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Doctoral Dissertation

Studies on the Iridium-Catalyzed Regioselective Silylation of Benzylic C-H Bonds in Azaarenes with Hydrosilanes

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January 2018

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Preface and Acknowledgements

The research presented in this thesis was carried out under the direction of Professor Naoto Chatani of the Department of Applied Chemistry, Faculty of Engineering, Osaka University between April 2012 and March 2018. The thesis is concerned with the development of the iridium-catalyzed direct silylation of $C(sp^3)$ -H bonds at the α -position in alkylazaarenes.

This thesis would not have been possible without help, advice, and support from many people and I would like to express my sincerest appreciation to all of them.

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General Introduction

Organosilicon compounds are one of the most important synthetic building blocks for a variety of chemical transformations¹ including Tamao-Fleming oxidation² and Hiyama cross coupling reactions.³ The conventional method for C-Si bond formation involves the reaction of an organolithium or an organomagnesium reagent with a silyl halide (eq. 1 in Scheme 1). However, this approach suffers from the poor functional group tolerance owing to the use of highly reactive organometallic reagents and silicon electrophiles and the fact that substoichiometric amounts of metal salts are produced as wastes. Therefore, the transition metal-catalyzed cross-coupling reaction of C-X bonds with hydrosilanes was developed in an attempt to synthesize organosilicon compounds under mild reaction conditions with good functional group compatibility (eq. 2).⁴ The catalytic direct silylation of C-H bonds, which are present ubiquitously in organic molecules, has recently emerged as one of the most useful and powerful methods in terms of atom- and stepeconomy (eq. 3).^{4,5} Whereas a large number of the direct silylation of C(sp²)-H bonds have appeared,⁶ the corresponding reaction of C(sp³)-H bonds has not been extensively developed. Especially, C(sp³)-H/H-Si coupling reaction has been much less reported, compared to the intramolecular reaction to afford 5- or 6-membered silacycles.⁷

Scheme 1. Methods for C-Si Bond Formation

$$C-M + X-Si \xrightarrow{M = Li, MgX} C-Si + M-X$$
 (1)

$$C-X + H-Si \xrightarrow{\text{cat. metal}} C-Si + H-X$$
 (2)

$$C-H + H-Si \xrightarrow{\text{cat. metal}} C-Si + H-H$$
 (3)

The chelation-assisted approach has been recognized as one of the more useful methodologies for the direct functionalization of C-H bonds,⁸ and was also applied to the intermolecular $C(sp^3)$ -H bond silylation.⁹ The $C(sp^3)$ -H bond is silylated regioselectively, due to guidance by a directing group, such as a pyridine ring.¹⁰ However, chelation-assisted silylation was restricted to the C-H bonds at the β -position (eq. 4 in Scheme 2)^{10b-f} and at the γ -position (eq. 5),^{10a} which involve the formation of stable 5-membered or 6-membered metallacycles as key intermediates in the catalytic cycle. There have been no reports on the chelation-assisted silylation of $C(sp^3)$ -H bond

of substituents at the α -position (the benzylic position) in azaarenes, which requires the formation of a 4-membered metallacycle as an intermediate.

Scheme 2. Chelation-Assisted $C(sp^3)$ -H Bond Silylation of the Substituent in Azaarenes at the β - and γ -Position

The KO'Bu-catalyzed silylation of C-H bonds was recently reported, 11 which probably proceeded via a radical chain mechanism and/or a mechanism involving a pentacoordinated silicon species as a key intermediate. 12 This catalyst system enabled the C(sp³)-H silylation of 2,6-dimethylpyridines at the α -position. 11b However, the results for the silylation of other azaarenes were not described in the literature.

Scheme 3. KO'Bu-Catalyzed $C(sp^3)$ -H Bond Silylation in 2,6-Dimethylpyridines at the α -Position

HSiEt₃ cat.
$$KO^tBu$$

$$N \qquad \alpha \qquad SiEt_3 \qquad (6)$$

Therefore, the development of new methods for the silylation of $C(sp^3)$ -H bond is highly anticipated. The objective of this research was to develop new types of silylation of $C(sp^3)$ -H bonds on substituents at the α -position in azaarenes via a new reaction mechanism. This thesis is composed of the following three chapters.

Chapter 1 contains a discussion of the Ir₄(CO)₁₂-catalyzed regioselective C(sp³)-H bond silylation of 4-alkylpyridines at the benzylic position with hydrosilanes.

Chapter 2 deals with the $Ir_4(CO)_{12}$ -catalyzed benzylic $C(sp^3)$ -H bond silylation of 2-alkylpyridines with hydrosilanes.

Chapter 3 is concerned with the cationic-pincer iridium-catalyzed C(sp³)-H bond silylation of

2-alkyl-1,3-azoles at the α -position in the alkyl group.

Finally, the findings are summarized in the conclusion section.

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Chapter 1

Iridium-Catalyzed Regioselective C(sp³)-H Silylation of 4-Alkylpyridines at the Benzylic Position with Hydrosilanes Leading to 4-(1-Silylalkyl)pyridines

1.1 Introduction

Pyridines derivatives are among the most versatile structural motifs and are widely utilized in medicinal and material chemistry. Thus, a variety of procedures for their efficient synthesis have been developed to date. Transition metal-catalyzed direct $C(sp^3)$ -H functionalization has also become a powerful tool for the introduction of new functional groups on a substituent, and a number of regioselective functionalizations of $C(sp^3)$ -H bonds on substituents at the 2-position of the pyridine ring has been demonstrated. A variety of catalytic α - $C(sp^3)$ -H bond additions in 2-alkylpyridines to alkenes, alkynes, aldehydes, imines, and N-N double bonds in diethyl azodicarboxylate and catalytic α -C-H bond substitutions by allylic compounds and aryl halides have also been reported. However, only a few reports on the transformation of α - $C(sp^3)$ -H bond in 4-alkylpyridines have been reported. Furthermore, the catalytic α -C-H bond functionalization of polyalkylpyridine derivatives such as 2,4-dimethylpyridine with C4 selectivity is rare.

Chapter 1 describes the regioselective C(sp³)-H silylation of 4-alkylpyridines with hydrosilanes in the presence of the iridium catalyst leading to the production of 4-(1-silylalkyl)pyridines (Scheme 1). As described in the general introduction, chelation-assisted silylation was restricted to the C-H bonds of C2-substituents in pyridines.⁵ This reaction represents the first example of an iridium-catalyzed non-directed C(sp³)-H bond silylation with a hydrosilane.

Scheme 1. Iridium-Catalyzed Regioselective C(sp³)-H Silylation of 4-Alkylpyridines at the Benzylic Position with Hydosilanes

$$R \stackrel{\text{II}}{\longrightarrow} H + HSiR"_3 \xrightarrow{\text{cat. Ir}_4(CO)_{12}} R \stackrel{\text{II}}{\longrightarrow} R$$

1.2 Results and Discussion

Optimization of reaction conditions was initiated by examining the coupling of 4-methylpyridine (1a, 1 mmol) and triethylsilane (2 mmol) using an iridium catalyst as shown in Table 1. To my delight, Ir₄(CO)₁₂ was found to promote the coupling reaction in toluene at 80 °C for 20 h to produce 4-[(triethylsilyl)methyl]pyridine (2a) in 26% yield, along with 40% of 1a being reaovered (entry 1). As is the case with previously reported C-H/H-Si coupling reactions,⁵ the addition of norbornene to the reaction of 1a as a hydrogen scavenger improved the yield of 2a to 85%. 2a was isolated in 79% yield by column chromatography on NH₂-modified silica-gel (entry 2). In this reaction, no C(sp²)-H bond silylation products were detected at the 2- or 3-positions on the pyridine ring.

Table 1. Catalyst Screening^a

entry	catalyst	yield of 2a (%)	remained 1a (%)
1^b	Ir ₄ (CO) ₁₂	26	40
2	$Ir_4(CO)_{12}$	85 (79) ^c	10
3^d	$Ir_4(CO)_{12}$	18	82
4^e	$Ir_4(CO)_{12}$	82 ^f	6
5^g	$Ir_4(CO)_{12}$	71^{h}	5
6^i	$[Ir(OMe)(cod)]_2$	0	88
7^i	[Ir(OMe)(cod)] ₂ /2Me ₄ pher	\mathbf{p}^{j} 0	78
8^i	$[IrCl(cod)]_2$	0	81
9^i	[IrCl(cod)] ₂ /2NaBF ₄	0	81
10^k	$Ir(acac)(CO)_2$	74^{l}	3
11^{i}	[Ir(OMe)(cod)] ₂ /CO (1 atm	n) 59	24
12	$Rh_4(CO)_{12}$	44	48
13^{k}	RhCl(PPh ₃) ₃	43	46
14^{m}	$Ru_3(CO)_{12}$	0	92

^aReaction conditions: **1a** (1 mmol), HSiEt₃ (2 mmol), catalyst (0.025 mmol), and norbornene (2 mmol) in toluene (1 mL) at 80 °C for 20 h under N₂. Yields of **2a** and the remaining **1a** were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^bIn the absence of norbornene. ^cThe number in parenthesis is the isolated yield of **2a**. ^dtert-Butylethylene (2 mmol). ^eAt 100 °C. ^f4-[Bis(triethylsilyl)methyl]pyridine (**2a**) was detected in 4% yield. ^gHSiEt₃ (4 mmol). ^h**2a** was produced in 10% yield. ⁱCatalyst (0.05 mmol). ^jMe₄phen: 3,4,7,8-tetramethylphenanthroline. ^kCatalyst (0.1 mmol). ^l**2a** was produced in 15% yield. ^mCatalyst (0.033 mmol).

On the other hand, tert-butylethylene had no effect on the present C-H silylation of 1a (entry 3). Attempts to achieve the complete consumption of 1a, such as increasing the reaction temperature to 100 °C and increasing the amount of HSiEt₃ to 4 mmol, resulted in the further C-H silylation of 2a to produce 4-[bis(triethylsilyl)methyl]pyridine (2a', entries 4 and 5). In other iridium-containing complexes, addition. some including $[Ir(OMe)(cod)]_2$ [Ir(OMe)(cod)]₂/2Me₄phen, [IrCl(cod)]₂, and [IrCl(cod)]₂/2NaBF₄, showed no catalytic activity (entries 6-9), but the use of Ir(acac)(CO)₂ resulting in the formation of **2a** in 74% (entry 10). Interestingly, the reaction catalyzed by [Ir(OMe)(cod)]₂ proceeded under 1 atmosphere of CO (entry 11). These results indicate that CO is crucial as a ligand for the present iridium-catalyzed reaction to proceed. Among complexes other than iridium that were examined, Rh₄(CO)₁₂ and RhCl(PPh₃)₃ also catalyzed reaction to give 2a in moderate yields (entries 12 and 13). No reaction occurred when Ru₃(CO)₁₂ was used as a catalyst (entry 14).

The results for the reaction of **1a** with a series of hydrosilanes are summarized in Table 2. Tributylsilane also reacted to give **3a** in 65% (entry 2). On the other hand, triisopropylsilane, triphenylsilane, trimethoxysilane, and dimethyl(trimethylsiloxy)silane failed to function as silylation reagents (entries 3-6).

Table 2. Screening of Hydrosilanes^a

+ HSiR₃
$$\frac{\text{cat. Ir}_4(\text{CO})_{12}}{\text{norbornene}}$$
 1a $\frac{\text{cat. Ir}_4(\text{CO})_{12}}{\text{2a-3a}}$

entry	hydrosilane	C-H silylationproduct	yield (%) ^b	remained 1a (%)
1	HSiEt ₃	2a	85 (79)	10
2	HSiBu ₃	3a	65 (65)	22
3	HSi ⁱ Pr ₃	-	0	94
4	HSiPh ₃	=	0	86
5	HSi(OEt) ₃	=	0	98
6	HSi(OSiMe ₃)Me	e ₂ -	0	78

^aReaction conditions: **1a** (1 mmol), hydrosilane (2 mmol), Ir₄(CO)₁₂ (0.025 mmol), norbornene (2 mmol), in toluene (1 mL) at 80 °C for 20 h under N₂. The yield of the remaining **1a** was determined from gas chromatogram with tridecane as an internal standard. The yield of the silylation product was determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^bThe number in parenthesis denotes the isolated yield.

To evaluate the scope of this catalytic reaction, the optimized reaction conditions were examined for a range of 4-alkylpyridines **1b-1o** (Table 3). The C-H silylation of 3,4-dimethylpyridine (**1b**) occurred regioselectively on the methyl group located at the 4-position to afford **2b** in 86% yield. Other alkyl and aryl groups at the 3-position also did not significantly affect product yields (**2c-2f**). Functional groups such as dimethylamino (**2g**), methoxy (**2h**), and chloro (**2i**) groups were compatible with the reaction, but the reaction of 3-bromopyridine (**1j**) afforded a mixture of **2j** and the reduction product **2a** in 61% and 9% yields, respectively. The more sterically hindered 3,4,5-trimethylpyridine (**1k**) was silylated to produce **2k** in 30% yield under the standard reaction conditions, but the product yield was improved to 50% by increasing the reaction temperature to 100 °C.

Table 3. Scope of Substrates^a

$$R = H \text{ (2a)}, \quad 79\% \quad \text{cyclo-C}_5H_9 \text{ (2d)}, \quad 84\% \quad \text{Me}_2N$$

$$R = H \text{ (2a)}, \quad 79\% \quad \text{cyclo-C}_5H_9 \text{ (2d)}, \quad 84\% \quad \text{Me}_2N$$

$$R = H \text{ (2a)}, \quad 79\% \quad \text{cyclo-C}_5H_9 \text{ (2d)}, \quad 74\%^b$$

$$C_4H_9 \text{ (2c)}, \quad 76\% \quad 2-\text{naph (2f)}, \quad 75\%$$

$$2g, \quad 84\% \quad 2g, \quad 84\%$$

$$2h, \quad 73\%^b \quad 2i, \quad 60\%^b \quad 2j, \quad 61\%^b.c \quad 2k, \quad 50\%^b$$

$$2h, \quad 73\%^b \quad 2i, \quad 60\%^b \quad 2j, \quad 61\%^b.c \quad 2k, \quad 50\%^b$$

$$2l, \quad 46\%^b \quad 2m, \quad 55\%^d \quad 2n, \quad 62\%^{d,e} \quad 2o, \quad 12\%^{d,e,f}$$

^aReaction conditions: **1** (1 mmol), HSiEt₃ (2 mmol), Ir₄(CO)₁₂ (0.025 mmol), and norbornene (2 mmol) in toluene (1 mL) at 80 °C for 20 h under N₂. Isolated yields are given. ^bAt 100 °C. ^c**2a** was formed in 9% yield. ^aHSiEt₃ (4 mmol) and norbornene (4 mmol) for 48 h. ^eAt 120 °C. ^fYield of **2o** was determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard.

Pyridines bearing a substituent other than a methyl group at the 4-position such as 4-hexylpyridine (11) also underwent the benzylic C-H silylation to produce 21 in 46% yield. The product 2m was obtained by the reaction of 5,6,7,8-tetrahydroisoquinoline (1m) in 55% yield, after reoptimization of the reaction conditions (Table 4). However, the silylation of 4-isopropylpyridine did not proceed at all under the reaction conditions used.

Table 4. Reoptimization of the Reaction Conditions for 1m^a

entry	HSiEt ₃ (mmol)	norbornene (mmol)	temp.	yield of $2m (\%)^b$	remained 1m (%)
1	2	2	80	34	55
2	2	2	80	42	48
3	4	2	80	50	25
4	4	4	80	61 (61)	23
5	4	4	120	39	36

^aReaction conditions:

1m (1 mmol), $Ir_4(CO)_{12}$ (0.025 mmol), in toluene (1 mL) for 48 h under N_2 . Yields of **2m** and the remaining **1m** were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^bThe number in parenthesis denotes the isolated yield of **2m**.

Finally, I examined some 2-subustituted 4-methylpyridines. Although 4-methylquinoline (1n) reacted to afford 2n in 62% at 120 °C, the reaction of 2,4-dimethylpyridine (1o) resulted in a low yield of the product 2o (12%) along with the recovery of 1o in 65% yield. The substrate was completely recovered in the case of the reaction of 2,4,6-trimethylpyridine. Steric hindrance around the pyridine nitrogen in the substrate retarded the reaction, although it is located some distance from the reaction site. However, the author was pleased to find that some pyridine derivatives with no substituent at the 2-position, such as pyridine, 3-methylpyridine, 3,5-dimethylpyridine, and 3-methyl-5-phenylpyridine, participated in the reaction (entries 2-5 in Table 5). In these cases, the silylation product of the methyl group at the 2-position 2o" and the disilylation product at 2- and 4-positions 2o" were also formed, although in low yields. Furthermore, the regioselective C(sp²)-H bond silylation of the added pyridine and 3-

methylpyridine at the 3-position was found to occur, with 3-triethylsilylpyridine and 3-methyl-5-triethylsilylpyridine, respectively, being formed (entries 2 and 3).⁶ On the other hand, the yield of **20** remained unchanged when 2,6-dimethylpyridine was added to the reaction mixture (entry 6). These results indicate that the nitrogen atom in the added pyridine ring plays an important role for the success of the reaction.

Table 5. Scope of Substrates^a

ontmi	a Aditivo	yields (%)			
entry	additive -	20	2o''	20'''	remained 10 (%)
1	none	10	0	0	78
2^b	pyridine	72	1	2	22
3^c	3-methylpyridine	75	2	3	17
4	3,5-dimethylpyridine	73	2	3	14
5	3-methyl-5-phenylpyridine	75	1	2	19
6	2,6-dimethylpyridine	14	0	0	60
7	phenanthlorine	40	0	0	41
8	Me ₄ phen	15	1	0	73
9	5,5'-dimethyl-2,2'-dipyridine	20	0	0	61
10	Et ₃ N	23	0	0	53
11	DBU	0	0	0	94
12^d	none	0	0	0	67
13^d	3,5-dimethylpyridine	34	1	0	51

"Reaction conditions: **10** (1 mmol), HSiEt₃ (2 mmol), Ir₄(CO)₁₂ (0.025 mmol), norbornene (2 mmol), and additive (0.2 mmol) in toluene (1 mL) at 100 °C for 20 h under N₂. Yields of **20** and the remaining **10** were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^b3-Triethylsilylpyridine was detected in 22% yield. ^c3-Methyl-5-triethylsilylpyridine was detected in 27% yield. ^dIr(acac)(CO)₂ (0.1 mmol) as a catalyst.

Some 2-substituted 4-methypyridine derivatives were next examined under the revised reaction conditions (Table 6). However, the reactivity of these derivatives was affected by the bulkiness of the substituent at the 2-position. 2-Hexyl-4-methylpyridine (1p) exhibited low reactivity to afford 2p in 30% yield, along with the recovery of 1p in 64% yield. Therefore, the reaction in the case of 1p was carried out at 120 °C, and 2p was produced in 59% yield. 4-Methypyridine

substituted with an isopropyl group at the 2-position was also silylated at 160 °C to furnish **2q** in 60% yield. The reaction conditions were also applicable to 2-dimethylamino-4-methylpyridine (**1r**) and 2,4-dimethylquinoline (**1s**) to afford **2r** and **2s** in 43% and 52% yields, respectively, and no silylated product of other methyl groups was detected. Nevertheless, 2,4,6-trimethylpyridine remained intact during the reaction.

Table 6. Scope of 2-Substituted 4-Methylpyridines^a

^aReaction conditions: **1** (1 mmol), HSiEt₃ (4 mmol), Ir₄(CO)₁₂ (0.025 mmol), norbornene (4 mmol), and 3,5-dimethylpyridine (0.2 mmol) in toluene (1 mL) at 120 °C for 48 h under N₂. Isolated yields are given. ^bAt 160 °C. ^cFor 5 days.

To gain some additional insights into the reaction mechanism, deuterium labeling experiments using DSiEt₃ were carried out, and the results are shown in Scheme 2. The results are summarized as follows:

- (1) whereas benzylic C-H bond cleavage occurred at the 2- and 4-positions, cleavage at the 3-position was not confirmed;
- (2) cleavage of the benzylic C-H bond at the 4-position took place more frequently rather than that at the 2-position;
- (3) The presence of pyridine clearly accelerated C-H bond cleavage;
- (4) no C-H bond cleavage was observed in the reaction of 2,4,6-trimethylpyridine even in the presence of 3,5-dimethylpyridine.

This fact indicates that the pyridine nitrogen atom in the substrate also appears to play an important role in this catalytic C-H bond cleavage.

Scheme 2. Deuterium Labelling Experiments^a

"Reaction conditions: **1** (1 mmol), DSiEt₃ (2 mmol), Ir₄(CO)₁₂ (0.025 mmol), norbornene (2 mmol), and additive (0.2 mmol) in toluene (1 mL) under N₂. Yields of **2** and the recovered **1**, and deuterium content were determined from ¹H and ²H NMR spectra. 1,3-Dihydro-2-benzofuran was used as an internal standard to determine the yields. ^bDue to the overlap of two peaks assigned as the hydrogens at the 2- and 6-positions in the pyridine ring, the location where deuterium was incorporated could not be determined. ^cDue to the overlap of two peaks assigned as the methyl groups at the 3- and 4-positions, the location where deuterium was incorporated could not be determined. ^dDeuterium was also incorporated into the recovered 3-methyl-5-phenylpyridine.

The applicability of the present C-H bond silylation to other substrates was investigated. The C-H bond at the 3-position of *N*-methylindole (4) participated in the reaction to afford 5 in 44% yield (Scheme 3).⁷ The presence of 3,5-dimethylpyridine was also crucial for the reaction to

proceed, based on the fact that only a trace amount of **5** was formed when the reaction was run in the absence of 3,5-dimethylpyridine.

Scheme 3. Reaction of *N*-Methylindole 4^a

"Reaction conditions: **4** (0.5 mmol), HSiEt₃ (1 mmol), Ir₄(CO)₁₂ (0.0125 mmol), norbornene (1 mmol), and 3,5-dimethylpyridine (0.1 mmol) in toluene (0.5 mL) at 120 °C for 20 h under N₂. NMR yields were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^bThe number in parenthesis denotes the isolated yield of **5**.

N,*N*-Dimethylaniline (6) could be also used as a substrate under the reaction conditions, with the *para*-silylated compound 7 being produced in 11% yield (Scheme 4). ^{7b,c,e} In light of previous studies on the mechanism of these reactions, it appears that the present iridium-catalyzed reaction involves the formation of an electrophilic silicon species as an intermediate. ⁸

Scheme 4. Reaction of N,N-Dimethylaniline 6^a

"Reaction conditions: **6** (0.5 mmol), HSiEt₃ (2 mmol), Ir₄(CO)₁₂ (0.0125 mmol), norbornene (0.5 mmol), and 3,5-dimethylpyridine (0.1 mmol) in toluene (0.5 mL) at 160 °C for 3 days under N₂. NMR yields were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. ^bThe number in parenthesis denotes the isolated yield of **7**.

A proposed mechanism for the regioselective benzylic C(sp³)-H bond silvlation of 4alkylpyridines is depicted in Scheme 5, although details are unclear at the present stage. An iridium complex reacts with a hydrosilane to form an electrophilic silicon species II, 9 which is an η^1 -silane complex \mathbf{H}_a , or a σ -silane complex \mathbf{H}_b , as a plausible key intermediate. ¹⁰ N-Silylpyridinium iridate III is formed by the reaction of II with the substrate pyridine, through an outer-sphere silicon transfer pathway. The author concludes that the π -acidic nature of CO as a ligand might promote the abstraction of a hydride from the hydrosilane to form III, 11 or, stabilize the anionic character of the hydridoiridate species in III. Proton abstraction at the benzylic position of the N-silylpyridinium species by a hydridoiridate species then takes place to afford a dihydridoiridium complex IV and the silylenamine V. Therefore, the reaction can be regarded as a new type of dehydrogenative silvlation of pyridines. The silvl group on the pyridine nitrogen hampers the abstraction of the benzylic proton at the 2-position because of steric factors. 12 IV reacts with norbornene to give norbornane, along with the regeneration of I. The enamine V undergoes silvlation with either the electrophilic silicon species II or the silvlpyridinium species III to furnish VI, 6c,13 and the subsequent elimination of the silyl group from the nitrogen in VI produces the product. Although the reason why the reaction is accelerated by the addition of 3,5dimethylpyridine is quite unclear, it might participate in the reaction as a ligand or, possibly, as the porter of the silvl group to 2,4-dimethylpyridine after the formation of the N-silvl-3,5dimethylpyridinium species. 14 The latter possibility might explain the reason why 3,5dimethylpyridine is required in the C-H silvlation of N-methylindoles. However, we currently have no direct evidence for shuffling of the silvl group between these pyridine derivatives.

1.3 Conclusion

The author has demonstrated the iridium-catalyzed regioselective C(sp³)-H bond silylation of 4-alkylpyridines with hydrosilanes. The reaction proceeded in the presence of a catalytic amount of Ir₄(CO)₁₂ or Ir(acac)(CO)₂, which possess CO as a ligand, or [Ir(OMe)(cod)]₂ under 1 atmosphere of CO. The low product yields in the reaction of 2-substituted 4-methylpyridines under the optimized reaction conditions could be markedly improved by adding other pyridine derivatives, such as 3,5-dimethylpyridine as additives.

Scheme 5. A Proposed Reaction Mechanism

$$Ir(CO)_{n}$$

$$Ir($$

1.4 Experimental Section

General Information.

 1 H, 2 H, and 13 C NMR spectra were recorded on 400, 61, and 100 MHz spectrometers, respectively, using CDCl₃ and C₆D₆ as solvents. Data are recorded as follows: chemical shifts in ppm (δ), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, c = complex), coupling constant (Hz), and integration. Infrared spectra (IR) were recorded using ATR. Absorption data are reported in reciprocal centimeters with the following relative intensities: s (strong), m (medium), or w (weak). Mass spectra were obtained using a spectrometer with a quadrupole mass analyzer at 70 eV. High-resolution mass spectra (HRMS) were obtained using a spectrometer with a double-focusing mass analyzer. Analytical gas chromatography (GC) was carried out on a chromatograph equipped with a flame ionization detector. Melting points were determined on a capillary point apparatus equipped with a digital thermometer and are uncorrected.

Materials.

Toluene was purified by passage through activated alumina under a positive pressure of N₂. Norbornene and DSiEt₃ were purchased and were used as received. Other organic compounds, except **1c-h**, **1k**, **1l**, and **1p-r**, were purchased and distilled over CaH₂ before use. Compounds **1c-h**, **1k**, and **1p** were prepared as described below. Compounds **1l**, ¹⁵ **1q**, ¹⁶ and **1r**¹⁷ were prepared following procedure described in the literature. All metal complexes were purchased and used without further purification.

Typical Procedure for the Ir₄(CO)₁₂-Catalyzed C-H Silylation of Pyridines with Hydrosilanes.

A 25 mL Schlenk tube was flame-dried and purged with N₂. After cooling to room temperature, Ir₄(CO)₁₂ (27.6 mg, 0.025 mmol), toluene (1 mL), hydrosilane (2 mmol), **1** (1 mmol), and norbornene (188 mg, 2 mmol) were placed in the Schlenk tube. The reaction mixture was stirred at 80 °C for 20 h. After cooling to room temperature, the volatile were removed *in vacuo*. The product was isolated by flash column chromatography on silica-gel.

Typical Procedure for the Deuterium Labelling Experiments using DSiEt3.

The reaction was carried out in the same manner as described above, except that DSiEt₃ (234 mg, 2 mmol) was added to the reaction mixture, in place of HSiEt₃. After the reaction, the reaction mixture was cooled to room temperature and concentrated under reduced pressure (150 mmHg) at 40 °C. The yields of the silylated product and the remaining unreacted starting pyridine were determined from ¹H NMR spectroscopy with 1,3-dihydro-2-benzofuran as the internal standard. They were then separated by flash column chromatography on silica-gel. However, in consideration of the high volatility of the compounds, the complete removal of eluent under reduced pressure was not made. The deuterium content of the compounds was determined from ¹H and ²H NMR spectroscopy.

The Deuterium Labelling Experiment of 1b with DSiEt3.

The reaction was carried out at 80 °C for 8 h. After cooling and evaporation of the reaction mixture, **1b** and **2b** were separated by flash column chromatography on silica-gel (**1b**: $R_f = 0.20$ in pentane/Et₂O = 1/1, **2b**: $R_f = 0.26$ in pentane/Et₂O = 1/1).

Labelling Experiment of 10 with DSiEt3 in the Presence of 3-Methyl-5-phenylpyridine. The reaction was carried out at 100 °C for 8 h. After cooling and evaporation of the reaction mixture, **10** and **20** were separated by flash column chromatography on silica-gel (**10**: $R_f = 0.14$ in pentane/Et₂O = 1/1, **20**: $R_f = 0.21$ in pentane/Et₂O = 1/1). However, **20** was obtained as a mixture with 3-methyl-5-phenylpyridine.

The Deuterium Labelling Experiment of 2,4,6-Trimethylpyridine with DSiEt₃ in the Presence of 3,5-Dimethylpyridine.

The reaction was carried out at 160 °C for 20 h. After cooling and evaporation of the reaction mixture, no production of the silylated product was confirmed from ¹H NMR spectroscopy. Therefore, the yield and deuterium content of 2,4,6-trimethylpyridine were determined from ¹H and ²H NMR spectroscopy of the unpurified reaction mixture.

3-Butyl-4-methylpyridine (1c).¹⁸

The procedure described in the patent¹⁹ was modified by using 3-bromo-4-methylpyridine (3.95 g, 23.0 mmol) and ZnBu₂,²⁰ in place of methyl 5-bromonicotinate and ZnEt₂, respectively, to produce **1c**, which was isolated by flash column chromatography on silica-gel ($R_f = 0.20$ in

hexane/EtOAc = 3/1) and subsequent distillation under reduced pressure (74 °C/8.3 mmHg) in 16% yield (570 mg, 3.68 mmol) as a colorless oil. 1 H NMR (CDCl₃) δ : 0.94 (t, J = 7.5 Hz, 3H), 1.36-1.40 (sextet, J = 7.5 Hz, 2H), 1.49-1.57 (quintet, J = 7.5 Hz, 2H), 2.39 (s, 3H), 2.64 (t, J = 7.5 Hz, 2H), 7.26 (d, J = 5.0 Hz, 1H), 8.44 (d, J = 5.0 Hz, 1H), 8.47 (s, 1H). 13 C NMR (CDCl₃) δ : 13.9, 19.2, 22.7, 30.4, 32.0, 126.8, 139.1, 145.8, 148.2, 150.5.

1-(4-Methylpyridin-3-yl)cyclopentan-1-ol (pre-1d-i).

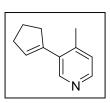
The procedure reported by Vacher²¹ was modified by using ethyl 4-methylnicotinate²² (6.61 g, 40.0 mmol) in place of ethyl 5-bromonicotinate to produce **pre-1d-i**, which was isolated by flash column chromatography on NH₂ modified silica-gel (R_f = 0.23 in hexane/EtOAc = 1/1) in 54% yield (3.86)

g, 21.6 mmol) as a white solid. Mp = 88.7-89.2 °C. ¹H NMR (CDCl₃) δ : 1.78-1.83 (m, 2H), 1.95-1.96 (m, 2H), 2.07-2.21 (m, 4H), 2.58 (s, 3H), 7.09 (d, J = 4.8 Hz, 1H), 8.35 (d, J = 4.8 Hz, 1H), 8.58 (s, 1H). ¹³C NMR (CDCl₃) δ : 21.4, 23.5, 39.6, 82.2, 127.0, 140.0, 146.1, 147.5, 148.0. IR

(ATR): 3131 w, 2952 w, 2871 w, 2806 w. MS, *m/z* (EI, relative intensity, %): 159 (90), 158 (100), 148 (23), 144 (39), 135 (10), 120 (93). Anal. Calcd for C₁₆H₁₃N: C, 74.54; H, 8.53; N, 7.90. Found: C, 74.29; H, 8.53; N, 7.86.

3-(Cyclopent-1-en-1-yl)-4-methylpyridine (pre-1d-ii).

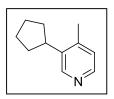
The procedure reported by Vacher²¹ was modified by using **pre-1d-i** (3.52 g, 20.0 mmol) in place of 1-(5-bromo-pyridin-3-yl)cyclopentanol to produce **pre-1d-ii**, which was isolated by distillation under reduced pressure (74 °C/7.5 mmHg) in 63% yield (2.01 g, 12.6 mmol) as a colorless oil. ¹H



NMR (CDCl₃) δ : 1.99-2.06 (quintet, J = 7.4 Hz, 2H), 2.35 (s, 3H), 2.52-2.58 (m, 2H), 2.66-2.71 (m, 2H), 5.86-5.88 (m, 1H), 7.07 (d, J = 4.8 Hz, 1H), 8.31 (d, J = 4.8 Hz, 1H), 8.39 (s, 1H). ¹³C NMR (CDCl₃) δ : 20.6, 23.6, 33.5, 36.2, 125.2, 131.5, 134.0, 140.0, 144.3, 147.5, 148.6. IR (ATR): 3044 w, 3020 w, 2952 w, 2845 w. MS, m/z (EI, relative intensity, %): 159 (M⁺, 98), 158 (100), 144 (69), 143 (21), 131 (29), 130 (23), 117 (18), 115 (11), 77 (15). HRMS Calcd for $C_{11}H_{13}NSi$ (M⁺): 159.1048. Found: 159.1047.

3-Cyclopentyl-4-methylpyridine (1d).

The procedure described in the patent²³ was modified by using **pre-1d-ii** (1.75 g, 11.0 mmol) in place of 5-cyclopentylpyridine-2-carboxylic acid to produce **1d**, which was isolated by distillation under reduced pressure (65 °C/8.2 mmHg) in 89% yield (1.58, 9.79 g) as a colorless oil. ¹H NMR (CDCl₃) δ:



1.61-1.74 (m, 4H), 1.83-1.85 (m, 2H), 2.01-2.08 (m, 2H), 2.33 (s, 3H), 3.11-3.19 (m, 1H), 7.01 (d, J = 4.8 Hz, 1H), 8.27 (d, J = 4.8 Hz, 1H), 8.44 (s, 1H). ¹³C NMR (CDCl₃) δ : 19.2, 25.5, 33.1, 39.9, 124.9, 139.8, 144.9, 146.8, 147.2. IR (ATR): 2952 w, 2868 w. MS, m/z (EI, relative intensity, %): 161 (M⁺, 72), 160 (22), 146 (17), 133 (37), 132 (100), 120 (52), 119 (74), 118. (47), 117 (45), 106 (20), 91 (16), 77 (12). Anal. Calcd for C₁₁H₁₅N: C, 81.94; H, 9.38; N, 8.69. Found: C, 81.87; H, 9.34; N, 8.67.

4-Methyl-3-phenylpyridine (1e).²⁴

The procedure reported by Horenstein²⁵ was modified by using 3-bromo-4-methylpyridine (3.62 g, 21.1 mmol) in place of 3,5-dibromopyridine to produce **1e**, which was isolated by column chromatography on silica-gel (R_f

= 0.08 in hexane/EtOAc = 6/1) and subsequent distillation under reduced pressure (100 °C/3.0 mmHg) in 69% (2.23 g, 14.6 g) as a colorless oil. 1 H NMR (CDCl₃) δ : 2.29 (s, 3H), 7.19 (d, J = 5.0 Hz, 1H), 7.31-7.33 (m, 2H), 7.38-7.48 (m, 3H), 8.44-8.45 (m, 2H). 13 C NMR (CDCl₃) δ : 19.8, 125.2 127.6, 128.4, 129.3, 137.7, 137.9, 144.5, 148.3, 150.0.

4-Methyl-3-(naphthalen-2-yl)pyridine (1f).²⁶

The procedure reported by Horenstein²⁵ was modified by using 3-bromo-4-methylpyridine (3.62 g, 21.1 mmol) in place of 3,5-dibromopyridine to produce **1f**, which was isolated by column chromatography on silica gel ($R_f = 0.19$ in hexane/EtOAc = 3/1) and

subsequent recrystallization from Hexane/EtOAc in 79% yield (3.28 g, 16.7 mmol) as a white solid. Mp = 60.9-61.4 °C. ¹H NMR (CDCl₃) δ : 2.34 (s, 3H), 7.23 (d, J = 5.0 Hz, 1H), 7.45 (d, J = 8.2 Hz, 1H), 7.53-7.55 (m, 2H), 7.79 (s, 1H), 7.89-7.92 (m, 3H), 8.49 (d, J = 5.0 Hz, 1H), 8.53 (s, 1H). ¹³C NMR (CDCl₃) δ : 20.0, 125.3, 126.4, 126.6, 127.4, 127.8, 128.09, 128.12, 128.3, 132.7, 133.3, 135.5, 137.8, 144.8, 148.5, 150.2.

N,*N*,4-Trimethylpyridin-3-amine (1g).

The procedure reported by Aron²⁷ was modified by using 4-methylpyridin-3-amine (5.00 g, 46.2 mmol) in place of pyridin-3-amine to produce **1g**, which was isolated by distillation under reduced pressure (55 °C/3.0 mmHg) in 96% yield (4.79 g, 44.4 mmol) as a colorless oil. 1 H NMR (CDCl₃) δ : 2.31 (s, 3H),

2.76 (s, 6H), 7.04 (d, J = 4.6 Hz, 1H), 8.15 (d, J = 4.6 Hz, 1H), 8.26 (s, 1H). ¹³C NMR (CDCl₃) δ : 18.1, 43.9, 125.8, 140.6, 140.8, 143.8, 148.6. IR (ATR): 3038 w, 2980 w, 2943 w, 2865 w, 2832 w, 2784 w. MS, m/z (EI, relative intensity, %): 136 (M⁺, 100), 135 (98), 121 (27), 119 (24), 105 (18), 94 (15), 93 (14), 92 (26), 65 (19), 53 (10). HRMS Calcd for C₈H₁₂N₂ (M⁺): 137.1079. Found: 137.1077.

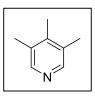
3-Methoxy-4-methylpyridine (1h).²⁸

The procedure reported by Buchwald²⁹ was modified by using 3-bromo-4-methylpyridine (2.58 g, 15.0 mmol) in place of 2-chloro-5-iodopyridine to produce **1h**, which was isolated by column chromatography on silica gel (R_f = 0.20 in pentane/Et₂O = 2/1) and subsequent distillation under reduced

pressure (42 °C/8.3 mmHg) in 18% yield (334 mg, 2.85 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 2.23 (s, 3H), 3.92 (s, 3H), 7.06 (d, J = 4.6 Hz, 1H), 8.12 (d, J = 4.6 Hz, 1H), 8.17 (s, 1H). ¹³C NMR (CDCl₃) δ : 15.7, 55.9, 125.4, 132.4, 135.5, 142.5, 154.4.

3,4,5-Trimethylpyridine (1k).³⁰

The procedure reported by Naveschuk³¹ was modified by using 3,5-dibromo-4-methylpyridine (5.00 g, 20.0 mmol) in place of ethyl 3,5-dibromoisonicotinate to produce **1k**, which was isolated by column chromatography on silica gel (R_f = 0.29 in Et₂O) and subsequent sublimation (38 °C/6.8 mmHg) in 6% (144 mg,



1.2 mmol) as a white solid. Mp = 33.1-33.6 °C. 1 H NMR (CDCl₃) δ : 2.18 (s, 3H), 2.25 (s, 6H), 8.19 (s, 2H). 13 C NMR (CDCl₃) δ : 15.0, 16.9, 131.4, 144.1, 148.2.

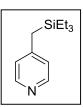
2-Hexyl-4-methylpyridine (1p).³²

The procedure with 2-bromo-4-methylpyridine (3.92 g, 22.8 mmol) as a starting material, reported by Delort, ³³ was modified by using 3.0 M hexylmagnesium bromide (11.5 mL, 34.8 mmol) in place of pentylmagnesium bromide to produce **1p**, which was isolated by

column chromatography on silica gel ($R_f = 0.23$ in hexane/EtOAc = 8/1) and subsequent distillation under reduced pressure (72 °C/9.5 mmHg) in 26% yield (1.09 g, 9.00 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.88 (t, J = 7.1 Hz, 3H), 1.28-1.37 (m, 6H), 1.68-1.71 (m, 2H), 2.31 (s, 3H), 2.73 (t, J = 7.8 Hz, 2H), 6.91 (d, J = 5.0 Hz, 1H), 6.96 (s, 1H), 8.36 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 14.1, 21.1, 22.6, 29.2, 30.0, 31.8, 38.2, 122.03, 123.7, 147.7, 148.6, 162.2.

4-[(Triethylsilyl)methyl]pyridine (2a).

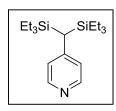
2a was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.17 in hexane) in 79% yield as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ : 0.52 (q, J = 7.9 Hz, 6H), 0.93 (t, J = 7.9 Hz, 9H), 2.11 (s, 2H), 6.94 (dd, J = 4.5, 1.7 Hz, 2H), 8.38 (dd, J = 4.5, 1.7 Hz, 2H). ¹³C NMR (CDCl₃) δ : 2.9, 7.3, 22.2,



123.7, 149.4, 150.7. IR (ATR): 2952 w, 2909 w, 2875 w, MS, *m/z* (EI, relative intensity, %): 207 (M⁺, 38), 150 (11), 122 (24), 120 (11), 115 (30), 87 (100), 59 (32). HRMS Calcd for C₁₂H₂₁NSi (M⁺): 207.1443. Found: 207.1444.

4-[Bis(triethylsilyl)methyl]pyridine (2a').

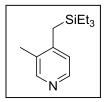
2a' was produced by the present $Ir_4(CO)_{12}$ -catalyzed reaction of **2a** (207 mg, 1.00 mmol) with $HSiEt_3$ (640 μ L, 4.00 mmol) in the presence of norbornene (376 mg, 4.00 mmol) at 160 °C for 48 h. The product was isolated by flash column chromatography on silica-gel (R_f = 0.11 in hexane/EtOAc = 5/1) in



14% yield (44.0 mg) as a pale yellow oil. ¹H NMR (CDCl₃) δ : 0.58 (q, J = 7.9 Hz, 12H), 0.90 (t, J = 7.9 Hz, 18H), 1.82 (s, 1H), 6.91 (d, J = 6.0 Hz, 2H), 8.33 (d, J = 6.0 Hz, 2H). ¹³C NMR (CDCl₃) δ : 4.8, 7.9, 23.4, 124.8, 149.1, 153.2. IR (ATR): 2952 w, 2910 w, 2876 w. MS, m/z (EI, relative intensity, %): 293 (18), 292 (31), 266 (12), 265 (100), 264 (58), 237 (42), 236 (38), 220 (53), 208 (14), 115 (21), 87 (60), 59 (48). HRMS Calcd for C₁₈H₃₅NSi₂ (M⁺): 321.2308. Found: 321.2303.

3-Methyl-4-[(triethylsilyl)methyl]pyridine (2b).

2b was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.10 in hexane) in 86% yield as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ : 0.54 (q, J = 8.0 Hz, 6H), 0.91 (t, J = 8.0 Hz, 9H), 2.11 (s, 2H), 2.22 (s, 3H), 6.88 (d, J = 5.0 Hz, 1H), 8.21 (d, J = 5.0 Hz, 1H), 8.26 (s, 1H). ¹³C



NMR (CDCl₃) δ: 3.5, 7.3, 17.0, 19.1, 123.6, 130.4, 146.9, 149.2, 150.6. IR (ATR): 2952w, 2909 w, 2875 w. MS, *m/z* (EI, relative intensity, %): 221 (M⁺, 38), 136 (12), 115 (28), 87 (100), 59 (31). HRMS Calcd for C₁₃H₂₃NSi (M⁺): 221.1600. Found: 221.1595.

3-Butyl-4-[(triethylsilyl)methyl]pyridine (2c).

2c was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.11 in hexane/EtOAc = 30/1) in 76% yield as a colorless oil. ¹H NMR (CDCl₃) δ : 0.54 (q, J = 7.8 Hz, δ H), 0.91-0.95 (m, 12H), 1.39-1.42 (sixtet, J = 7.4 Hz, 2H), 1.52-1.60 (quintet, J = 7.4 Hz, 2H),

2.13 (s, 2H), 2.55 (t, J = 7.4 Hz, 2H), 6.87 (d, J = 5.0 Hz, 1H), 8.20 (d, J = 5.0 Hz, 1H), 8.26 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.3, 14.1, 18.6, 22.8, 30.4, 32.7, 124.0, 135.0, 146.8, 148.4, 150.2. IR (ATR): 2953 w, 2933 w, 2912 w, 2874 w. MS, m/z (EI, relative intensity, %): 248 (27), 234 (18), 221 (100), 134 (11), 115 (57), 106 (19), 87 (90), 59 (34). HRMS Calcd for C₁₆H₃₀NSi (M⁺+1): 264.2148. Found: 264.2148.

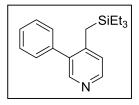
3-Cyclopentyl-4-[(triethylsilyl)methyl]pyridine (2d).

2d was obtained by flash column chromatography on silica-gel ($R_f = 0.16$ in hexane/EtOAc = 5/1) in 84% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.20$ in hexane/EtOAc = 30/1). ¹H NMR (CDCl₃)

δ: 0.54 (q, J = 8.0 Hz, 6H), 0.91 (t, J = 8.0 Hz, 9H), 1.63-1.69 (m, 4H), 1.84-1.88 (m, 2H), 2.00-2.06 (m, 2H), 2.17 (s, 2H), 3.04-3.12 (quintet, J = 8.4 Hz, 1H), 6.85 (d, J = 4.6 Hz, 1H), 8.17 (d, J = 4.6 Hz, 1H), 8.38 (s, 1H). ¹³C NMR (CDCl₃) δ: 3.5, 7.4, 19.0, 25.9, 34.3, 39.8, 123.9, 138.7, 146.2, 147.9, 148.3. IR (ATR): 2952 w, 2909 w, 2873 w. MS, m/z (EI, relative intensity, %): 275 (M⁺, 11), 274 (18), 247 (23), 246 (25), 235 (19), 234 (100), 232 (26), 132 (13), 121 (11), 1115 (44), 87 (77), 59 (36). HRMS Calcd for $C_{17}H_{29}NSi$ (M⁺): 275.2069. Found: 275.2066.

3-Phenyl-4-[(triethylsilyl)methyl]pyridine (2e).

2e was obtained by flash column chromatography on silica-gel (R_f = 0.23 in hexane/EtOAc = 10/1) in 74% yield as a colorless oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.26 in hexane). ¹H NMR (CDCl₃) δ : 0.33 (q, J = 7.8 Hz,



6H), 0.70 (t, J = 7.8 Hz, 9H), 2.26 (s, 2H), 7.02 (d, J = 5.0 Hz, 1H), 7.33-7.38 (m, 3H), 7.44 (t, J = 7.6 Hz, 2H), 8.33-8.35 (m, 2H). ¹³C NMR (CDCl₃) δ : 3.4, 7.1, 18.8, 124.2, 127.5, 128.6, 129.8, 136.7, 138.6, 147.9, 148.4, 150.5. IR (ATR): 2952 w, 2908 w, 2874 w. MS, m/z (EI, relative intensity, %): 283 (M⁺, 26), 282 (89), 254 (12), 115 (38), 87 (100), 59 (46). HRMS Calcd for $C_{18}H_{25}NSi$ (M⁺): 283.1756. Found: 283.1746.

3-(Naphthalen-2-yl)-4-[(triethylsilyl)methyl]pyridine (2f).

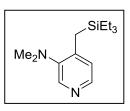
2f was obtained by flash column chromatography on silica-gel (R_f = 0.20 in hexane/EtOAc = 4/1) in 75% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.09 in

hexane/EtOAc = 30/1). ¹H NMR (CDCl₃) δ : 0.33 (q, J = 8.0 Hz, 6H), 0.67 (t, J = 8.0 Hz, 9H), 2.31 (s, 2H), 7.07 (d, J = 5.0 Hz, 1H), 7.47 (d, J = 8.2 Hz, 1H), 7.52-7.54 (m, 2H), 7.79 (s, 1H), 7.87-7.92 (m, 3H), 8.39 (d, J = 5.0 Hz, 1H), 8.43 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.4, 7.1, 19.1, 124.2, 126.3, 126.5, 127.9, 128.7, 132.6, 133.5, 136.1, 136.7, 148.2, 148.6, 150.8. IR (ATR):

2952 w, 2908 w, 2874 w. MS, *m/z* (EI, relative intensity, %): 333 (M⁺, 38), 332 (100), 304 (34), 219 (18), 218 (16), 217 (14), 115 (32), 87 (65), 59 (26). Anal. Calcd for C₂₂H₂₇NSi: C, 79.22; H, 8.16; N, 4.20. Found: C, 79.17; H, 8.26; N, 4.20.

N,*N*-Dimethyl-4-[(triethylsilyl)methyl]pyridin-3-amine (2g).

2g was obtained by flash column chromatography on silica-gel ($R_f = 0.14$ in hexane/EtOAc = 5/1) in 73% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.09$ in hexane). ¹H NMR (CDCl₃) δ : 0.49 (q, J = 8.0 Hz,



6H), 0.90 (t, J = 8.0 Hz, 9H), 2.17 (s, 2H), 2.71 (s, 6H), 6.89 (d, J = 5.0 Hz, 1H), 8.10 (d, J = 5.0 Hz, 1H), 8.25 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.3, 17.6, 44.5, 124.6, 142.1, 144.5, 145.9, 148.0. IR (ATR): 2951 w, 2908 w, 2874 w, 2828 w, 2782 w. MS, m/z (EI, relative intensity, %): 250 (M⁺, 17), 235 (16), 221 (47), 115 (39), 87 (100), 59 (39). HRMS Calcd for C₁₄H₂₆N₂Si (M⁺): 250.1865. Found: 250.1865.

3-Methoxy-4-[(triethylsilyl)methyl]pyridine (2h).

2h was obtained by flash silica-gel column chromatography ($R_f = 0.11$ in hexane/EtOAc = 2/1) in 73% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.09$ in hexane/EtOAc = 30/1). ¹H NMR (CDCl₃) δ : 0.50 (q,

J = 7.9 Hz, 6H), 0.91 (t, J = 7.9 Hz, 9H), 2.13 (s, 2H), 3.88 (s, 3H), 6.92 (t, J = 4.7 Hz, 1H), 8.06 (d, J = 4.7 Hz, 1H), 8.10 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.3, 7.3, 15.8, 55.5, 124.2, 132.1, 139.2, 142.3, 153.3. IR (ATR): 2952 w, 2908 w, 2875 w. MS, m/z (EI, relative intensity, %): 237 (M⁺, 33), 236 (32), 209 (15), 208 (88), 193 (23), 164 (20), 115 (54), 87 (100), 59 (36). HRMS Calcd for C₁₃H₂₃NOSi (M⁺): 237.1549. Found: 237.1543.

3-Chloro-4-[(triethylsilyl)methyl]pyridine (2i).

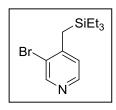
2i was obtained by flash column chromatography on silica-gel ($R_f = 0.10$ in hexane/EtOAc = 10/1) in 60% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.26$ in hexane). ¹H NMR (CDCl₃) δ : 0.57 (q, J = 7.9 Hz, 6H),

0.92 (t, J = 7.9 Hz, 9H), 2.31 (s, 2H), 6.98 (d, J = 5.0 Hz, 1H), 8.25 (d, J = 5.0 Hz, 1H), 8.45 (s,

1H). 13 C NMR (CDCl₃) δ : 3.4, 7.3, 20.0, 124.5, 131.0, 147.0, 148,9, 149.3. IR (ATR): 2953 w, 2909 w, 2876 w. MS, m/z (EI, relative intensity, %): 243 (13), 241 (M⁺, 38), 212 (25), 156 (13), 115 (53), 87 (100), 59 (26). HRMS Calcd for $C_{12}H_{20}CINSi$ (M⁺): 241.1054. Found: 241.1055.

3-Bromo-4-[(triethylsilyl)methyl]pyridine(2j).

2j was obtained by flash column chromatography on silica-gel ($R_f = 0.20$ in hexane/EtOAc = 8/1) in 61% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.26$ in hexane). ¹H NMR (CDCl₃) δ : 0.58 (q, J = 7.9 Hz, 6H),



0.92 (t, J = 7.9 Hz, 9H), 2.34 (s, 2H), 6.99 (d, J = 4.8 Hz, 1H), 8.27 (d, J = 4.8 Hz, 1H), 8.58 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.3, 22.7, 122.4, 124.7, 147.6, 150.8, 151.9. IR (ATR): 2952 w, 2909 w, 2875 w. MS, m/z (EI, relative intensity, %): 287 (12), 285 (M⁺, 12), 258 (10), 256 (10), 115 (47), 87 (100), 59 (33). HRMS Calcd for C₁₂H₂₀BrNSi (M⁺): 285.0548. Found: 285.0551.

3,5-Dimethyl-4-[(triethylsilyl)methyl]pyridine (2k).

2k was obtained by flash column chromatography on silica-gel ($R_f = 0.17$ in hexane/EtOAc = 1/1) in 50% yield as a pale yellow oil. Analytically pure sample was obtained by column chromatography on NH₂-modified silica-gel ($R_f = 0.09$ in hexane/EtOAc = 30/1). ¹H NMR (CDCl₃) δ : 0.56 (q, J = 7.9 Hz,

6H), 0.89 (t, J = 7.9 Hz, 9H), 2.13 (s, 2H), 2.21 (s, 6H), 8.12 (s, 2H). ¹³C NMR (CDCl₃) δ : 4.8, 7.3, 16.2, 17.6, 129.9, 148.2. 148.3. IR (ATR): 2952 w, 2911 w, 2875 w. MS, m/z (EI, relative intensity, %): 236 (18), 235 (M⁺, 90), 178 (10), 150 (12), 115 (38), 87 (100), 59 (23). HRMS Calcd for C₁₄H₂₅NSi (M⁺): 235.1756. Found: 235.1758.

4-[1-(Triethylsilyl)hexyl]pyridine (21).

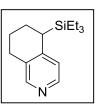
21 was obtained by flash column chromatography on silica-gel ($R_f = 0.23$ in hexane/EtOAc = 5/1) in 50% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.14$ in hexane). ¹H

NMR (CDCl₃) δ : 0.51 (q, J = 7.9 Hz, 6H), 0.83 (t, J = 6.6 Hz, 3H), 0.89 (t, J = 7.9 Hz, 9H), 1.05-1.28 (m, 6H), 1.64-1.72 (m, 1H), 1.80-1.90 (m, 1H), 2.18 (dd, J = 12.4, 3.1 Hz, 1H), 6.97 (dd, J = 4.6, 1.5 Hz, 2H), 8.40 (dd, J = 4.6, 1.5 Hz, 2H). 13 C NMR (CDCl₃) δ : 2.3, 7.6, 14.2, 22.6, 28.9,

29.2, 31.7, 34.6, 123.4, 149.4, 154.0. IR (ATR): 2953 m, 2928 m, 2874 w. MS, m/z (EI, relative intensity, %): 277 (M⁺, 2), 234 (10), 221 (23), 220 (100), 115 (54), 87 (98), 59 (64). HRMS Calcd for $C_{17}H_{31}NSi$ (M⁺): 277.2226. Found: 277.2225.

5-Triethylsilyl-5,6,7,8-tetrahydroisoquinoline (2m).

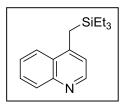
2m was obtained by flash silica-gel column chromatography on $(R_f = 0.20 \text{ in hexane/EtOAc} = 8/1)$ in 55% yield as a pale yellow oil. Analytically pure sample was obtained by column chromatography on NH₂-modified silica-gel $(R_f = 0.13 \text{ in hexane/EtOAc} = 30/1)$. ¹H NMR (CDCl₃) δ : 0.59 (dd, J = 15.8, 8.0 Hz, 6H),



0.89 (t, J = 8.0 Hz, 9H), 1.81-2.00 (m, 4H), 2.50 (t, J = 5.7 Hz, 1H), 2.62-2.78 (m, 2H), 6.86 (d, J = 5.0 Hz, 1H), 8.18 (d, J = 5.0 Hz, 1H), 8.22 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.1, 7.5, 22.0, 24.2, 25.9, 26.8, 123.0, 132.4, 146.2, 149.6, 150.3. IR (ATR): 2947 w, 2912 w, 2874 w. MS, m/z (EI, relative intensity, %): 247 (M⁺, 77), 246 (100), 232 (12), 132 (14), 115 (35), 87 (81), 59 (40). HRMS Calcd for C₁₅H₂₅NSi (M⁺): 247.1756. Found: 247.1750.

4-[(Triethylsilyl)methyl]quinoline (2n).

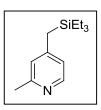
2n was obtained by flash column chromatography on silica-gel (R_f = 0.57 in hexane/EtOAc = 1/1) in 62% yield as a pale yellow oil. Analytically pure sample was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.14 in hexane/EtOAc = 6/1). ¹H NMR (CDCl₃) δ : 0.53 (q, J



= 7.9 Hz, 6H), 0.89 (t, J = 7.9 Hz, 9H), 2.61 (s, 2H), 7.09 (d, J = 4.4 Hz, 1H), 7.51-7.53 (m, 1H), 7.66-7.70 (m, 1H), 7.97 (d, J = 8.7 Hz, 1H), 8.07 (d, J = 8.2 Hz, 1H), 8.70 (d, J = 4.4 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.3, 18.9, 120.5, 124.2, 125.6, 127.7, 129.0, 130.2, 148.4, 148.5, 149.7. IR (ATR): 2952 w, 2908 w, 2875 w. MS, m/z (EI, relative intensity, %): 257 (M⁺, 39), 143 (21), 115 (59), 87 (100), 59 (32). HRMS Calcd for C₁₆H₂₃NSi (M⁺): 257.1600. Found: .257.1595.

2-Methyl-4-[(triethylsilyl)methyl]pyridine (20).

20 was obtained by flash column chromatography on silica-gel ($R_f = 0.06$ in hexane/EtOAc = 5/1) in 72% yield as a pale yellow oil. Analytically pure sample was obtained by column chromatography on NH₂-modified silicagel ($R_f = 0.13$ in hexane/EtOAc = 6/1). ¹H NMR (CDCl₃) δ : 0.52 (q, J = 8.0

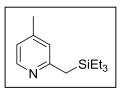


Hz, 6H), 0.92 (t, J = 8.0 Hz, 9H), 2.06 (s, 2H), 2.48 (s, 3H), 6.74 (d, J = 5.0 Hz, 1H), 6.80 (s,

1H), 8.26 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.0, 7.3, 22.0, 24.4, 120.9, 123.1, 148.7, 150.9, 157.8. IR (ATR): 2952 w, 2910 w, 2876 w. MS, m/z (EI, relative intensity, %): 222 (12), 221 (M⁺, 62), 164 (12), 136 (21), 115 (49), 87 (100), 59 (23). HRMS Calcd for C₁₃H₂₃NSi (M⁺): 221.1600. Found: 221.1597.

Alternative Sample Preparation of 4-Methyl-2-[(triethylsilyl)methyl]pyridine (20").

20" was produced by the lithiation of **10** (1.1 mL, 10 mmol) with 1.6 M BuLi (6.3 mL, 10 mmol) to generate 2-lithiomethyl-4-methylpyridine,³⁴ and subsequent trapping with chlorotriethylsilane (1.7 mL, 10 mmol). The product was isolated by flash column chromatography on NH₂-modified



silica-gel (R_f = 0.29 in hexane/EtOAc = 10/1) in 63% yield (1.40 g, 6.33 mmol) as a colorless oil. 1H NMR (CDCl₃) δ : 0.54 (q, J = 7.9 Hz, 6H), 0.90 (t, J = 7.9 Hz, 9H), 2.27 (s, 3H), 2.31 (s, 2H), 6.79 (s, 2H), 8.27-8.27 (m, 1H). 13 C NMR (CDCl₃) δ : 3.4, 7.3, 21.0, 25.2, 120.3, 123.1, 146.6, 148.8, 161.4. IR (ATR): 2952 w, 2909 w, 2874 w. MS, m/z (EI, relative intensity, %): 221 (M⁺, 3), 220 (10), 193 (19), 192 (100), 164 (14), 137 (11), 136 (10), 134 (16). HRMS Calcd for $C_{13}H_{23}NSi$ (M⁺+1): 222.1678. Found: 222.1681.

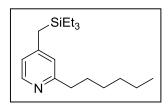
Alternative Sample Preparation of 2,4-Bis[(triethylsilyl)methyl|pyridine (20").

20" was produced by the Ir₄(CO)₁₂-catalyzed silylation of **20** (437 mg, 0.97 mmol) at the 2-position with HSiEt₃ (320 μ L, 2 mmol) in the presence of norbornene (188 mg, 2 mmol) at 160 °C for 20 h. The product was isolated by flash column chromatography on silica-gel ($R_f = 0.11$ in

hexane/EtOAc =10/1) in 31% yield (101 mg) as a pale yellow oil. 1 H NMR (CDCl₃) δ : 0.51-0.55 (m, 12H), 0.90-0.94 (m, 18H), 2.03 (s, 2H), 2.28 (s, 2H), 6.63-6.64 (m, 2H), 8.19 (dd, J = 4.4, 1.6 Hz, 1H). 13 C NMR (CDCl₃) δ : 3.1, 3.4, 7.4, 21.9, 25.2, 119.6, 122.1, 148.6, 150.2, 161.1. IR (ATR): 2952 w, 2909 w, 2874 w. MS, m/z (EI, relative intensity, %): 335 (M⁺, 7), 334 (11), 308 (10), 307 (32), 306 (100), 251 (18), 192 (12), 115 (11), 87 (41), 59 (23). HRMS Calcd for $C_{19}H_{37}NSi_2$ (M⁺): 335.2465. Found: 335.2469.

2-Hexyl-4-[(triethylsilyl)methyl]pyridine (2p).

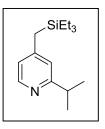
2p was obtained by flash column chromatography on silica-gel (R_f = 0.31 in hexane/EtOAc = 6/1) in 59% yield as a pale yellow oil. Analytically pure sample was obtained by column chromatography on NH₂-modified silica-gel (R_f = 0.29 in hexane/EtOAc = 30/1). ¹H NMR



(CDCl₃) δ : 0.51 (q, J= 7.9 Hz, 6H), 0.88-0.92 (m, 12H), 1.21-1.39 (br m, 6H), 1.65-1.73 (quintet, J= 7.6 Hz, 2H), 2.08 (s, 2H), 2.70 (t, J= 7.6 Hz, 2H), 6.75 (d, J= 5.5 Hz, 1H), 6.79 (s, 1H), 8.29 (d, J= 5.5 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.0, 7.3, 14.2, 22.1, 22.7, 29.2, 30.1, 31.9, 38.5, 121.0, 122.6, 148.8, 150.7, 162.0. IR (ATR): 2952 w, 2926 w, 2875 w, 2857 w. MS, m/z (EI, relative intensity, %): 291 (M⁺, 3), 248 (14), 234 (31), 222 (20), 221 (100), 206 (11), 120 (10), 115 (25), 107 (24), 87 (63), 59 (23). HRMS Calcd for C₁₈H₃₃NSi (M⁺): 291.2382. Found: 291.2382.

2-Isopropyl-4-[(triethylsilyl)methyl]pyridine (2q).

2q was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.11 in hexane) in 60% yield as a colorless oil. ¹H NMR (CDCl₃) δ : 0.51 (q, J = 8.0 Hz, 6H), 0.92 (t, J = 8.0 Hz, 9H), 1.28 (d, J = 6.9 Hz, 6H), 2.09 (s, 2H), 2.98 (septet, J = 6.9 Hz, 1H), 6.75 (d, J = 5.0 Hz, 1H), 6.82 (s, 1H), 8.30 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.0, 7.3, 22.2, 22.7, 36.3, 120.5,



121.2, 148.7, 150.8, 166.9. IR (ATR): 2955 w, 2909 w, 2875 w. MS, m/z (EI, relative intensity, %): 250 (11), 249 (M⁺, 50), 234 (10), 192 (11), 164 (13), 115 (70), 88 (11), 87 (100), 59 (24). HRMS Calcd for $C_{15}H_{27}NSi$ (M⁺): 249.1913. Found: 249.1911.

N,*N*-Dimethyl-4-[(triethylsilyl)methyl]pyridin-2-amine (2r).

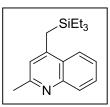
2r was obtained by flash column chromatography on silica-gel ($R_f = 0.31$ in hexane/EtOAc = 6/1) in 43% yield as a pale yellow oil. Analytically pure sample was obtained by column chromatography on NH₂-modified silica-gel ($R_f = 0.09$ in hexane). ¹H NMR (CDCl₃) δ : 0.53 (q, J = 7.9 Hz,

6H), 0.94 (t, J = 7.9 Hz, 9H), 2.02 (s, 2H), 3.05 (s, 6H), 6.17 (s, 1H), 6.25 (d, J = 5.3 Hz, 1H), 7.97 (d, J = 5.3 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.2, 7.4, 22.3, 38.3, 105.1, 112.8, 147.4, 151.2, 159.8. IR (ATR): 2949 w, 2911 w, 2871 w. MS, m/z (EI, relative intensity, %): 251 (17), 250 (M⁺, 80), 235 (35), 222 (20), 221 (100), 206 (14), 193 (20), 178 (10), 165 (17), 134 (38), 120 (11), 115 (34), 87 (100), 75 (10), 59 (32). HRMS Calcd for $C_{14}H_{26}N_2Si$ (M⁺): 250.1865. Found:

250.1865.

2-Methyl-4-[(triethylsilyl)methyl]quinoline (2s).

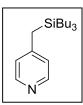
2s was obtained by flash column chromatography on silica-gel ($R_f = 0.20$ in hexane/EtOAc = 1/10) in 52% yield as a pale yellow oil. Analytically pure sample was obtained by bulb-to-bulb distillation (142 °C/0.9 mmHg). ¹H NMR (CDCl₃) δ : 0.58 (q, J = 7.9 Hz, 6H), 0.93 (t, J = 7.9 Hz, 9H), 2.51 (s,



2H), 2.64 (s, 3H), 6.96 (s, 1H), 7.44 (t, J = 8.0 Hz, 1H), 7.62 (t, J = 8.0 Hz, 1H), 7.91 (d, J = 8.0 Hz, 1H), 7.95 (d, J = 8.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.6, 7.4, 18.8, 26.7, 122.5, 123.6, 124.7, 126.0, 128.9, 129.0, 143.4, 148.1, 162.0. IR (ATR): 2951 w, 2908 w, 2874 w. MS, m/z (EI, relative intensity, %): 271 (M⁺, 23), 270 (34), 243 (24), 242 (100), 184 (24). HRMS Calcd for $C_{17}H_{25}NSi$ (M⁺): 271.1756. Found: 271.1754.

4-[(Tributylsilyl)methyl]pyridine (3a).

3a was obtained by flash column chromatography on NH₂-modified silica-gel ((R_f = 0.20 in hexane/EtOAc = 30/1) in 65% yield as a colorless oil. ¹H NMR (CDCl₃) δ : 0.50 (t, J = 8.2 Hz, 6H), 0.87 (t, J = 7.1 Hz 9H), 1.21-1.32 (m, 12H), 2.10 (s, 2H), 6.92 (d, J = 6.0 Hz, 2H), 8.37 (d, J = 6.0 Hz, 2H). ¹³C NMR (CDCl₃) δ : 11.7,



13.8, 23.2, 26.0, 26.8, 123.7, 149.4, 150.7. IR (ATR): 2955 w, 2921 w, 2870 w, 2858 w. MS, *m/z* (EI, relative intensity, %): 292 (19), 291 (M⁺, 78), 178 (12), 144 (13), 143 (100), 122 (23), 120 (15), 101 (68), 87 (41), 73 (19), 59 (42). Anal. Calcd for C₁₈H₃₃NSi: C, 74.15; H, 11.41; N, 4.80. Found: C, 73.81; H, 11.62; N, 4.89.

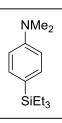
1-Methyl-3-triethylsilyl-1*H*-indole (5).⁷ⁱ

5 was obtained by flash column chromatography on silica-gel ($R_f = 0.29$ in hexane/EtOAc/Et₃N =20/1/1) in 44% yield as a colorless oil. ¹H NMR (CDCl₃) δ : 0.86 (q, J = 7.7 Hz, 6H), 0.99 (t, J = 7.7 Hz, 9H), 3.79 (s, 3H), 7.04 (s, 1H), 7.11 (d, J = 7.3 Hz, 1H), 7.22 (t, J = 7.1 Hz, 1H), 7.32 (d, J = 8.2 Hz, 1H), 7.68

(d, J = 7.8 Hz, 1H). ¹³C NMR (CDCl₃) δ : 4.43, 7.30, 32.9, 105.7, 109.4, 119.3, 121.5, 122.3, 133.4, 136.1, 138.4.

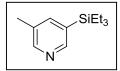
N,N-Dimethyl-4-(triethylsilyl)aniline (7).7b

7 was obtained by flash column chromatography on silica-gel ($R_f = 0.31$ in hexane/EtOAc =12/1) in 10% yield as a pale yellow oil. ¹H NMR (CDCl₃) δ : 0.75 (q, J = 7.8 Hz, 6H), 0.96 (t, J = 7.8 Hz, 10H), 2.96 (s, 6H), 6.74 (d, J = 8.8 Hz, 2H), 7.37 (dt, J = 8.8, 2.2 Hz, 2H). ¹³C NMR (CDCl₃) δ : 3.73, 7.69, 40.4, 111.2, 122.4, 135.4, 150.9.



Alternative Sample Preparation of 3-Methyl-5-(triethylsilyl)pyridine.

The procedure with 3-bromo-5-methylpyridine (340 mg, 1.98 mmol) as a starting material, reported by Reuman, was modified using chlorotriethylsilane (340 μ L, 2.0 mmol) in place of chlorotrimethylstannane



to produce the title compound, which was isolated by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.20$ in hexane/EtOAc =10/1) in 29% yield (112 mg, 0.58 mmol) as a pale yellow oil. ¹H NMR (CDCl₃) δ : 0.81 (q, J = 7.7 Hz, 6H), 0.97 (t, J = 7.7 Hz, 9H), 2.33 (s, 3H), 7.26 (s, 1H), 8.41 (s, 1H), 8.45 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.2, 7.3, 18.7, 131.5, 132.3, 142.4, 150.5, 151.8. IR (ATR): 2953 w, 2910 w, 2875 w. MS, m/z (EI, relative intensity, %): 207 (M⁺, 7), 178 (65), 151 (12), 150 (100), 122 (72). HRMS Calcd for $C_{12}H_{21}NSi$ (M⁺): 207.1443. Found: 207.1441.

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Chapter 2

Ir₄(CO)₁₂-Catalyzed Benzylic C(sp³)-H Silylation of 2-Alkylpyridines with Hydrosilanes Leading to 2-(1-Silylalkyl)pyridines

2.1 Introduction

As described in the general introduction, there is only one example of the direct $C(sp^3)$ -H silylation of 2-alkylpyridine at the α -positions with HSiEt₃ catalyzed by potassium *tert*-butoxide.¹ In this reaction, the product yield was moderate, which probably proceeded via a radical chain mechanism² and/or the mechanism involving a pentacoordinate silicon species as a key intermediate.³ In the course of the study of $C(sp^3)$ -H silylation of 4-alkylpyridine, described in Chapter 1, when 2,4-dimethylpyridine was used as a substrate, a trace amount of a product in which the methyl group was silylated at the 2-position was also found in the reaction mixture. This motivated me to explore the reactivity of α -C-H silylation of C2-substituents in pyridine. Chapter 2 describes the $C(sp^3)$ -H silylation of 2-alkylpyridines with hydrosilanes, leading to the formation of 2-(1-silylalkyl)pyridines, catalyzed by an iridium complex/3,5-dimethylpyridine system (scheme 1).

Scheme 1. Ir₄(CO)₁₂-Catalyzed Benzylic C(sp³)-H Silylation of 2-Alkylpyridines with Hydrosilanes

2.2 Results and Discussion

After a series of explorative experiments intended to optimize the reaction conditions, the reaction of 2-methylpyridine (1a) with triethylsilane in the presence Ir₄(CO)₁₂, norbornene, and 3,5-dimethylpyridine in the absence of a solvent, at 160 °C for 40 h under a N₂ atmosphere, afforded a mixture of the desired product 2a (67%, Table 1, entry 1), the disilylation product 2a' (16%), and the unreacted 1a (8%). Both 2a and 2a' were isolated by flash chromatography on NH₂-modified silica-gel in 59% and 16%, respectively, as the sole products. Deviation from these optimized reaction conditions led to decreased yields. Decreasing the reaction temperature

to 140 °C and the use of toluene as a solvent resulted in a decrease in the yield of **2a** in 38% and 22%, respectively (entries 2 and 3). As is the case with the previous report on the catalytic C(sp³)-H silylation of 4-alkylpyridines, the addition of both norbornene (entry 4) and 3,5-dimethylpyridine (entry 5) was required for the reaction to proceed. Screening some catalysts revealed that Ir(acac)(CO)₂ (entry 6) and [Ir(OMe)(cod)]₂ under an atmosphere of CO (entry 8, versus entry 7) showed catalytic activity in the reaction, indicating that CO is also an essential ligand for the reaction. Transition metal carbonyls other than Ir₄(CO)₁₂, such as Rh₄(CO)₁₂ (entry 9) and Ru₃(CO)₁₂ (entry 10), were also found to be effective catalysts for the reaction. When triphenylsilane was used in the present reaction, **3a** was produced in 80% yield, as determined from the ¹H NMR spectrum (entry 11). However, when the product was isolated by NH₂-modified silica-gel column chromatography, the yield of **3a** was decreased significantly, to 54%, probably due to its susceptibility to hydrolysis while it was on the silica-gel column. On the other hand, when HSi(OEt)₃ was used in the reaction, the product was not produced.

Table 1. Optimization of the Reaction Conditions^a

entry	deviation from the standard reaction conditions	yields (%) ^b		
		2a	2a'	1a
1	None	$67 (59)^c$	$16(16)^c$	8
2	At 140 °C	38	7	34
3	In toluene (0.5 mL)	22	2	73
4	Without norbornene	24	trace	30
5	Without 3,5-dimethylpyridine	11	0	74
6	$Ir(acac)(CO)_2 (0.05 \text{ mmol})$	54	8	16
7	$[Ir(OMe)(cod)]_2$ (0.025 mmol)	0	0	79
8	[Ir(OMe)(cod)] ₂ (0.025 mmol) under CO (1 atm)	63	6	15
9	$Rh_4(CO)_{12}$ (0.0125 mmol)	61	9	11
10	$Ru_3(CO)_{12}$ (0.017 mmol)	32	3	57
11	HSiPh ₃ in place of HSiEt ₃	80 (54, 3	a) 0	8

^aReaction conditions: **1a** (0.5 mmol), HSiEt₃ (1 mmol), Ir₄(CO)₁₂ (0.0125 mmol), norbornene (1 mmol), and 3,5-dimethylpyridine (0.1 mmol) 160 °C for 40 h under N₂. ^bYields of **2a** and **2a**' were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. The yield of the unreacted **1a** was determined by GC with tridecane as an internal standard. ^cThe number shown in parentheses is the isolated yield.

With the optimized reaction conditions in hand, the author next evaluated the substrate scope of the C(sp³)-H silvlation of **1b-1o**, and the results are shown in Table 2. When a 2methylpyridne derivative containing a methyl group at the 3-position was used, no silylation occurred at this position, and 2b was isolated in 75% yield. Further silvlation of 2b to produce **2b'** also did not occur, probably due to steric hindrance by the methyl group at the 3-position. Although the author did not examine the reactions of 4-methyl- and 4-primary alkyl-substituted 2-methylpyridines, because the benzylic C-H bond in the 4-substituents were readily silylated, ¹⁸ the 4-isopropyl-substituted derivatives 1d could be applied to the reaction to afford the desired product 2d in 64%, along with the formation of 2d' in 4% yield. The reactions of 2methylpyridines bearing substituents such as methyl (1e), phenyl (1f), and dimethylamino (1g) groups at the 5-position gave, as suspected, a mixture of the monosilylated and disilylated products. In the case of 2-methyl-5-phenoxypyridine (1h), 2h was isolated in 39% yield, along with 1h being recovered in 53% yield. Efforts to increase the product yield of 2h were in vain. Ethyl-, hexyl-, and β-phenetyl-substituted pyridines at the 2-position (1i-1k), reacted to produce 2i, 2j, and 2k in 71%, 58%, and 57% yield, respectively, and no silylated compound at the homobenzylic position was observed in any of these cases. ⁴ The 6-membered ring-fused pyridine, 11, also underwent regioselective silvlation to give 21 in 73% yield and 2m was formed from 2methylquinoline (1m) in 66% yield. In the case of harmine (1n), a harmala alkaloid belonging to the β-carboline family of compounds, the amount of HSiEt₃ was increased to 1.5 equivalent relative to 1n, because the N-H bond in 1n was also silvlated, and the desired product 2n was isolated in 86% after chromatographic purification. However, the reaction of 2,6dimethylpyridine resulted in the quantitative recovery of the starting material. This result indicates that the present reaction proceeds via a mechanism different from that catalyzed by potassium tert-butoxide. Although the reaction system was also applicable to the silylation of 2-methylthiazole (10) to afford 20 in 53% yield, the reaction of 2-methyloxazole gave a complex reaction mixture.

Table 2. Scope of Substrates^a

"Reaction conditions: **1** (0.5 mmol), HSiEt₃ (1 mmol), Ir₄(CO)₁₂ (0.0125 mmol), norbornene (1 mmol), and 3,5-dimethylpyridine (0.1 mmol) 160 °C for 40 h under N₂. Isolated yields are given. The number shown in parentheses is the isolated yield of disilylation product **2**°. ^bFor 4 days. ^cFor 3 days. ^dHSiEt₃ (1.5 mmol).

A proposed mechanism for the regioselective $C(sp^3)$ -H silylation of 2-alkylpyridines is depicted in Scheme 2, which is almost the same as that proposed in chapter 1, including an electrophilic silicon species as a key intermediate. The coordination of a hydrosilane to an iridium carbonyl complex **I** occurs to form the electrophilic silicon species, either an η^1 -silane iridium complex **II** or a σ -silane iridium complex **II**. Next, the pyridine nitrogen attacks the silicon atom in **II** or **II** to cleave the Si-H bond heterolytically to form an *N*-silylpyridinium hydridoiridate **III**. Deprotonation at the benzylic position in the *N*-silyl pyridinium by the hydrideiridate then results in the formation of a dihydridoiridium species **IV** and an *N*-silyl enamine **V**. 1,3-Silyl migration of the latter from the nitrogen atom to the carbon atom occurs to produce the desired product. **IV** reacts with norbornene to regenerate the iridium catalyst **I** with the concomitant formation of norbornane. The author speculates that the added 3,5-dimethylpyridine functions as a transporter

of the silyl group via the formation of III.⁶ The reason that the present reaction requires higher reaction temperature of 160 °C can be attributed to the steric hindrance around the alkyl group at the 2-position for the abstraction of the benzylic proton in III. The possibility that the methyl group at the 2-position hinders the formation of III by steric hindrance can be ruled out. This is because, as described in chapter 1, a 2-methyl substituted silylpyridinium species is likely to be generated in the catalytic cycle under the almost same reaction conditions at a reaction temperature of 100 °C, leading to the C(sp³)-H silylation of methyl group at the 4-position of 2,4-dimethylpyridine. On the other hand, the decreased reaction rate in the case of the reactions of 2i-2m and the fact that 2,6-dimethylpyridine failed to participate in the reaction might be also related, partly or significantly, to steric hindrance by the substituents adjacent to the pyridine nitrogen atom to form III.

Scheme 2. A Plausible Reaction Mechanism

$$| Ir(CO)_n | Ir(CO)_n | H | (CO)_n | Ir(CO)_n | Ir(CO$$

2.3 Conclusion

The author has demonstrated the iridium-catalyzed regioselective C(sp³)-H silylation of 2-alkylpyridines with hydrosilanes at the benzylic position. Both norbornene and 3,5-dimethylpyridine are essential additives for the reaction to proceed. Ir₄(CO)₁₂ and Ir(acac)(CO)₂, and [Ir(OMe)(cod)]₂ under ambient atmosphere of CO all showed catalytic activity for the reaction, which indicates that carbon monoxide plays an important role in the catalytic cycle by functioning as a ligand. Other transition metal carbonyls such as Rh₄(CO)₁₂ and Ru₃(CO)₁₂ can also be used as catalysts for the present C-H silylation. The formation of an electrophilic silicon species is likely to be involved in the reaction as a key intermediate in the catalytic cycle. The added 3,5-dimethylpyridine is thought to serve as the transporter of the silyl group to form a silylpyridinium intermediate.

2.4 Experimental Section

General Information.

¹H and ¹³C NMR spectra were recorded on 400, and 100 MHz spectrometers, respectively, using CDCl₃ as the solvent. Data are recorded as follows: chemical shifts in ppm (δ), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, c = complex), coupling constant (Hz), and integration. Infrared spectra (IR) were recorded by an ATR technique. Absorption data are reported in reciprocal centimeters with the following relative intensities: s (strong), m (medium), or w (weak). Mass spectra were obtained using a spectrometer with a quadrupole mass analyzer at 70 eV. High-resolution mass spectra (HRMS) were obtained using a spectrometer with a double-focusing mass analyzer. Analytical gas chromatography (GC) was carried out on a chromatograph equipped with a flame ionization detector. Melting points were determined on a capillary point apparatus equipped with a digital thermometer and are uncorrected.

Materials.

Toluene was purified by passage through activated alumina under a positive pressure of N₂. Norbornene and **1n** were purchased and were used as received. Other organic compounds, except **1c**, **1d**, **1f-1h**, **1j**, and **1k** were purchased from commercial sources and distilled over CaH₂ before use. Compounds **1c**, **1d**, **1g**, and **1h** were prepared as described below. Compounds **1f**, ⁷ **1j**, ⁸ and **1k** were prepared following procedure described in the literature. All metal complexes were purchased from commercial sources and used without further purification.

Typical Procedure for the Ir₄(CO)₁₂-Catalyzed C-H Silylation of 2-Alkylpyridines with Hydrosilanes.

A 8 mL Schlenk tube was flame-dried and purged with N₂. After cooling to room temperature, Ir₄(CO)₁₂ (13.8 mg, 0.0125 mmol), 3,5-dimethylpyridine (11 mg, 0.1 mmol), hydrosilane (1 mmol), **1** (0.5 mmol), and norbornene (94 mg, 1 mmol) were placed in the Schlenk tube. The reaction mixture was stirred at 160 °C for 40 h. After cooling to room temperature, the volatiles were removed *in vacuo*. The product was isolated by flash column chromatography on silica-gel and subsequent filtration through a pad of NH₂-modified silica-gel to remove silicon impurities.

2-Methyl-3-phenylpyridine (1c).¹⁰

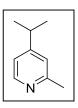
The procedure reported by Tobisu and Chatani¹¹ was modified by using 3-bromo-2-methylpyridine (0.86 g, 5.0 mmol) in place of (2-bromophenyl)phenyl sulfide to produce **1c**, which was isolated by distillation under reduced pressure



(140 °C/5 mmHg) in 82% yield (0.69 g, 4.1 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 2.51 (s, 3H), 7.18 (dd, J = 7.8, 4.7 Hz, 1H), 7.31-7.32 (m, 2H), 7.35-7.39 (m, 1H), 7.42-7.45 (m, 2H), 7.51 (dd, J = 7.8, 1.7 Hz, 1H), 8.50 (dd, J = 4.7, 1.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 23.5, 121.1, 127.6, 128.5, 129.1, 137.1, 137.3, 140.1, 148.0, 155.9.

4-Isopropyl-2-methylpyridine (1d). 12

The procedure reported by Comins¹² was modified by using DDQ¹³ in place of sulfur for the aromatization of dihydropyridine. **1d** was isolated from the crude product by distillation under reduced pressure (58 °C/3.2 mmHg) in 6% yield (173.5 mg, 1.28 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 1.24 (d, J = 6.9 Hz,



6H), 2.53 (s, 3H), 2.79-2.89 (m, 1H), 6.94 (d, J = 5.3 Hz, 1H), 7.00 (s, 1H), 8.36 (d, J = 5.3 Hz, 1H). ¹³C NMR (CDCl₃) δ : 23.2, 24.5, 33.6, 119.2, 121.6, 149.1, 157.9, 158.3.

N,N-6-Trimethylpyridin-3-amine (1g). 14

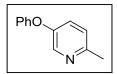
The procedure reported by $Aron^{15}$ was modified by using 2-methylpyridin-5-amine (1.00 g, 9.2 mmol) in place of pyridin-3-amine to produce $\mathbf{1g}$, which was isolated by distillation under reduced pressure (57 °C/1.2 mmHg) in 61%

yield (0.76 g, 5.60 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 2.45 (s, 3H), 2.93 (s, 6H), 6.95 (dd, J = 8.5, 3.0 Hz, 1H), 6.99 (d, J = 8.5 Hz, 1H), 8.04 (d, J = 3.0 Hz, 1H). ¹³C NMR (CDCl₃)

δ: 23.1, 40.5, 120.2, 122.9, 134.5, 144.4, 146.0.

2-Methyl-5-phenoxypyridine (1h).¹⁶

The procedure reported by Olofsson¹⁷ was modified by using 6-methylpyridin-3-ol (1.39 g, 12.8 mmol) in place of pyridin-3-ol to produce **1h**, which was isolated by distillation under reduced pressure (85 °C/0.8



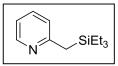
mmHg) in 81% yield (1.92 g, 10.4 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 2.55 (s, 3H), 6.99 (dd, J = 8.7, 0.9 Hz, 2H), 7.10-7.14 (m, 2H), 7.22 (dd, J = 8.2, 2.7 Hz, 1H), 7.32-7.37 (m, 2H), 8.30 (d, J = 2.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 23.7, 118.3, 123.5, 123.7, 126.6, 129.9, 140.9, 151.3, 153.2, 157.0.

2-[(Triethylsilyl)methyl]pyridine (2a).

After reacting **1a** (46.3 mg, 0.50 mmol) under the standard reaction conditions, the reaction mixture was passed through a flash silica-gel column to separate **2a** (62.2 mg, $R_f = 0.17$ in hexane/EtOAc = 20/1) and 2-[bis(triethylsilyl)methyl]pyridine (**2a'**) (32.2 mg, $R_f = 0.46$ in hexane/EtOAc = 20/1), although both products still contained some silicon impurities.

2-[(Triethylsilyl)methyl]pyridine (2a).

The crude mixture containing 2a (62.2 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure 2a in 59% yield (60.9 mg) as a colorless



oil. HNMR (CDCl₃) δ : 0.54 (q, J = 8.0 Hz, δ H), 0.90 (t, J = 8.0 Hz, θ H), 2.36 (s, 2H), 6.95-6.97 (m, 2H), 7.48 (td, J = 7.8, 1.8 Hz, 1H), 8.41 (d, J = 4.6 Hz, 1H) 13 C NMR (CDCl₃) δ : 3.4, 7.4, 25.5, 119.2, 122.4, 136.0, 149.0, 161.7. IR (ATR): 2952 w, 2909 w, 2874 w. MS, m/z (CI, relative intensity, %): 208 ([M+H]⁺, 100), 178 (12). HRMS Calcd for C₁₂H₂₂NSi ([M+H]⁺): 208.1522. Found: 208.1525.

2-[Bis(triethylsilyl)methyl]pyridine (2a').

The crude mixture containing **2a'** (32.2 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2a'** in 16% yield (26.2 mg) as a colorless oil. However, since **2a'** gradually decomposed during handling, it was not possible to collect spectroscopic spectra, except for ¹H NMR. Therefore, we modified the procedure for producing **2a'** as the major product by using an excess amount of HSiEt₃ and norbornene. Details are described below.

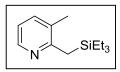
2-[Bis(triethylsilyl)methyl]pyridine (2a').

2a' was produced by the present $Ir_4(CO)_{12}$ -catalyzed reaction of **1a** (46.6 mg, 0.50 mmol) with HSiEt₃ (480 mL, 3.00 mmol) in the presence of norbornene (282 mg, 3.00 mmol) at 160 °C for 4 days. The reaction mixture was passed through a flash silica-gel column to separate **2a** (3.1 mg, $R_f = 0.17$ in

hexane/EtOAc = 20/1) and 2-[bis(triethylsilyl)methyl]pyridine (**2a'**) (102.4 mg, R_f = 0.46 in hexane/EtOAc = 20/1), although both products still contained some silicon impurities. The crude mixture containing **2a'** was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2a'** in 61% yield (98.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.59 (m, 12H), 0.88 (t, J = 8.0 Hz, 18H), 2.11 (s, 1H), 6.88-6.92 (m, 2H), 7.41 (td, J = 7.7, 1.5 Hz, 1H), 8.38 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 4.9, 7.9, 26.2, 118.3, 123.3, 135.3, 148.8, 164.1. IR (ATR): 2950 m, 2909 w, 2875 m. MS, m/z (CI, relative intensity, %): 322 ([M+H]⁺, 100), 182 (27). HRMS Calcd for C₁₈H₃₅NSi₂ ([M+H]⁺): 322.2386. Found: 322.2383.

3-Methyl-2-[(triethylsilyl)methyl|pyridine (2b).

2b was obtained by flash column chromatography on silica-gel (R_f = 0.23 in hexane/EtOAc = 12/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent



in 75% yield (83 mg) as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ : 0.57 (q, J = 7.9 Hz, 6H), 0.88 (t, J = 7.9 Hz, 9H), 2.25 (s, 3H), 2.36 (s, 2H), 6.90 (dd, J = 7.3, 5.0 Hz, 1H), 7.33 (d, J = 7.8 Hz, 1H), 8.28 (dd, J = 5.0, 1.4 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.9, 7.3, 19.9, 22.5, 119.4, 129.6, 137.1, 146.5, 160.7. IR (ATR): 2951 w, 2909 w, 2874 w. MS, m/z (CI, relative intensity, %): 222 ([M+H]⁺, 100), 192 (12). HRMS Calcd for C₁₃H₂₄NSi ([M+H]⁺): 222.1678. Found: 222.1676.

3-Phenyl-2-[(triethylsilyl)methyl]pyridine (2c).

2c was obtained by flash column chromatography on silica-gel (R_f = 0.54 in hexane/EtOAc = 5/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent

in 61% yield (86.0 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.37 (q, J = 7.8 Hz, 6H), 0.69 (t, J = 7.8 Hz, 9H), 2.49 (s, 2H), 7.03 (dd, J = 7.5, 4.9 Hz, 1H), 7.32-7.36 (m, 3H), 7.39-7.43 (m, 3H), 8.42 (dd, J = 4.9, 1.8 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.8, 7.2, 22.4, 119.3, 127.3, 128.5, 129.5,

135.8, 137.3, 140.9, 147.9, 159.7. IR (ATR): 2952 w, 2874 w. MS, *m/z* (CI, relative intensity, %): 284 ([M+H]⁺, 100), 254 (40). HRMS Calcd for C₁₈H₂₆NSi ([M+H]⁺): 284.1835. Found: 284.1832.

4-Isopropyl-2-[(triethylsilyl)methyl]pyridine (2d) and 2-[Bis(triethylsilyl)methyl]-4-isopropylpyridine (2d').

After reacting **1d** (68.5 mg, 0.507 mmol) under the standard reaction conditions, the reaction mixture was passed through a flash silica-gel column to separate **2d** (88.5 mg, $R_f = 0.19$ in hexane/EtOAc = 20/1) and **2d'** (21.8 mg, $R_f = 0.43$ in hexane/EtOAc = 20/1), although both products still contained some silicon impurities.

4-Isopropyl-2-[(triethylsilyl)methyl]pyridine (2d).

The crude mixture containing **2d** (88.5 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2d** in 64% yield (81.2 mg) as a colorless oil. 1 H NMR (CDCl₃) δ : 0.54 (q, J = 7.9 Hz, 6H), 0.90 (t, J = 7.9 Hz, 9H), 1.22 (d, J

= 6.9 Hz, 6H), 2.33 (s, 2H), 2.80 (septet, J = 6.9 Hz, 1H), 6.82 (d, J = 2.3 Hz, 2H), 8.29-8.30 (m, 1H). ¹³C NMR (CDCl₃) δ : 3.4, 7.3, 23.2, 25.4, 33.6, 117.7, 120.5, 149.0, 157.3, 161.5. IR (ATR): 2955 w, 2908 w, 2874 w. MS, m/z (EI, relative intensity, %): 249 (M⁺, 4), 248 (12), 234 (11), 221 (21), 220 (100), 205 (10), 204 (11), 165 (15). HRMS Calcd for C₁₅H₂₇NSi ([M+H]⁺): 250.1991. Found: 250.1987.

2-[Bis(triethylsilyl)methyl]-4-isopropylpyridine (2d').

The crude mixture containing **2d'** (21.8 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2d'** in 4% yield (8.2 mg) as a colorless oil. 1 H NMR (CDCl₃) δ : 0.58-0.61 (m, 12H), 0.88 (t, J = 7.8 Hz, 18H), 1.21 (d, J = 6.9 Hz, 6H), 2.10 (s, 1H), 2.78 (septet, J = 6.9 Hz, 1H), 6.75 (dd, J =

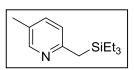
5.0, 1.8 Hz, 1H), 6.78 (s, 1H), 8.25 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 4.9, 7.9, 23.3, 26.3, 33.6, 117.0, 121.5, 148.8, 156.6, 163.8. IR (ATR): 2953 w, 2911 w, 2875 w. MS, m/z (CI, relative intensity, %): 364 ([M+H]⁺, 100), 334 (28). HRMS Calcd for C₂₁H₄₂NSi ([M+H]⁺): 364.2856. Found: 364.2854.

5-Methyl-2-[(triethylsilyl)methyl]pyridine (2e) and 2-[Bis(triethylsilyl)methyl]-5-methylpyridine (2e').

After reacting **1e** (53.6 mg, 0.50 mmol) under the standard reaction conditions, the reaction mixture was passed through a flash silica-gel column to separate **2e** (68.2 mg, $R_f = 0.20$ in hexane/EtOAc = 12/1) and **2e'** (18.2 mg, $R_f = 0.51$ in hexane/EtOAc = 12/1), although both products still contained some silicon impurities.

5-Methyl-2-[(triethylsilyl)methyl]pyridine (2e).

The crude mixture containing 2e (68.2mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure 2e in 60% yield (66.8 mg) as a



colorless oil. ¹H NMR (400 MHz, CDCl₃) δ : 0.53 (q, J = 7.9 Hz, 6H), 0.90 (t, J = 7.9 Hz, 9H), 2.25 (s, 3H), 2.31 (s, 2H), 6.86 (d, J = 7.9 Hz, 1H), 7.30 (dd, J = 7.9, 2.2 Hz, 1H), 8.24 (d, J = 2.2 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.3, 7.4, 18.0, 24.7, 121.8, 128.1, 136.6, 149.2, 158.4. IR (ATR): 2952 w, 2909 w, 2874 w. MS, m/z (CI, relative intensity, %): 222 ([M+H]⁺, 100), 192 (12). HRMS Calcd for C₁₃H₂₄NSi ([M+H]⁺): 222.1678. Found: 222.1677.

2-[Bis(triethylsilyl)methyl]-5-methylpyridine (2e').

The crude mixture containing **2e'** (18.2 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id \times 5 cm h) using 120 mL of hexane as the eluent to give pure **2e'** in 10% yield (17.4 mg) as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ : 0.54-0.61 (m, 12H), 0.88 (t, J

= 8.0 Hz, 18H), 2.06 (s, 1H), 2.23 (s, 3H), 6.80 (d, J = 7.9 Hz, 1H), 7.22 (dd, J = 7.9, 2.2 Hz, 1H), 8.20 (d, J = 2.2 Hz, 1H). 13 C NMR (CDCl₃) δ : 4.8, 7.9, 18.1, 25.3, 122.8, 127.2, 136.2, 149.1, 160.8. IR (ATR): 2952 m, 2910 w, 2874 w. MS, m/z (CI, relative intensity, %): 336 ([M+H]⁺, 100), 306 (29). HRMS Calcd for C₁₉H₃₈NSi₂ ([M+H]⁺): 336.2543. Found: 336.2539.

5-Phenyl-2-[(triethylsilyl)methyl]pyridine (2f) and 2-[Bis(triethylsilyl)methyl]-5-phenylpyridine (2f').

After reacting **1f** (84.4 mg, 0.50 mmol) under the standard reaction conditions, the reaction mixture was passed through a flash silica-gel column to separate **2f** (94.2 mg, $R_f = 0.20$ in hexane/EtOAc = 12/1) and **2f'** (30.2 mg, $R_f = 0.43$ in hexane/EtOAc = 12/1), although both products still contained some silicon impurities.

5-Phenyl-2-[(triethylsilyl)methy])pyridine (2f).

The crude mixture containing **2f** (94.2 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id \times 5 cm h) using 120 mL of hexane as the eluent to give pure **2f** in 66% yield (93.7 mg) as a

colorless oil. ¹H NMR (CDCl₃) δ : 0.58 (q, J = 8.0 Hz, 6H), 0.93 (t, J = 8.0 Hz, 9H), 2.41 (s, 2H), 7.04 (d, J = 8.2 Hz, 1H), 7.35 (t, J = 7.3 Hz, 1H), 7.45 (t, J = 7.5 Hz, 2H), 7.56 (d, J = 7.3 Hz, 2H), 7.70 (dd, J = 8.2, 2.3 Hz, 1H), 8.67 (d, J = 2.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.3, 7.4, 25.2, 122.2, 126.8, 127.6, 129.0, 132.0, 134.3, 138.2, 147.4, 160.7. IR (ATR): 2951 w, 2908 w, 2874 w. MS, m/z (CI, relative intensity, %): 284 ([M+H]⁺, 100), 254 (11). HRMS Calcd for C₁₈H₂₆NSi ([M+H]⁺): 284.1835. Found: 284.1834.

2-[Bis(triethylsilyl)methyl]-5-phenylpyridine (2f').

The crude mixture containing **2f'** (30.2 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2f'** in 15% yield (29.0 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.56-0.67 (m, 12H), 0.90 (t, J = 7.8

Hz, 18H), 2.16 (s, 1H), 6.97 (d, J = 8.0 Hz, 1H), 7.34 (t, J = 7.5 Hz, 1H), 7.44 (t, J = 7.5 Hz, 2H), 7.58 (d, J = 7.5 Hz, 2H), 7.65 (d, J = 8.0 Hz, 1H), 8.65 (s, 1H). ¹³C NMR (CDCl₃) δ : 4.9, 7.9, 25.9, 123.2, 126.7, 127.4, 129.0, 131.0, 133.7, 138.4, 147.1, 163.2. IR (ATR): 2950 w, 2909 w, 2874 w. MS, m/z (CI, relative intensity, %): 398 ([M+H]⁺, 100), 368 (19). HRMS Calcd for C₂₄H₄₀NSi₂ ([M+H]⁺): 398.2699. Found: 398.2700.

N,N-Dimethyl-2-[(triethylsilyl)methyl]pyridin-5-amine (2g).

After reacting **1g** (68.3 mg, 0.5 mmol) under the standard reaction conditions, the reaction mixture was passed through a flash silica-gel column to separate **2g** (76.7 mg, $R_f = 0.14$ in hexane/EtOAc = 6/1) and **2g'** (29.8 mg, $R_f = 0.49$ in hexane/EtOAc = 6/1), although both products still contained some silicon impurities.

N, N-Dimethyl-2-[(triethylsilyl)methyl]pyridin-5-amine (2g).

The crude mixture containing 2g (76.7 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure 2g in 57% yield (71.8 mg)

as a colorless oil. ¹H NMR (CDCl₃) δ : 0.53 (q, J = 7.9 Hz, 6H), 0.91 (t, J = 7.9 Hz, 9H), 2.23 (s,

2H), 2.90 (s, 6H), 6.82 (d, J = 8.7 Hz, 1H), 6.92 (dd, J = 8.7, 3.2 Hz, 1H), 8.00 (d, J = 3.2 Hz, 1H). 13 C NMR (CDCl₃) δ : 3.4, 7.5, 23.5, 40.8, 120.7, 122.1, 134.7, 143.5, 149.5. IR (ATR): 2950 w, 2873 w. MS, m/z (EI, relative intensity, %): 250 (M⁺, 20), 235 (27), 222(21), 221(100), 82(14). HRMS Calcd for $C_{14}H_{26}N_{2}Si$ (M⁺): 250.1865. Found: 250.1862.

2-[Bis(triethylsilyl)methyl]-N,N-dimethylpyridin-5-amine (2g').

The crude mixture containing **2g'** (29.8 mg) was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2g'** in 15% yield (26.6 mg) as a colorless oil. However, since **2g'** gradually decomposed during handling, it was not possible to collect spectroscopic spectra, except for ¹H NMR. Therefore, we modified the procedure for producing **2g'** as the major product by using an excess amount of HSiEt₃ and norbornene. Details are described below.

2-[Bis(triethylsilyl)methyl]-N,N-dimethylpyridin-5-amine (2g').

2g' was produced by the present Ir₄(CO)₁₂-catalyzed reaction of **1a** (68.1 mg, 0.50 mmol) with HSiEt₃ (480 mL, 3.00 mmol) in the presence of norbornene (282 mg, 3.00 mmol) at 160 °C for 4 days. The reaction mixture was passed through a flash silica-gel column to

$$\begin{picture}(20,10) \put(0,0){\line(1,0){100}} \put(0,0){\line(1,0){10$$

separate **2g** (6.6 mg, $R_f = 0.14$ in hexane/EtOAc = 6/1) and **2g'** (108.3 mg, $R_f = 0.49$ in hexane/EtOAc = 6/1), although both products still contained some silicon impurities. The crude mixture containing **2g'** was subsequently filtered through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent to give pure **2g'** in 57% yield (104.1 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.54-0.62 (m, 12H), 0.88 (t, J = 8.0 Hz, 18H), 1.98 (s, 1H), 2.89 (s, 6H), 6.77 (d, J = 8.7 Hz, 1H), 6.87 (dd, J = 8.7, 3.2 Hz, 1H), 7.98 (d, J = 3.2 Hz, 1H). ¹³C NMR (CDCl₃) δ : 4.9, 7.9, 23.7, 40.8, 120.3, 123.0, 134.5, 142.8, 151.8. IR (ATR): 2950 m, 2874 m. MS, m/z (CI, relative intensity, %): 365 ([M+H]⁺, 100), 364 (M⁺, 24), 335(18). HRMS Calcd for C₁₈H₃₅NSi₂ ([M+H]⁺): 365.2808. Found: 365.2804.

5-Phenoxy-2-[(triethylsilyl)methyl]pyridine (2h).

2h was obtained by flash column chromatography on silica-gel ($R_f = 0.50$ in hexane/EtOAc = 5/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane

as the eluent in 39% yield (59.0 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.59 (q, J = 7.9 Hz,

6H), 0.95 (t, J = 7.9 Hz, 9H), 2.37 (s, 2H), 6.96-6.99 (m, 3H), 7.11 (t, J = 7.4 Hz, 1H), 7.19 (dd, J = 8.7, 2.7 Hz, 1H), 7.34 (t, J = 7.4 Hz, 2H), 8.26 (d, J = 2.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.4, 7.4, 24.6, 118.1, 122.7, 123.4, 126.9, 129.9, 140.9, 150.2, 156.8, 157.5. IR (ATR): 2952 w, 2874 w. MS, m/z (CI, relative intensity, %): 300 ([M+H]⁺, 100), 270 (14). HRMS Calcd for C₁₈H₂₆NOSi ([M+H]⁺): 300.1784. Found: 300.1783.

2-[1-(Triethylsilyl)ethyl]pyridine (2i).

2i was obtained by flash column chromatography on silica-gel ($R_f = 0.31$ in hexane/EtOAc = 5/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 71% yield (78.1 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.54 (q, J =

8.0 Hz, 6H), 0.87 (t, J = 8.0 Hz, 9H), 1.43 (d, J = 7.5 Hz, 3H), 2.57 (q, J = 7.5 Hz, 1H), 6.95-7.01 (m, 2H), 7.50 (td, J = 7.5, 1.7 Hz, 1H), 8.45 (d, J = 5.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 2.3, 7.6, 14.5, 30.2, 119.5, 121.6, 135.8, 148.9, 166.2. IR (ATR): 2952 m, 2909 w, 2874 m. MS, m/z (CI, relative intensity, %): 222 ([M+H]⁺, 100), 192 (16). HRMS Calcd for C₁₃H₂₄NSi ([M+H]⁺): 222.1678. Found: 222.1675.

2-[1-(Triethylsilyl)hexyl]pyridine (2j).

2j was obtained by flash column chromatography on silica-gel ($R_f = 0.43$ in hexane/EtOAc = 30/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 58% yield (80.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.52 (q, J = 7.8

$$\begin{array}{|c|c|}\hline \\ N & SiEt_3 \\\hline \\ C_5H_{11} \\\hline \end{array}$$

Hz, 6H), 0.86-0.89 (m, 12H), 1.15-1.22 (m, 6H), 1.65-1.71 (m, 1H), 2.03-2.08 (m, 1H), 2.44 (dd, J = 12.1, 3.0 Hz, 1H), 6.95-7.00 (m, 2H), 7.49 (td, J = 7.7, 2.0 Hz, 1H), 8.47 (d, J = 3.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 2.6, 7.6, 14.2, 22.7, 29.2, 31.8, 37.5, 119.4, 122.5, 135.6, 149.1, 164.8. IR (ATR): 2952 w, 2928 w, 2874 w. MS, m/z (EI, relative intensity, %): 277 (M⁺, 7), 276 (24), 246 (23), 248 (100), 221 (11), 220 (55), 192 (44), 120 (10), 115 (60), 106 (11), 93 (11), 87 (63), 59 (38). HRMS Calcd for $C_{17}H_{32}NSi$ ([M+H]⁺): 278.2304. Found: 278.2300.

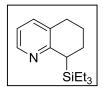
2-[2-Phenyl-1-(triethylsilyl)ethyl]pyridine (2k).

2k was obtained by flash column chromatography on silica-gel (R_f = 0.16 in hexane/EtOAc = 50/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 57% yield (84.8 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.60 (q, J = 7.7

Hz, 6H), 0.91 (t, J = 7.7 Hz, 9H), 2.77 (dd, J = 11.9, 3.1 Hz, 1H), 3.04 (dd, J = 14.5, 3.1 Hz, 1H), 3.42 (dd, J = 14.5, 11.9 Hz, 1H), 6.86 (d, J = 7.7 Hz, 1H), 6.91-6.93 (m, 1H), 7.02-7.07 (m, 3H), 7.12 (t, J = 7.6 Hz, 2H), 7.39 (td, J = 7.7, 1.8 Hz, 1H), 8.47 (d, J = 3.7 Hz, 1H). ¹³C NMR (CDCl₃) 8: 2.7, 7.6, 35.3, 39.3, 119.6, 123.1, 125.6, 128.1, 128.5, 135.6, 142.9, 149.1, 163.6. IR (ATR): 2950 w, 2909 w, 2874 w. MS, m/z (EI, relative intensity, %): 297 (M⁺, 13), 296 (29), 269 (20), 268 (77), 240 (26), 239 (13), 238 (59), 192 (38), 183 (27), 182 (100), 121 (15), 120 (15), 115 (14), 106 (20), 91 (12), 87 (51), 59 (32). HRMS Calcd for $C_{19}H_{28}NSi$ ([M+H]⁺): 298.1991. Found: 298.1992.

8-(Triethylsilyl)-5,6,7,8-tetrahydroquinoline (21).

21 was obtained by flash column chromatography on silica-gel ($R_f = 0.27$ in hexane/EtOAc = 12/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 73% yield (92.1 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.61 (q, J = 7.9 Hz, 6H),



0.86 (t, J = 7.9 Hz, 9H), 1.60-1.70 (m, 1H), 1.83-1.88 (m, 2H), 2.02-2.07 (m, 1H), 2.66-2.74 (m, 3H), 6.89 (dd, J = 7.3, 4.8 Hz, 1H), 7.24-7.26 (m, 1H), 8.27 (d, J = 4.8 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.6, 22.6, 25.2, 29.5, 29.7, 119.3, 131.3, 136.2, 146.5, 161.1. IR (ATR): 2947 w, 2911 w, 2873 w. MS, m/z (EI, relative intensity, %): 247 (M⁺, 19), 246 (66), 219 (32), 218 (100), 205 (26), 191 (23), 190 (12), 288 (11), 163 (29), 132 (25), 87 (37), 59 (32). HRMS Calcd for C₁₅H₂₅NSi (M⁺): 247.1756. Found: 247.1750.

2-[(Triethylsilyl)methyl]quinoline (2m).

2m was obtained by flash column chromatography on silica-gel (R_f = 0.24 in hexane/EtOAc = 10/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane

as the eluent in 66% yield (85.6 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.58 (q, J = 8.0 Hz, 6H), 0.93 (t, J = 8.0 Hz, 9H), 2.57 (s, 2H), 7.11 (d, J = 8.2 Hz, 1H), 7.42 (t, J = 7.7 Hz, 1H), 7.63

(t, J = 7.7 Hz, 1H), 7.72 (d, J = 8.2 Hz, 1H), 7.95 (d, J = 7.7 Hz, 2H). ¹³C NMR (CDCl₃) δ : 3.5, 7.4, 26.9, 121.8, 124.9, 125.9, 127.5, 128.5, 129.2, 135.6, 148.2, 162.3. IR (ATR): 2951 w, 2908 w, 2874 w. MS, m/z (CI, relative intensity, %): 258 ([M+H]⁺, 100), 228 (8). HRMS Calcd for $C_{16}H_{24}NSi$ ([M+H]⁺): 258.1678. Found: 258.1675.

7-Methoxy-1-[(triethylsilyl)methyl]-9*H*-pyrido[3,4-*b*]indole (2n).

2n was obtained by flash column chromatography on silica-gel ($R_f = 0.06$ in hexane/EtOAc = 5/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 86% yield (140.4 mg)

as a pale yellow solid. Mp = 106-108 °C. ¹H NMR (CDCl₃) δ : 0.60 (q, J = 7.8 Hz, 6H), 0.89 (t, J = 7.8 Hz, 9H), 2.58 (s, 2H), 3.91 (s, 3H), 6.88 (dd, J = 8.7, 2.3 Hz, 1H), 6.96 (d, J = 2.3 Hz, 1H), 7.59 (d, J = 5.5 Hz, 1H), 7.93 (d, J = 8.7 Hz, 1H), 8.01 (br s, 1H), 8.26 (d, J = 5.5 Hz, 1H). ¹³C NMR (CDCl₃) δ : 4.0, 7.4, 21.2, 55.8, 94.9, 109.7, 110.7, 116.3, 122.7, 128.3, 133.9, 139.2, 141.5, 144.9, 160.7. IR (ATR): 3064 w, 2952 w, 2873 w, 1629 m. MS, m/z (CI, relative intensity, %): 327 ([M+H]⁺, 100), 326 (11), 297(8), 213(7). HRMS Calcd for C₁₉H₂₇N₂OSi ([M+H]⁺): 327.1893. Found: 327.1889.

2-[(Triethylsilyl)methyl]thiazole (20).

20 was obtained by flash column chromatography on NH₂-modified silicagel ($R_f = 0.26$ in hexane/EtOAc = 50/1) in 53% yield (57.3 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.60 (q, J = 8.0 Hz, 6H), 0.94 (t, J = 8.0

Hz, 9H), 2.59 (s, 2H), 7.04 (d, J = 3.4 Hz, 1H), 7.57 (d, J = 3.4 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.2, 7.3, 20.2, 116.8, 142.1, 169.4. IR (ATR): 2952 w, 2909 w, 2875 w. MS, m/z (EI, relative intensity, %): 185 (17), 184 (100), 170 (12), 157 (14), 156 (36), 129 (20), 128 (22), 115 (11), 87 (70), 86 (21), 59 (45), 58 (17). HRMS Calcd for C₁₀H₁₉NSSi (M⁺): 213.1007. Found: 284.1009.

2-[(Triphenylsilyl)methyl]pyridine (3a).

3a was obtained by flash column chromatography on silica-gel ($R_f = 0.19$ in hexane/EtOAc = 10/1) and subsequent filtration through a pad of NH₂-modified silica-gel (1.8 cm id × 5 cm h) using 120 mL of hexane as the eluent in 54% yield (95.7 mg) as a white solid. Mp = 61-63 °C. ¹H

NMR (CDCl₃) δ : 3.22 (s, 2H), 6.68 (d, J = 8.3 Hz, 1H), 6.93 (t, J = 6.2 Hz, 1H), 7.31-7.34 (m, 7H), 7.38-7.42 (m, 3H), 7.46 (dd, J = 6.6, 1.1 Hz, 6H), 8.35 (d, J = 4.6 Hz, 1H). ¹³C NMR (CDCl₃) δ : 27.1, 119.8, 123.4, 127.9, 129.7, 134.2, 135.8, 136.1, 149.0, 159.5. IR (ATR): 3066 w, 3042 w, 3013 w, 2925 w, 2853 w. MS, m/z (EI, relative intensity, %): 351 (M⁺, 5), 350 (40), 275 (17), 274 (100), 259 (55), 181 (15). HRMS Calcd for $C_{24}H_{21}NSi$ ([M+H]⁺): 351.1522. Found: 352.1525.

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Chapter 3

The Cationic Iridium-Catalyzed $C(sp^3)$ -H Silylation of 2-Alkyl-1,3-azoles at the α -Position in the 2-Alkyl Group Leading to 2-(1-Silylalkyl)-1,3-azoles

3.1 Introduction

As mentioned in Chapters 1 and 2, the author found that the $Ir_4(CO)_{12}$ complex is effective for the regioselective $C(sp^3)$ -H silylation of alkylpyridines at the α -position in the C2- or C4-substituent, which appeared to proceed via the formation of a key intermediate, an electrophilic silicon species, formed by the reaction of $Ir_4(CO)_{12}$ with hydrosilanes, in the catalytic cycle. On the other hand, the reaction of 2-methyloxazole, which represents a class of 5-membered heterocyclic compounds, gave a complex mixture. The author's next focus was on the use of other complexes that have the ability to electrophilically activate a hydrosilane derivatives, and were examined in the regioselecitive α - $C(sp^3)$ -H silylation of 2-alkyloxazoles and other 2-alkyl-1,3-azoles. Among the reported complexes that were examined, the author was pleased to find that Brookhart's complex, (POCOP'Bu)IrHCl, proved to be an ideal candidate. Chapter 3 described the $(POCOP'Bu)IrHCl/NaBAr^F_4$ -catalyzed regioselective $C(sp^3)$ -H silylation of 2-alkyl-1,3-azoles at the α -position in a 2-alkyl group leading to the production of 2-(1-silylalkyl)-1,3-azole derivatives (Scheme 1).

Scheme 1. (POCOP'Bu)IrHCl/NaBAr^F₄-Catalyzed Regioselective C(sp³)-H Silylation of 2-Alkyl-1,3-azoles with Hydrosilanes

cat. (POCOP^tBu)IrHCl

NaBAr^F₄

3,5-dimethylpyridine cyclopentene

$$X = O, NR''', S$$

3.2 Results and Discussion

Table 1 shows the results for the reaction of 2-methylbenzoxazaole (1a) with triethylsilane under optimized reaction conditions with (POCOP'Bu)IrHCl and NaBAr^F₄ as a catalyst system, and under several different sets of reaction conditions. When 3,5-dimethylpyridine and cyclopentene, the latter of which acted as a hydrogen acceptor, were added to the reaction mixture, the silylated compound 2a was produced in 98% yield, as determined from the ¹H NMR spectrum, and was isolated in 90% yield by silica-gel column chromatography (entry 1). The results cited in entries 2 and 3 indicate that a cationic iridium complex generated in situ is the actual active species for the reaction.

Table 1. Optimization of the Reaction Conditions^a

cat.
$$(POCOP^tBu)IrHCI$$

$$NaBAr_4^F$$

$$3,5-dimethylpyridine$$
cyclopentene

2a

antw.	deviation from the standard reaction conditions	yields $(\%)^b$		
entry		2a	1a	
1	None	98 (90) ^c	0	
2	NaB(C ₆ F ₅) ₄ instead of NaBAr ^F	89	8	
3	without NaBAr ^F	0	90	
4	without 3,5-dimethylpyridine	0	67	
5	without cyclopentane	61	20	
6	cyclohexene instead of cyclopentene	93	7	
7	^t BuCH=CH ₂ instead of cyclopentene	83	0	
8	norbornene instead of cyclopentene	20	67	
9	at 60 °C	4	94	
10	$\operatorname{Ir}_{4}(\operatorname{CO})_{12}{}^{d,e}$	0	81	
11	$Cp*IrCl[\eta^2-2-(2-Py)Ph]/NaBAr_4^{F_d}$	trace	75	
12	$B(C_6F_5)_3^d$	0	99	
13	FeCl ₂ /NaBAr ^F ₄ ^d	0	98	
14	$Zn(OTf)_2$	0	99	
15	(POCOP ^t Bu)RhHCl/NaBAr ^F ₄	trace	81	

^aReaction conditions: **1a** (0.3 mmol), HSiEt₃ (1.2 mmol), (POCOP'Bu)IrHCl (0.0075 mmol), NaBAr^F₄ (0.0075 mmol), 3,5-dimethylpyridine (0.06 mmol), and cyclopentene (0.6 mmol) at 80 °C for 20 h under N₂. ^bYields of **2a** and **2a**' were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. The yield of the unreacted **1a** was determined from the GC spectrum with tridecane as an internal standard. ^cThe number shown in parentheses is the isolated yield. ^dThe complex was used as a catalyst instead of (POCOP'Bu)IrHCl/NaBAr^F₄. ^eIr₄(CO)₁₂ (0.0019 mmol)

The addition of 3,5-dimethylpyridine was necessary for the reaction to proceed (entry 4). The fact that the reaction occurred, even in the absence of cyclopentene, indicates that the hydrogen atoms at the α-position and that on the silicon atom were released in the form of H₂ (entry 5). Nonetheless, the addition of other hydrogen acceptors, such as cyclohexene and *tert*-butylethylene, also resulted in better product yields, compared to the case in which they were not present (entries 6 and 7). However, in the presence of norbornene, the reaction was retarded (entry 8). Ir₄(CO)₁₂ was not an effective catalyst in this reaction (entry 10). Other complexes, such as Cp*IrCl[η²-2-(2-Py)Ph]/NaBAr^F₄,² B(C₆F₅)₃,³ FeCl₂/NaBAr^F₄,⁴ and Zn(OTf)₂,⁵ did not catalyze the reaction, although they are also known to activate hydrosilanes in a manner similar to the (POCOP'Bu)IrHCl/NaBAr^F₄ catalyst. When the iridium in the catalyst was replaced with rhodium, only trace amounts of 2a were produced (entry 15).

Some other hydrosilanes were also examined in the reaction of **1a** (Scheme 2). Although other trialkylsilanes such as HSiEt₂Me and HSiEtMe₂ were found to be applicable to the present reaction, the final reaction was comprised of mixtures of monosilylated products (**3a** and **4a**) and disilylated products (**3a** and **4a**). However, proper tuning of the reaction conditions in the reaction with HSiEt₂Me, by decreasing the amount of HSiEt₂Me to 0.6 mmol, using NaB(C₆F₅)₄ and cyclohexene instead of NaBAr^F₄ and cyclopentene, respectively, and at a lower reaction temperature of 60 °C, permitted the monosilylated product **3a** to be selectively produced in 87% yield (Scheme 3). On the other hand, the reactions with HSiPhMe₂ and HSi(OSiMe₃)Me₂ gave complex reaction mixtures containing only small amounts of the desired products.

Scheme 2. Screening of hydrosilanes^a

cat. (POCOP^tBu)IrHCl NaBAr
$$_{4}^{F}$$
 3,5-dimethylpyridine cyclopentene 2a-4a 2a'-4a'

HSiEt $_{3}$ 98% (2a) 0% HSiEt $_{2}$ Me 21% (3a) 72% (3a') HSiEtMe $_{2}$ 17% (4a) 73% (4a')

"Reaction conditions: **1a** (0.3 mmol), hydrosilane (1.2 mmol), (POCOP'Bu)IrHCl (0.0075 mmol), NaBAr^F₄ (0.0075 mmol), 3,5-dimethylpyridine (0.06 mmol), and cyclopentene (0.6 mmol) at 80 °C for 20 h under N₂. Yields of products were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard.

Scheme 3. Tuning the Reaction Conditions When Using HSiEt₂Me^a

^aReaction conditions: **1a** (0.3 mmol), HSiEt₂Me (0.6 mmol), (POCOP'Bu)IrHCl (0.0075 mmol), NaB(C₆F₅)₄ (0.0075 mmol), 3,5-dimethylpyridine (0.06 mmol), and cyclohexene (0.6 mmol) at 60 °C for 20 h under N₂. Yields of products were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard.

A number of 2-alkyl-1,3-azoles were examined under the optimized reaction conditions (Table 2). None of the methyl groups on the phenyl ring in 2-methylbenzoxazole participated in the silylation and **2b-2d** were produced, in good yields, expect for **2b**. Whereas the naphthalenefused oxazole derivatives **1e** and **1f** reacted under the reaction conditions to give **2e** and **2f** in 88% and 80% yields, respectively, the reaction of the [1,2-d]-fused oxazole **1g** resulted in the complete recovery of the starting material at 80 °C, and, gave **2g** in 31% yield even at the elevated reaction temperature of 160 °C. The reaction of the ethyl-substituted benzoxazole **1h** took place regioselectively at the α-position in the butyl group to produce **2h** in 81%. However, no reaction occurred when 2-isopropylbenzoxazole was used as the substrate. Functional groups, such as methoxy (**2i**), bromo (**2k**), chloro (**2k**), and methoxycarbonyl (**2l**) groups, were all tolerated in the reaction. Although the reaction of 2-methyloxazole gave a complex reaction mixture, probably due to instability of the starting material and/or the desired product, 2,4-dimethyloxazole could be converted into **2m** in 52% yield. To my delight, 1,2-dimethylbenzimidazole (**1n**) and thiazole derivatives (**1o-1p**) were also applicable to the present reaction, giving **2n-2p**.

A plausible mechanism for the regioselective silylation of 2-alkyl-1,3-azoles in the absence of a hydrogen acceptor, which would be simpler than that in the presence of an acceptor, is shown in Scheme 4. A cationic iridium species **I** is formed by the in situ reaction of (POCOP'Bu)IrHCl with NaBAr^F₄. A hydrosilane is next activated by **I** via the formation of an η^1 -hydrosilane complex **II**.⁶ The silyl group in **II** is abstracted by the added pyridine derivative to form a silylpyridinium species **III** and a dihydridoiridium complex **IV**,⁷ from the latter of which, the reductive elimination of H₂ occurs to form **V**. The coordination of the pyridine derivative to the

Table 2. Scope of Substrates^a

^aReaction conditions: **1** (0.3 mmol), HSiEt₃ (1.2 mmol), (POCOP^tBu)IrHCl (0.0075 mmol), NaBAr^F₄ (0.0075 mmol), 3,5-dimethylpyridine (0.06 mmol), and cyclopentene (0.6 mmol) at 80 °C for 20 h under N₂. Isolated yields are given. ^bAt 100 °C. ^cAt 120 °C. ^dAt 130 °C. ^eAt 160 °C.

iridium center in **IV** might accelerate the elimination of H₂, as was reported for the generation of H₂ from **IV**, which proceeded smoothly under an atmosphere of CO.⁸ The transfer of the silyl group from **III** to the 2-alkyl-1,3-azole substrate occurs to form an *N*-silylazolinium species **VI**.⁹ The subsequent deprotonation of **VI** by **V** then affords a 2-methylene-*N*-silyl-2,3-dihydroazole, with regeneration of **I**. Finally, the migration of the silyl group on the nitrogen atom to the methylene carbon atom produces the desired product. The reason for the low product yields in the reactions of **1b** and **1g** can be attributed to steric factors in case of the formation of **VI**.

Scheme 4. A Plausible Reaction Mechanism in the Absence of a Hydrogen Acceptor

In the case of the silvlation in the presence of a hydrogen acceptor, the hydrogen atoms at the α-position in the substrate and that on the silicon atom are removed from the reaction via the formation of alkane. One possibility involves the hydrogen atoms being trapped via the hydrogenation of the acceptor with the dihydridoiridium complex IV, as shown in Scheme 4. As shown in Table 1, the presence of added norbornene resulted in a lower product yield than that with other alkenes, even in its absence. Although the author has no information regarding the reactivity of IV toward norbornene, it is difficult to conclude that hydrogenation of norbornene with IV under the present reaction conditions would not occur. Another possibility is that the hydrogen acceptor participates in an earlier stage of the reaction, prior to the formation of V. Given these considerations, a proposed reaction mechanism for the reaction under the above conditions is shown in Scheme 5. First, the monohydridoiridium complex I reacts with the hydrogen acceptor to afford an alkyl complex VII, which activates the hydrosilane via the formation of VIII. Hartwig and a co-worker proposed a reaction mechanism in which two hydrogen atoms are trapped by the acceptor alkene starting from the insertion of the alkene into the Rh-H bond in the monohydridorhodium complex, and not the dihydridorhodium complex. 10 Bulkier substituents on the silicon atom and the hydrogen acceptor would make the formation of VIII more difficult, due to steric repulsion, thus leading to a lower reaction rate. The abstraction of the silvl group from VIII by the added pyridine occurs to form III and the complex IX. The subsequent reductive elimination of an alkane from IX gives V. The remaining reaction path

from **III** to the final product and the regeneration of **I** is the same as that shown in Scheme 4. However, the possibility that the mechanism involves the hydrogenation of the acceptor with **IV** cannot be completely ruled out at the present time.

Scheme 5. A Plausible Reaction Mechanism in the Presence of a Hydrogen Acceptor

3.3 Conclusion

The author demonstrated that a cationic iridium complex, formed by the reaction of (POCOP'Bu)IrHCl with NaBAr^F₄, catalyze the regioselective C(sp³)-H silylation of 2-alkyl-1,3-azoles to produce 2-(1-silylalkyl)-1,3-azoles. Although the reaction proceeded both in the presence and absence of a hydrogen acceptor, the presence of an added acceptor gave better results, in terms of the efficiency of the reaction. The proposed reaction mechanism involves the formation of an electrophilic silicon species as a key intermediate

3.4 Experimental Section

General Information.

¹H and ¹³C NMR spectra were recorded on 400 and 100 MHz spectrometers, respectively, using CDCl₃ and acetone- d_6 as solvents. Data were recorded as follows: chemical shifts in ppm (δ), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, c = complex), coupling constants (Hz), and integration. Infrared spectra (IR) were recorded by an ATR technique. Absorption data were reported in reciprocal centimeters with the following relative intensities: s (strong), m (medium), or w (weak). Mass spectra were obtained using a spectrometer with a

quadrupole mass analyzer at 70 eV. High-resolution mass spectra (HRMS) were obtained using a spectrometer with a double-focusing mass analyzer. Analytical gas chromatography (GC) was carried out on a chromatograph equipped with a flame ionization detector. Melting points were determined on a capillary point apparatus equipped with a digital thermometer and were uncorrected.

Materials.

1b, **1e-g**, **1j-k**, **1m-n**, and **1p** were purchased and were used as received. Other organic compounds, except **1d**, **1h**, **1l**, and 2-isopropylbenzo[*d*]oxazole were purchased from commercial sources and distilled over CaH₂ before use. Compounds **1d**, **1h**, **1l**, and 2-isopropylbenzo[*d*]oxazole were prepared as described below. Complexes (POCOP'Bu)IrHCl, ¹¹ (POCOP'Bu)RhHCl, ¹² NaBAr^F₄, ¹³ and NaB(C₆F₅)₄ ¹⁴ were prepared by following the procedure described in the literature. Other metal complexes were purchased from commercial sources and used without further purification.

2,6-Dimethylbenzo[d]oxazole (1d).¹⁵

The procedure reported by Hartwig¹⁶ was modified by using 2-amino-5-methylphenol (1.00 g, 8.12 mmol) in place of 2-amino-4-methoxy-phenol to produce **1d**, which was isolated by distillation under reduced pressure

(42 °C/52 mmHg) in 74% yield (0.88 g, 5.98 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ: 2.42 (s, 3H), 2.57 (s, 3H), 7.06 (m, 1H), 7.22 (d, J = 1.0 Hz, 1H), 7.49 (d, J = 8.2 Hz, 1H). ¹³C NMR (CDCl₃) δ: 14.4, 21.5, 110.3, 118.6, 125.1, 134.5, 139.2, 151.1, 163.0.

2-Ethylbenzo[d]oxazole (1h).¹⁷

The procedure reported by Hartwig¹⁶ was modified by using 2-aminophenol (1.50 g, 13.7 mmol) and 1,1,1-trimethoxy-propane (1.89 g, 13.7 mmol) in place of 2-amino-4-methoxyphenol and trimethyl orthoacetate, respectively,

to produce **1h**, which was isolated by distillation under reduced pressure (61 °C/84 mmHg) in 74% yield (1.49 g, 10.1 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 1.44 (t, J = 7.6 Hz, 3H), 2.94 (q, J = 7.6 Hz, 2H), 7.27 (tt, J = 5.5, 1.9 Hz, 2H), 7.44-7.46 (m, 1H), 7.65-7.68 (m, 1H). ¹³C NMR (CDCl₃) δ : 10.9, 22.2, 110.2, 119.5, 124.0, 124.4, 141.4, 150.8, 168.1.

Methyl 2-methylbenzo[d]oxazole-5-carboxylate (11).¹⁸

The procedure reported by Hartwig¹⁶ was modified by using methyl 3-amino-4-hydroxybenzoate¹⁹ (0.61g, 3.65 mmol), in place of 2-amino-4-methoxyphenol to produce **11**, which was isolated by flash column chromatography on silica-gel ($R_f = 0.09$ in hexane/EtOAc = 10/1) in

68% yield (52.6 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 2.68 (s, 3H), 3.95 (s, 3H), 7.51 (d, J = 8.6 Hz, 1H), 8.06 (dd, J = 8.6, 1.6 Hz, 1H), 8.34 (s, 1H). ¹³C NMR (actetone- d_6) δ : 14.3, 52.5, 110.1, 121.5, 127.0, 127.4, 142.7, 154.7, 166.4, 166.8.

2-Isopropylbenzo[d]oxazole.²⁰

The procedure reported by Hartwig¹⁵ was modified by using 2-aminophenol (0.87 g, 7.98 mmol) and 1,1,1-trimethoxy-2-methylpropane (1.19 g, 8.03 mmol) in place of 2-amino-4-methoxyphenol and trimethyl orthoacetate,

respectively, to produce 2-isopropylbenzo[d]oxazole, which was isolated by distillation under reduced pressure (68 °C/54 mmHg) in 70% yield (0.90 g, 5.58 mmol) as a colorless oil. ¹H NMR (CDCl₃) δ : 1.45 (d, J = 7.1 Hz, 6H), 3.19-3.27 (m, 1H), 7.27 (t, J = 3.7 Hz, 2H), 7.46 (dd, J = 7.3, 3.7 Hz, 1H), 7.67-7.69 (m, 1H). ¹³C NMR (CDCl₃) δ : 20.4, 28.9, 110.3, 119.7, 124.0, 124.4, 141.3, 150.7, 171.3.

Typical Procedure for the (POCOP'Bu)IrHCl/NaBAr^F4-Catalyzed C-H Silylation of 2-Alkyl-1,3-azoles with Hydrosilanes.

An 8 mL Schlenk tube was flame-dried and purged with N_2 . After the tube was cooled to room temperature, (POCOP'Bu)IrHCl (4.7 mg, 7.5 x 10^{-3} mmol), NaBAr^F₄ (6.6 mg, 7.5 x 10^{-3} mmol), 3,5-dimethylpyridine (6.4 mg, 0.06 mmol), 1 (0.3 mmol), cyclopentene (40.8 mg, 0.6 mmol), and hydrosilane (1.2 mmol) were placed in the Schlenk tube. The reaction mixture was stirred at 80 °C for 20 h. After cooling the mixture to room temperature, the volatiles were removed in vacuo. The product was isolated by flash column chromatography on NH₂-modified silica gel.

2-[(Triethylsilyl)methyl]benzo[d]oxazole (2a).

2a was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.17$ in hexane/EtOAc = 40/1) in 90% yield (66.8 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.54 (q, J = 7.8 Hz, 6H), 0.96 (t, J =

7.8 Hz, 9H), 2.42 (s, 2H), 7.20-7.28 (m, 2H), 7.41-7.44 (m, 1H), 7.59-7.61 (m, 1H). ¹³C NMR (CDCl₃) δ: 3.5, 7.2, 15.0, 109.9, 118.9, 123.6, 123.9, 142.2, 150.9, 167.4. IR (ATR): 2953 w, 2876 w. MS, *m/z* (EI, relative intensity, %): 247 (M⁺, 4), 246 (12), 219 (17), 218 (100), 191 (11), 190 (62), 162 (16), 87 (67), 86 (19), 59 (35). HRMS Calcd for C₁₄H₂₁NOSi (M⁺): 247.1392; found: 247.1396.

4-Methyl-2-[(triethylsilyl)methyl]benzo[d]oxazole (2b).

2b was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.21$ in hexane/EtOAc = 40/1) in 67% yield (52.6 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.65 (q, J = 8.1 Hz, 6H), 0.96 (t, J = 8.1 Hz, 9H), 2.43 (s, 2H), 2.57 (s, 3H), 7.04 (d, J = 7.3 Hz, 1H), 7.10 (d,

J = 7.8 Hz, 1H), 7.24 (d, J = 8.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.6, 7.2, 15.1, 16.7, 107.3, 123.2, 124.5, 129.2, 141.2, 150.6, 166.6. IR (ATR): 2953 w, 2876 w. MS, m/z (EI, relative intensity, %): 261 (M⁺, 16), 246 (21), 233 (20), 232 (100), 204 (27), 115 (23), 88 (12), 87 (97), 86 (20), 59 (41). HRMS Calcd for C₁₅H₂₄NOSi ([M+H]⁺): 262.1627; found: 262.1624.

5-Methyl-2-[(triethylsilyl)methyl]benzo[d]oxazole (2c).

2c was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.26 in hexane/ EtOAc = 40/1) in 94% yield (72.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.64 (q, J = 8.0 Hz, 6H), 0.95

(t, J = 8.0 Hz, 9H), 2.40 (s, 2H), 2.44 (s, 3H), 7.02 (d, J = 8.3 Hz, 1H), 7.27 (d, J = 8.3 Hz, 1H), 7.38 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.6, 7.2, 15.1, 21.5, 109.3, 119.0, 124.6, 133.6, 142.4, 149.1, 167.5. IR (ATR): 2953 w, 2876 w. MS, m/z (EI, relative intensity, %): 261 (M⁺, 6), 233 (17), 232 (100), 205 (11), 204 (59), 177 (11), 176 (14), 87 (57), 86 (14), 59 (31). HRMS Calcd for C₁₅H₂₃NOSi (M⁺): 261.1549; found: 261.1549.

6-Methyl-2-[(triethylsilyl)methyl]benzo[d]oxazole (2d).

2d was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.19 in hexane/ EtOAc = 40/1) in 86% yield (67.5 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.63 (q, J = 8.0 Hz, 6H), 0.95

(t, J = 8.0 Hz, 9H), 2.38 (s, 2H), 2.44 (s, 3H), 7.05 (d, J = 8.2 Hz, 1H), 7.22 (s, 1H), 7.45 (d, J = 8.2 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.2, 14.9, 21.7, 110.2, 118.2, 124.9, 133.8, 139.9, 151.1,

166.7. IR (ATR): 2953 w, 2876 w. MS, *m/z* (EI, relative intensity, %): 261 (M⁺, 16), 260 (17), 233 (22), 232 (100), 205 (16), 204 (58), 177 (16), 176 (19), 87 (50), 86 (16), 59 (28). HRMS Calcd for C₁₅H₂₄NOSi ([M+H]⁺): 262.1627; found: 262.1627.

2-[(Triethylsilyl)methyl]naphtho[2,1-d]oxazole (2e).

2e was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.20 in hexane/ EtOAc = 40/1) in 88% yield (78.5 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.67 (q, J = 8.0 Hz, 6H), 0.98 (t, J = 8.0 Hz, 9H), 2.52 (s, 2H), 7.44 (t, J = 7.6 Hz, 1H), 7.56 (t, J =

7.3 Hz, 1H), 7.68 (d, J = 8.7 Hz, 1H), 7.71 (d, J = 8.7 Hz, 1H), 7.91 (d, J = 7.8 Hz, 1H), 8.13 (d, J = 8.2 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.2, 15.1, 118.3, 119.8, 120.2, 124.5, 124.9, 126.6, 128.6, 131.0, 138.4, 146.2, 166.5. IR (ATR): 2953 w, 2875 w. MS, m/z (EI, relative intensity, %): 297 (M⁺, 34), 296 (20), 269 (27), 268 (100), 241 (14), 240 (44), 214 (13), 213 (16), 212 (19), 115 (10), 87 (49), 86 (15), 59 (35). HRMS Calcd for $C_{18}H_{24}NOSi$ ([M+H]⁺): 298.1627; found: 298.1623.

2-[(Triethylsilyl)methyl]naphtho[2,3-d]oxazole (2f).

2f was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.18$ in hexane/ EtOAc = 40/1) in 80% yield (71.4 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.67 (q, J = 7.9 Hz, 6H),

0.97 (t, J = 7.9 Hz, 9H), 2.48 (s, 2H), 7.41-7.48 (m, 2H), 7.80 (s, 1H), 7.89-7.92 (m, 1H), 7.93-7.96 (m, 1H), 8.00 (s, 1H). ¹³C NMR (CDCl₃) δ : 3.6, 7.2, 15.6, 105.6, 115.7, 124.5, 125.0, 127.9, 128.4, 131.0, 131.3, 142.3, 150.0, 170.0. IR (ATR): 2953 w, 2875 w. MS, m/z (EI, relative intensity, %): 297 (M⁺, 33), 296 (17), 269 (26), 268 (100), 241 (16), 240 (59), 213 (15), 212 (18), 210 (16), 183 (17), 115 (13), 114 (13), 87 (45), 86 (15), 59 (29). HRMS Calcd for C₁₈H₂₄NOSi ([M+H]⁺): 298.1627; found: 298.1628.

2-[(Triethylsilyl)methyl]naphtho[1,2-d]oxazole (2g).

2g was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.21 in hexane/ EtOAc = 40/1) in 34% yield (30.3 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.67 (q, J = 7.9 Hz, 6H), 0.97 (t, J = 7.9 Hz, 9H), 2.53 (s, 2H), 7.48 (ddd, J = 8.2, 6.9, 0.9 Hz, 1H),

7.59 (d, J = 8.9 Hz, 1H), 7.60 (ddd, J = 8.2, 6.9, 0.9 Hz, 1H), 7.68 (d, J = 8.7 Hz, 1H), 7.92 (d, J = 8.2 Hz, 1H), 8.45 (d, J = 8.2 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.7, 7.2, 15.2, 110.6, 122.2, 124.3, 125.0, 126.1, 126.6, 128.5, 131.0, 137.2, 147.9, 166.5. IR (ATR): 2952 w, 2875 w. MS, m/z (EI, relative intensity, %): 298 (M⁺, 16), 297 (65), 296 (71), 282 (34), 281 (50), 269 (19), 268 (79), 267 (24), 240 (16), 212 (11), 210 (14), 166 (14), 115 (23), 106 (14), 88 (11), 87 (100), 86 (25), 59 (52). HRMS Calcd for $C_{18}H_{24}NOSi$ ([M+H]⁺): 298.1627; found: 298.1627.

2-[1-(Triethylsilyl)ethyl]benzo[d]oxazole (2h).

2h was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.26 in hexane/EtOAc = 40/1) in 81% yield (63.5 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.65 (q, J = 7.9 Hz, 6H), 0.94 (t, J =

7.9 Hz, 9H), 1.53 (d, J = 7.4 Hz, 3H), 2.70 (q, J = 7.4 Hz, 1H), 7.20-7.27 (m, 2H), 7.43 (d, J = 7.5 Hz, 1H), 7.62 (d, J = 7.5 Hz, 1H). ¹³C NMR (CDCl₃) δ : 2.5, 7.3, 12.6, 21.5, 109.9, 119.0, 123.6, 123.9, 142.0, 150.6, 171.5. IR (ATR): 2954 w, 2876 w. MS, m/z (EI, relative intensity, %): 261 (M⁺, 7), 260 (22), 233 (22), 232 (100), 205 (13), 204 (47), 177 (10), 115 (16), 87 (80), 86 (18), 59 (36). HRMS Calcd for C₁₅H₂₄NOSi ([M+H]⁺): 262.1627; found: 262.1629.

5-Methoxy-2-[(triethylsilyl)methyl]benzo[d]oxazole (2i).

2i was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.09$ in hexane/ EtOAc = 40/1) in 88% yield (73.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.64 (q, J = 7.8 Hz, 6H),

0.96 (t, J = 7.8 Hz, 9H), 2.39 (s, 2H), 3.83 (s, 3H), 6.81 (dd, J = 8.7, 2.5 Hz, 1H), 7.11 (d, J = 2.5 Hz, 1H), 7.28-7.31 (m, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.2, 15.1, 56.0, 102.5, 109.9, 111.6, 143.0, 145.5, 157.0, 168.3. IR (ATR): 2953 w, 2876 w. MS, m/z (EI, relative intensity, %): 277 (M⁺, 19), 276 (10), 249 (21), 248 (100), 221 (14), 220 (59), 207 (11), 193 (14), 192 (19), 87 (51), 86 (15), 59 (32). HRMS Calcd for C₁₅H₂₃NO₂Si (M⁺): 277.1498; found: 277.1498.

5-Bromo-2-[(triethylsilyl)methyl]benzo[d]oxazole (2j).

2j was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.20$ in hexane/ EtOAc = 40/1) in 84% yield (82.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.63 (q, J = 7.9 Hz, 6H),

0.95 (t, J = 7.9 Hz, 9H), 2.41 (s, 2H), 7.28 (d, J = 8.4 Hz, 1H), 7.33 (dd, J = 8.4, 1.8 Hz, 1H),

7.72 (d, J = 1.8 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.6, 7.2, 15.2, 111.1, 116.7, 121.9, 126.6, 143.8, 149.9, 168.9. IR (ATR): 2953 w, 2876 w. MS, m/z (EI, relative intensity, %): 326 (M⁺[⁸¹Br], 12), 324 (M⁺[⁷⁹Br], 11), 299 (15), 298 (75), 297 (15), 296 (73), 270 (39), 268 (39), 242 (12), 240 (10), 115 (33), 114 (10), 88 (10), 87 (100), 86 (30), 59 (39), 58 (10). HRMS Calcd for C₁₄H₂₁⁸¹BrNOSi ([M+H]⁺): 326.0576; found: 326.0574.

5-Chloro-2-[(triethylsilyl)methyl]benzo[d]oxazole (2k).

2k was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.24$ in hexane/ EtOAc = 40/1) in 89% yield (75.3 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.64 (q, J = 8.0 Hz, 6H),

0.95 (t, J = 8.0 Hz, 9H), 2.41 (s, 2H), 7.18 (dd, J = 8.5, 2.1 Hz, 1H), 7.32 (d, J = 8.5 Hz, 1H), 7.57 (d, J = 2.1 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.1, 15.2, 110.5, 118.9, 123.8, 129.3, 143.3, 149.4, 169.0. IR (ATR): 2954 w, 2876 w. MS, m/z (EI, relative intensity, %): 283 (M⁺[³⁷Cl], 5), 283 (M⁺[³⁵Cl], 1), 280 (11), 254 (29), 253 (16), 252 (81), 226 (17), 225 (10), 224 (49), 196 (11), 115 (28), 87 (100), 86 (28), 59 (41). HRMS Calcd for C₁₄H₂₁³⁷ClNOSi [M+H]⁺: 282.1081; found: 282.1084.

Methyl 2-[(triethylsilyl)methyl[benzo[d]oxazole-5-carboxylate (21).

21 was obtained by flash column chromatography on NH₂-modified silica-gel ($R_f = 0.17$ in hexane/EtOAc = 40/1) in 58% yield (53.1 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.65 (q.

J = 8.0 Hz, 6H), 0.96 (t, J = 8.0 Hz, 9H), 2.45 (s, 2H), 3.94 (s, 3H), 7.45 (d, J = 8.6 Hz, 1H), 8.00 (dd, J = 8.6, 1.7 Hz, 1H), 8.28 (d, J = 1.7 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.5, 7.2, 15.2, 52.4, 109.7, 120.8, 125.9, 126.4, 142.2, 153.8, 167.1, 168.9. IR (ATR): 2953 w, 2877 w, 1722 m, 1287 m. MS, m/z (EI, relative intensity, %): 277 (M⁺, 21), 276 (100), 249 (12), 248 (56), 220 (11), 160 (14), 159 (18), 115 (15), 87 (62), 86 (15), 59 (31). HRMS Calcd for C₁₆H₂₄NO₃Si ([M+H]⁺): 306.1525; found: 306.1523.

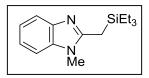
4-Methyl-2-[(triethylsilyl)methyl]oxazole (2m).

2m was obtained by flash column chromatography on NH₂-modified silicagel ($R_f = 0.14$ in hexane/EtOAc = 40/1) in 52% yield (30.9 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.58 (q, J = 8.0 Hz, 6H), 0.94 (t, J = 8.0

Hz, 9H), 2.10 (s, 3H), 2.20 (s, 2H), 7.18 (c, 1H). 13 C NMR (CDCl₃) δ : 3.5, 7.1, 11.7, 14.2, 133.0, 136.1, 164.3. IR (ATR): 2952 w, 2876 w. MS, m/z (EI, relative intensity, %): 182 (85), 154 (47), 126 (19), 115 (11), 87 (100), 86 (17), 59 (47). HRMS Calcd for $C_{11}H_{21}NOSi$ (M⁺): 211.1392; found: 211.1391.

1-Methyl-2-[(triethylsilyl)methyl]-1*H*-benzo[*d*]imidazole (2n).

2n was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.11 in hexane/EtOAc = 1/1) in 52% yield (41.0 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.64 (q, J = 7.9 Hz, 6H), 0.93 (t, J = 7.9 Hz, 9H), 2.38 (s, 2H), 3.67 (s, 3H), 7.17-7.24 (c, 3H), 7.63-7.67



(c, 1H). 13 C NMR (CDCl₃) δ : 3.7, 7.3, 13.6, 30.2, 108.6, 118.5, 121.2, 121.6, 135.8, 143.2, 155.2. IR (ATR): 2951 w, 2874 w. MS, m/z (EI, relative intensity, %): 260 (M⁺, 32), 259 (13), 245 (27), 232 (28), 231 (100), 204 (18), 176 (48), 175 (13), 173 (16) HRMS Calcd for $C_{15}H_{24}N_2Si$ ([M+H]⁺): 260.1709; found: 260.1709.

2-[(Triethylsilyl)methyl]benzo[d]thiazole (20).

20 was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.19 in hexane/EtOAc = 40/1) in 85% yield (67.2 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.65 (q, J = 7.9 Hz, 6H), 0.96 (t, J =

7.9 Hz, 9H), 2.68 (s, 2H), 7.26 (td, J = 7.7, 1.9 Hz, 1H), 7.38 (td, J = 7.7, 1.9 Hz, 1H), 7.75 (d, J = 8.0 Hz, 1H), 7.87 (d, J = 8.0 Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.4, 7.3, 21.8, 121.2, 121.8, 124.0, 125.7, 135.3, 153.7, 170.3. IR (ATR): 2952 w, 2874 w. MS, m/z (EI, relative intensity, %): 263 (M⁺, 7) 262 (12), 248 (23), 236 (10), 235 (20), 234 (100), 206 (22), 87 (44), 86 (17), 59 (25). HRMS Calcd for C₁₄H₂₂NSSi ([M+H]⁺): 264.1242; found: 264.1243.

2-[(Triethylsilyl)methyl]thiazole (2p).

2p was obtained by flash column chromatography on NH₂-modified silica-gel (R_f = 0.26 in hexane/EtOAc = 50/1) in 73% yield (47.0 mg) as a colorless oil. ¹H NMR (CDCl₃) δ : 0.60 (q, J = 8.0 Hz, 6H), 0.94 (t, J = 8.0 Hz, 9H), 2.59 (s,

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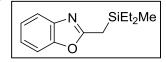
2H), 7.04 (d, J = 3.4, Hz, 1H), 7.57 (d, J = 3.4, Hz, 1H). ¹³C NMR (CDCl₃) δ : 3.2, 7.3, 20.2, 116.8, 142.1, 169.4. IR (ATR): 2952 w, 2875 w. MS, m/z (EI, relative intensity, %): 213 (M⁺, 0.2), 198 (40), 185 (17), 184 (100), 170 (12), 157 (14), 156 (36), 129 (20), 128 (22), 115 (11),

2-[(Diethylmethylsilyl)methyl]benzo[d]oxazole (3a) and 2-[Bis(diethylmethylsilyl)methyl]benzo[d]oxazole (3a').

After reacting **1a** (68.3 mg, 0.3 mmol) with HSiEt₂Me (122.7 mg, 1.2 mmol) under the standard reaction conditions, the reaction mixture was passed through an NH₂-modified silica-gel column to isolate **3a** in 17% yield (12.1 mg, $R_f = 0.17$ in hexane/ EtOAc = 40/1) and **3a**' in 69% yield (68.9 mg, $R_f = 0.49$ in hexane/ EtOAc = 40/1), as colorless oils, respectively.

2-[(Diethylmethylsilyl)methyl]benzo[d]oxazole (3a).

¹H NMR (CDCl₃) δ: 0.08 (s, 3H), 0.63 (q, J = 7.9 Hz, 4H), 0.96 (t, J = 7.9 Hz, 9H), 2.42 (s, 2H), 7.20-7.22 (c, 1H), 7.41-7.43 (c, 1H), 7.59-7.61 (c, 1H). ¹³C NMR (CDCl₃) δ: -5.8, 5.2, 7.1, 16.7, 109.9, 118.9,



123.7, 123.9, 142.2, 150.9, 167.8. IR (ATR): 2954 w, 2876 w. MS, m/z (EI, relative intensity, %): 233 (M⁺, 13), 232 (28), 218 (10), 205 (19), 204 (100), 190 (10), 177 (26), 176 (45), 150 (12), 101 (15), 73 (77), 72 (25), 45 (26). HRMS Calcd for $C_{13}H_{20}NOSi$ ([M+H]⁺): 234.1314; found: 234.1310. **3a** was produced as a sole product by the reaction under the revised reaction conditions shown in Scheme 3, and isolated in 69% yield.

2-[Bis(diethylmethylsilyl)methyl]benzo[d]oxazole (3a').

¹H NMR (CDCl₃) δ: 0.13 (s, 6H), 0.49-0.68 (c, 8H), 0.88 (t, J = 7.9 Hz, 6H), 0.93 (t, J = 7.9 Hz, 6H), 2.22 (s, 1H), 7.17-7.26 (c, 2H), 7.40 (d, J = 7.5 Hz, 1H), 7.57 (d, J = 7.5 Hz, 1H). ¹³C NMR (CDCl₃) δ: -

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4.5, 6.1, 6.3, 7.4, 7.5, 18.2, 109.7, 118.7, 123.1, 123.7, 142.5, 150.8, 169.6. IR (ATR): 2953 w, 2876 w. MS, m/z (EI, relative intensity, %): 333 (M⁺, 10), 332 (13), 306 (10), 305 (30), 304 (100), 277 (16), 276 (62), 248 (11), 246 (12), 218 (12), 101 (53), 87 (14), 73 (84), 59 (18). HRMS Calcd for $C_{18}H_{31}NSi_2$ (M⁺): 333.1944; found: 333.1940.

2-[(Ethyldimethylsilyl)methyl]benzo[d]oxazole (4a) and <math>2-[Bis(ethyldimethylsilyl)methyl]-benzo[d]oxazole (4a').

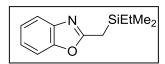
After reacting **1a** (68.3 mg, 0.3 mmol) with HSiEtMe₂ (105.9 mg, 1.2 mmol) under the standard reaction conditions, the reaction mixture was passed through an NH₂-modified silica-gel column to isolate **4a**' in 66% yield (61.0 mg, $R_f = 0.49$ in hexane/ EtOAc = 40/1) as a colorless oil. However, no **4a** was obtained, probably due to decomposition during chromatographic

separation.

1a was also reacted with HSiEtMe₂ under the reaction conditions shown in Scheme 3, instead of HSiEt₂Me, to afford a mixture of **4a** and **4a'** in 41% and 51% yields, respectively, which were determined from ¹H NMR spectra with 1,3-dihydro-2-benzofuran as an internal standard. After the reaction, the reaction mixture was passed through an NH₂-modified silica-gel column to isolate **4a** in 3% yield (8.0 mg, $R_f = 0.17$ in hexane/ EtOAc = 40/1) and **4a'** in 47% yield (43.1 mg, $R_f = 0.49$ in hexane/ EtOAc = 40/1), as colorless oils, respectively.

2-[(Ethyldimethylsilyl)methyl]benzo[d]oxazole (4a).

¹H NMR (CDCl₃) δ: 0.11 (s, 6H), 0.63 (q, J = 8.0 Hz, 2H), 0.98 (t, J = 8.0 Hz, 3H), 2.43 (s, 2H), 7.21-7.29 (c, 2H), 7.42-7.46 (c, 1H), 7.60 (dd, J = 7.1, 2.1 Hz, 1H). ¹³C NMR (CDCl₃) δ: -3.5, 7.0, 7.2, 18.4,



119.3, 110.0, 119.0, 123.7, 123.9, 142.1, 150.9, 167.3. IR (ATR): 2955 w, 2876 w. MS, m/z (EI, relative intensity, %): 219 (M⁺, 25), 218 (68), 204 (42), 191 (28), 190 (38), 176 (38), 150 (23), 135 (21), 87 (25), 86 (10), 59 (100), 58 (22). HRMS Calcd for $C_{12}H_{18}NOSi$ ([M+H]⁺): 220.1158; found: 220.1156.

2-[Bis(ethyldimethylsilyl)methyl]benzo[d]oxazole (4a').

¹H NMR (CDCl₃) δ: 0.11 (s, 6H), 0.14 (s, 6H), 0.56 (c, 4H), 0.91 (t, J = 8.1 Hz, 6H), 2.18 (s, 2H), 7.17-7.26 (c, 2H), 7.38-7.41 (c, 1H), 7.57-7.59 (c, 1H). ¹³C NMR (CDCl₃) δ: -2.4, -2.2, 7.3, 8.1, 20.8,

$$\begin{picture}(20,0) \put(0,0){\line(1,0){100}} \put(0,0){\line(1,0){100$$

109.7, 118.7, 123.1, 123.7, 142.5, 150.8, 169.6. IR (ATR): 2954 w, 2876 w. MS, m/z (EI, relative intensity, %): 305(M $^+$, 20), 304 (30), 290 (33), 277 (21), 276 (64), 262 (14), 249 (12), 248 (43), 232 (17), 222 (17), 208 (14), 204 (13), 178 (10), 87 (68), 73 (33), 59 (100). HRMS Calcd for $C_{16}H_{27}NOSi_2$ (M $^+$): 305.1631; found: 305.1629.

5.5 References and Notes

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Conclusion

Some new types of the $C(sp^3)$ -H bond silylations catalyzed by $Ir_4(CO)_{12}$ or $(POCOP'Bu)IrHCl/NaBAr^F_4$ was developed and reported herein. The key to success for these reactions is the generation of an electrophilic silicon species as a key intermediate in the catalytic cycle.

Chapter 1 contains a discussion of the iridium-catalyzed regioselective $C(sp^3)$ -H silylation of 4-alkypyridines at the α -position with hydrosilanes. The low product yields of 2-substituted 4-methylpyridines were improved markedly by the addition of a catalytic amounts of 3,5-dimethylpyridine. This is the rare example of non-chelation-assisted $C(sp^3)$ -H bond silylations with hydrosilanes in transition metal catalysis. An unprecedented C4 selectivity was observed when a polyalkyl-substituted pyridine such as 2,4-dimethylpyridine was used as a substrate.

Chapter 2 deals with the iridium-catalyzed silylation of C(sp³)-H bonds in 2-alkylpyridine derivatives with hydrosilanes. The addition of 3,5-dimethylpyridine was also essential for the success of the reaction.

Chapter 3 is concerned with the iridium-catalyzed $C(sp^3)$ -H silylation of 2-alkyl-1,3-azoles at the α -position in the alkyl group. Screening of complexes as catalysts that have the ability to electrophilically activate the hydrosilane, revealed that a cationic pincer iridium complex (POCOP'Bu)IrHCl showed a high degree of catalytic activity.

A number of catalytic silylations of $C(sp^2)$ -H bonds, involving the generation of electrophilic silicon species followed by an electrophilic attack on the silicon atom, such as Friedel-Crafts type C-H silylation, have been reported to date. On the other hand, to the best my knowledge, this is the first example of the electrophilic silylation of $C(sp^3)$ -H bonds. These findings provide a promising starting point for the further development of a variety of the electrophilic silylation of $C(sp^3)$ -H bonds.

One advantage of the present reactions, compared to the conventional method in which organometallic reagents such as Grignard reagents are used, is its broader functional group compatibility. A variety of functional groups, even bromide and chloride, did not affect the reaction. This result shows that the reactions reported in this thesis provide a simple and useful method for the synthesis of various types of benzylated silyl compounds.

List of Publications

- (1) Iridium-Catalyzed Regioselective C(sp³)–H Silylation of 4-Alkylpyridines at the Benzylic Position with Hydrosilanes Leading to 4-(1-Silylalkyl)pyridines Yoshiya Fukumoto, Masaya Hirano, Naoto Chatani.

 ACS Catal. 2017, 7, 3152.
- (2) Ir₄(CO)₁₂-Catalyzed Benzylic C(sp³)-H Silylation of 2-Alkylpyridines with Hydrosilanes Leading to 2-(1-Silylalkyl)pyridines Yoshiya Fukumoto, <u>Masaya Hirano</u>, Nao Matsubara, Naoto Chatani. *J. Org. Chem.* 2017, 82, 13649.
- (3) An Cationic Iridium-Catalyzed C(sp³)-H Bond Silylation of 2-Alkyl-1,3-azoles at the α-Position in the 2-Alkyl Group Leading to 2-(1-Silylalkyl)-1,3-azoles

 <u>Masaya Hirano</u>, Yoshiya Fukumoto, Nao Matsubara, Naoto Chatani. *Chem. Lett.* in press.

Supplementary List of Publication

(1) Rhodium-Catalyzed Anti-Markovnikov Hydrohydrazination of Terminal Alkynes with *N*-Alkyl- and *N*,*N*-Dialkylhydrazines
Yoshiya Fukumoto, Akihiro Ohmae, Masaya Hirano, Naoto Chatani. *Asian J. Org. Chem.* **2013**, *2*, 1036.