracted much attention from both fundamental scientific and practical industrial view points, because novel concepts such as photonic band gap and various new types of applications have been proposed [1-2]. As examples we prepared synthetic opals by the sedimentation of silica spheres dispersed in water and also their polymer inverse opals. These opals have interconnected nano-size voids. We demonstrated that various materials can be infiltrated in these void [3-6]. After infiltration of polymer in these opals and then removing silica by washing with HF, polymer inverse opal can be prepared, which has also interconnected regular array of voids. We have proposed tunable photonic crystal crystals utilizing opals and inverse opals infiltrated with various functional materials and demonstrated several examples of tunable photonic crystals [7-10]. In this paper, effects of infiltration of various liquids and liquid crystals in opals and inverse opals are studied and the application of these effects for the evaluation of refractive index of liquid and also for the sensors of liquids and liquid compositions are proposed.

EXPERIMENTAL

Three-dimensional periodic structures in thin films were fabricated by the sedimentation of mono-dispersed SiO_2 spheres of several hundreds nm in diameter in thin cells. These thin three-dimensional periodic structures of SiO_2 (synthetic opals) were heat treated to make them robust. The opal films were confirmed to have a face-centered cubic structure and have interconnected periodic array of voids. The structure was observed by a scanning electron microscope (Hitachi, S-2100). The transmission spectrum and reflection spectrum was measured using a spectrophotometer (Hitachi 330).

RESULTS AND DISCUSSION

Figure 1 shows transmission spectra of an opal and opals infiltrated with various liquids. As evident in this figure, we can observe a sharp dip at some wavelength in the transmission

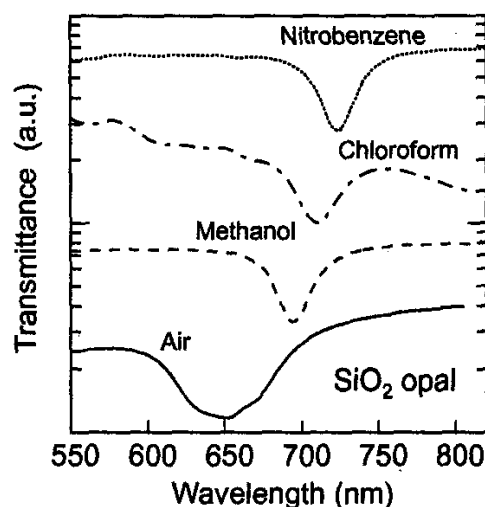


Fig. 1. Transmission spectra of SiO_2 opal infiltrated with various liquids.

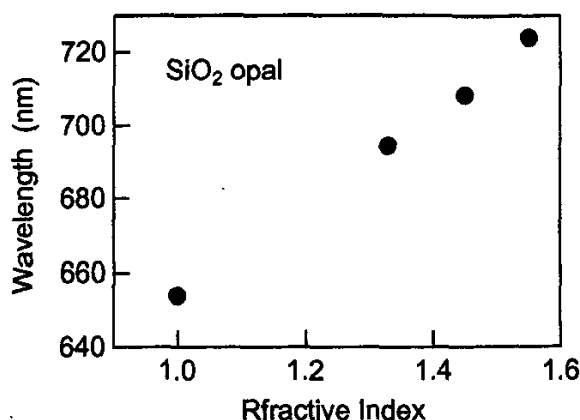


Fig. 2. Transmission dip wavelength of liquid-infiltrated SiO_2 opal as a function of refractive index of liquid.

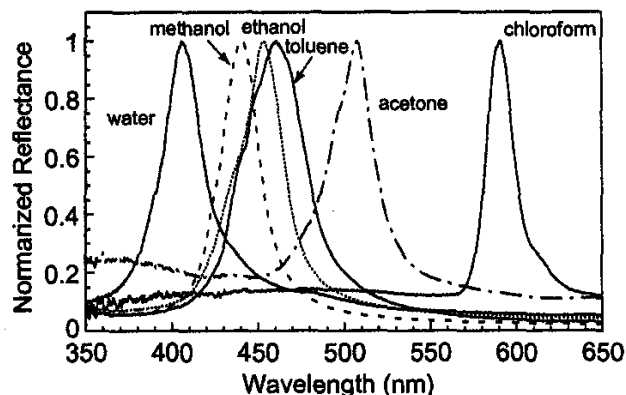


Fig. 3. Reflection spectra of polymer inverse opal infiltrated with various liquids.

spectra, which depends on the sort of liquids. A sharp peak was also observed in a reflection spectrum of the infiltrated opals, which also shifted upon the change of liquids. These spectral shifts of the peak and the dip in the transmission and the reflection spectra can be explained by the difference in the refractive index of the liquids infiltrated in opal.

For normal incidence geometry, Bragg reflection wavelength λ is determined using the following equation,

$$\lambda = 2d\sqrt{fn_s^2 + (1-f)n_v^2} \quad (1)$$

where d is the spacing of the (111) planes, which is normal to

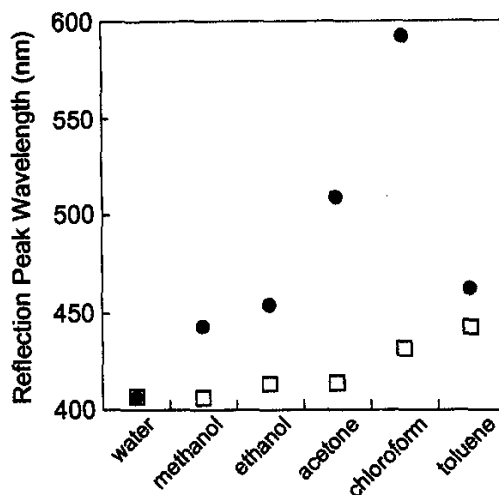


Fig. 4. Reflection peak wavelength of polymer inverse opal infiltrated with various solvents. Closed circles indicate experimental results. Open squares shows wavelength theoretically evaluated assuming constant periodicity.

the light propagation direction, f is the filling factor of the fcc close-packed structure, and n_s and n_v are refractive indices of the SiO_2 spheres and substances in the voids of opal, respectively. In the case of non-infiltrated opal, n_v is the refractive index of air. When liquids is infiltrated in opal, n_v is for liquids. In this case, $f = 0.74$, $n_s = 1.46$.

Figure 2 shows the dependence of the dip wavelengths on the refractive index of the solvent. The results are well coincident with the theoretical dependence. That is, the dip wavelength shifts as following the equation (1).

Similar spectral shifts were also observed upon infiltration of liquids in polymer inverse opal as shown in Fig.3. This shift can be explained also by the difference of refractive index of the liquids. However, the shift in the infiltrated polymer inverse opals was much bigger than the case of infiltrated opals, which can be explained in terms of the much larger volume of voids and therefore much larger occupation ratio of liquids in polymer inverse opals compared with the case of opals. In a case of the inverse opal with gel polymer, the shift of the peak and dip was much larger than the value theoretically evaluated utilizing refractive index of the liquids as shown in Fig.4. This anomaly can be explained by the gel characteristics, that is, by the change in volume, therefore the change in periodicity with the liquids. Namely, in this case, changes in not only refractive index but also periodicity must be taken into consideration. It should also be mentioned that graphite inverse opals also exhibited drastic color change with changing solvent.

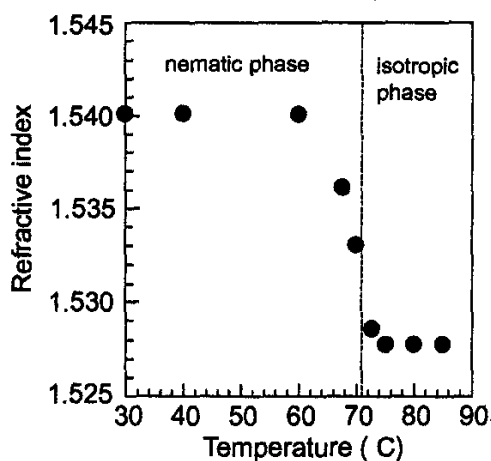


Fig. 5. Temperature dependence of reflection peak wavelength of 5CB-infiltrated opal.

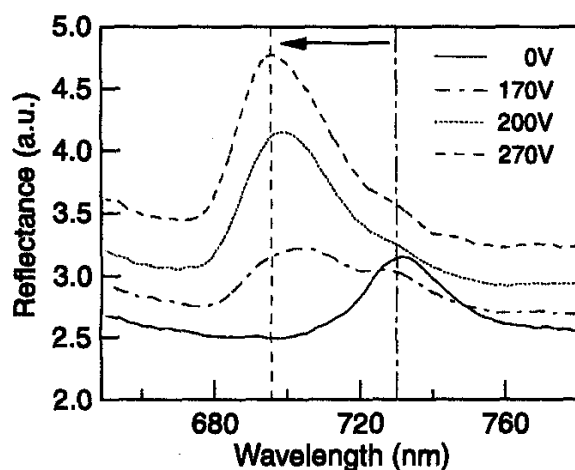


Fig. 6. Reflection spectra of 5CB-infiltrated polymer inverse opal as a function of applied voltage.

Photonic crystals infiltrated with various liquid crystals such as nematic and smectic liquid crystals have also been explored. The reflection peak of the opal infiltrated with liquid crystal shifts with the temperature, especially at the phase transition points, which can be explained also by the change of refractive index with temperature, especially step wise change at the phase transition, as shown in Fig.5. This result is also consistent with the theoretical analysis of photonic crystals infiltrated with liquid crystals.

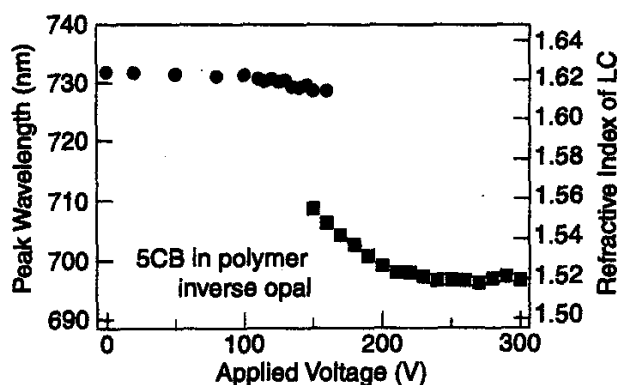


Fig. 7. Voltage dependence of peak wavelength of reflection spectrum of 5CB-infiltrated polymer inverse opal. Refractive index of liquid crystal evaluated from the peak wavelength of reflection spectrum is also shown.

The peak in the reflection spectrum in photonic crystals infiltrated with liquid crystals shifted with applying voltage. Figure 6 shows the reflection spectra of 5CB-infiltrated polymer inverse opal as a function of applied voltage. 5CB (4-pentyl-4'-cyanobiphenyl) is a quite common nematic liquid crystal. As evident from the figure, the reflection peak shifts toward shorter wavelength upon applying voltage. The dependence of the peak wavelength on the applied voltage is more clearly indicated in Fig.7. This shift can be explained by the change of refractive index accompanying with the reorientation orientation of nematic liquid crystals with the applied voltage. The existence of the threshold and the hysteresis due to the anchoring force between the wall of voids and the liquid crystal molecules. The response speed of this voltage induced spectral shift was much faster than the same liquid crystal in a conventional liquid crystal cell.

SUMMARY

We prepared synthetic opals by the sedimentation of silica spheres dispersed in water, and also fabricated polymer inverse opals from the silica opal. Upon infiltration of various liquids in the opal and inverse opal, a sharp peak in the reflection spectrum and a dip in the transmission spectrum shifted drastically. The wavelength shift strongly depended on the sort of infiltrated liquids. For the polymer inverse opal, larger wavelength shift compared with that theoretically predicted was confirmed, which was interpreted in terms of the volume change in the polymer gel matrix. The optical stop band in the transmission and reflection spectra in the opal and inverse opal infiltrated with liquid crystal could be tuned upon the temperature change or upon the electric field application.

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