

Title	Control on the density and optical properties of color centers at SiO <sub>2</sub> /SiC interfaces by oxidation and annealing
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1 **Control on the density and optical properties of color centers at**  
2 **SiO<sub>2</sub>/SiC interfaces by oxidation and annealing**

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11  
12 Color centers in solids can serve as single photon emitters (SPEs) that are important in many quantum  
13 applications. Silicon carbide (SiC) is a promising host for color centers because of its well-established  
14 crystal growth and device technologies. Although color centers with extremely high brightness were  
15 found at the silicon dioxide (SiO<sub>2</sub>)/SiC interface, controlling their density and optical properties  
16 remains a challenge. In this study, we demonstrate control over the color centers at the SiO<sub>2</sub>/SiC  
17 interface by designing the oxidation and annealing conditions. We report that post-oxidation CO<sub>2</sub>  
18 annealing has the ability to reduce the color centers at the interface and form well-isolated SPEs with  
19 bright emission. We also discuss the correlation between the color centers and electrically active  
20 defects.

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1 Solid-state single photon emitters (SPEs) find various applications in quantum computing<sup>1</sup>, quantum  
2 cryptography<sup>2</sup>, and quantum sensing<sup>3</sup>. Optically active point defects (i.e., color centers) in wide  
3 bandgap semiconductors can serve as SPEs. Furthermore, if the color center possesses a non-zero  
4 spin ground state, it may act as a spin-to-photon interface<sup>2,4</sup>. The nitrogen-vacancy center (NV center)  
5 in diamond is a leading example; the spin state of NV centers can be optically initialized, manipulated,  
6 and detected even at room temperature<sup>5-7</sup>. Group-IV color centers such as silicon-vacancy (SiV)<sup>8</sup> and  
7 germanium-vacancy (GeV)<sup>9</sup> centers in diamond are also attractive because of their extremely high  
8 Debye-Waller factor and sharp zero phonon transition.

9 Silicon carbide (SiC) is a wide gap semiconductor that shares favorable properties for quantum  
10 technology with diamond<sup>10,11</sup>. Wafer-scale crystal growth, well-controlled *n*- and *p*-type doping, and  
11 mature device technologies make this material attractive as a host for SPEs<sup>12,13</sup>. A number of color  
12 centers have been found in bulk SiC so far, e.g. silicon-vacancy ( $V_{\text{Si}}$ )<sup>14,15</sup>, divacancy ( $V_{\text{Si}}V_{\text{C}}$ )<sup>16,17</sup>,  
13 carbon antisite-vacancy complex ( $C_{\text{Si}}V_{\text{C}}$ )<sup>18</sup>, and nitrogen-vacancy center ( $N_{\text{C}}V_{\text{Si}}$ )<sup>19-21</sup>. Among them,  
14 coherent control of defect spins for  $V_{\text{Si}}$ <sup>15</sup>,  $V_{\text{Si}}V_{\text{C}}$ <sup>17</sup>, and  $N_{\text{C}}V_{\text{Si}}$ <sup>21</sup> have been demonstrated at room  
15 temperature. While these defects are important candidates for qubits, color centers with extremely  
16 bright emission have also been found at the SiO<sub>2</sub>/SiC interface<sup>22-26</sup>. These centers exhibit single  
17 photon emission within the visible spectrum at room temperature, with a count rate exceeding that of  
18 the NV center in diamond<sup>23,24</sup>. Moreover, they can be readily implemented in semiconductor devices,  
19 enabling control of their luminescence by electrical current or voltage<sup>23,24</sup>. However, achieving  
20 control over their density and optical properties is not straightforward. A previous study suggested  
21 that stable color centers start to form during the oxidation of SiC at 550°C, but their density remains  
22 almost constant within a wide oxidation temperature range of 700–1100°C<sup>22</sup>. It was also noted that  
23 the color center density is oxide thickness dependent<sup>25</sup>, but the magnitude of the change is rather small.  
24 This may impede the formation of spatially well-separated SPEs at the SiO<sub>2</sub>/SiC interface.  
25 Additionally, achieving controllability over the optical properties is challenging. Unstable color  
26 centers are frequently found at the interface unless the process conditions are carefully designed<sup>22,25</sup>.  
27 It is also known that the emission spectrum from the defects strongly varies among them, even within  
28 the very same sample chip<sup>23-25</sup>. Understanding the microscopic origin of defects is crucial for better  
29 control over them. Although carbon defects<sup>27-31</sup> and/or oxygen defects (i.e., defect complexes in  
30 SiC<sup>32,33</sup> and oxygen vacancies in SiO<sub>2</sub><sup>34,35</sup>) in SiO<sub>2</sub>/SiC structures are possible candidates, the details  
31 are still unclear. Investigating the correlation between optically and electrically active defects should  
32 provide a hint regarding the origin of defects. Such attempts have been made in this regard<sup>25,26</sup>, but

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1 further investigation involving samples processed under various conditions is needed to deepen the  
2 understanding.

3 In this study, we aim to control the density and optical properties of color centers at the SiO<sub>2</sub>/SiC  
4 interface by designing the oxide formation conditions. In particular, we investigate the impact of high-  
5 temperature oxidation<sup>36</sup> and post-oxidation annealing in a carbon dioxide (CO<sub>2</sub>) ambient<sup>37</sup>. Both  
6 methods are known to effectively reduce the electrical defects in SiC metal-oxide-semiconductor  
7 (MOS) devices. Although post-oxidation annealing in nitric oxide (NO) ambient is the most common  
8 method to improve the SiO<sub>2</sub>/SiC interface quality, it is known that the NO annealing leads to the  
9 formation of very fast interface states due to the incorporation of a high amount of nitrogen atoms<sup>38</sup>.  
10 We use the high-temperature oxidation and CO<sub>2</sub> annealing to avoid introducing any foreign atoms  
11 into the SiO<sub>2</sub>/SiC system. This enables us to focus on the intrinsic defects in the SiO<sub>2</sub>/SiC structure  
12 (excluding possible defects involving nitrogen donors), which simplifies the situation. It should be  
13 noted that the oxide growth during the CO<sub>2</sub> annealing can be ignored, because the activation energy  
14 of oxidation by CO<sub>2</sub> is much high (7.5 eV) compared with that by O<sub>2</sub> (2.9 eV)<sup>37</sup>. Therefore, the CO<sub>2</sub>  
15 annealing is likely effective in reducing the interface carbon defects and oxide defects while  
16 preventing the oxide growth. We also discuss the correlation between the density of color centers and  
17 electrically active defects to gain insights into the origin of color centers.

18 The flow of sample preparation and measurements is shown in Fig. 1. *N*-type 4H-SiC(0001)  
19 substrates with 5 μm-thick epilayers (donor density: 1.0×10<sup>16</sup> cm<sup>-3</sup>) are used in this study. After wet  
20 cleaning and the removal of surface oxide by 10%-diluted hydrofluoric (HF) acid, we form the  
21 SiO<sub>2</sub>/SiC structures through two different processes: (i) thermal oxidation at 1200–1500°C in 100%  
22 O<sub>2</sub> ambient and (ii) thermal oxidation at 1200°C followed by annealing in CO<sub>2</sub> at 1050–1200°C for  
23 30 min. Note that the samples were cooled down without gas replacement. Hereafter, these samples  
24 are labeled by conditions, e.g., Ox. 1200°C and Ox.+CO<sub>2</sub> 1050°C. The resulting oxide thickness  
25 ranges from 18–34 nm. Optical measurements are performed on SiO<sub>2</sub>/SiC samples, and electrical  
26 measurements are conducted on SiC MOS capacitors with Al gate electrodes. All measurements are  
27 conducted at room temperature.

28 We describe the details of optical measurements. Figure 1(b) shows the schematic of our  
29 homemade confocal microscope. An excitation laser with a wavelength of  $\lambda = 532$  nm is focused onto  
30 the SiO<sub>2</sub>/SiC interface using an objective lens with NA=0.9 (MPLAPON60X, Olympus). The  
31 resulting photoluminescence (PL) is collected by the same objective lens, passed through a dichroic  
32 mirror (550 nm cut-on), and detected either by the spectrometer (HRS-300 and Pixis100BRX,

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1 Teledyne Princeton Instruments) or by the avalanche photodiode-based single photon counting  
2 modules (SPCM-AQRH-14, Excelitas Technologies). The objective lens is placed on top of the  
3 piezoelectric stage, which enables a three-dimensional mapping of the PL intensity. A 600-nm long-  
4 pass filter is inserted after the dichroic mirror to cutoff the Raman lines when taking the PL mapping  
5 image. Hanbury-Brown and Twiss (HBT) measurement<sup>39,40</sup> is also performed to evaluate the single  
6 photon property of the defects. For this measurement, the photons are passed through the 600-nm  
7 long-pass filter, split by the 50:50 beam splitter, and detected by either of the two single photon  
8 counting modules. Then, the difference in the timing of single photon detection by the two modules  
9 is analyzed with the multiscaler (MCS8A, FAST Comtec).

10 Figure 2 shows typical PL mapping images of SiO<sub>2</sub>/SiC samples. A large number of color centers  
11 is observed in the Ox. 1200°C sample (Fig. 2(a)), which is comparatively low in the Ox. 1500°C  
12 sample (Fig. 2(b)). Hence, it seems that high-temperature oxidation is effective in reducing the color  
13 center density. For quantitative discussion, we evaluate the density of color centers with count rates  
14 greater than  $5.0 \times 10^4$  counts/sec through image processing. As a result, color center densities are  
15  $5.2 \times 10^7$ ,  $2.0 \times 10^7$ ,  $1.6 \times 10^7$ , and  $1.2 \times 10^7$  cm<sup>-2</sup> for samples oxidized at 1200, 1300, 1400, and 1500°C,  
16 respectively. Thus, the density decreases monotonically as the oxidation temperature increases.  
17 However, the density stays in the same order of magnitude ( $10^7$  cm<sup>-2</sup>) even when the temperature is  
18 increased up to 1500°C. As explained in the following paragraph, reducing the density of color centers  
19 is crucial to make them act as SPEs. We then take a look into the PL mapping image for the CO<sub>2</sub>-  
20 annealed samples (Figs. 2(c) and 2(d)). We see that the color center density is substantially lower in  
21 the annealed samples compared with the as-oxidized ones (Figs. 2(a) and 2(b)). In particular, when  
22 the CO<sub>2</sub> anneal is performed at a high temperature of 1200°C, only three spots with a count rate higher  
23 than  $5.0 \times 10^4$  counts/sec are observed in the  $30 \times 30$  μm<sup>2</sup> area (Fig. 2(d)). Again, image processing is  
24 performed to quantitatively investigate the color center density as a function of CO<sub>2</sub> anneal  
25 temperature. The resulting color center densities are  $2.9 \times 10^6$ ,  $1.4 \times 10^6$ ,  $5.6 \times 10^5$ , and  $3.3 \times 10^5$  cm<sup>-2</sup> for  
26 samples annealed at 1050, 1100, 1150, and 1200°C, respectively. Compared with the Ox. 1200°C  
27 sample, the color center density becomes two orders of magnitude smaller after CO<sub>2</sub> annealing at  
28 1200°C. Thus, CO<sub>2</sub> annealing has the ability to reduce not only the electrically active defects<sup>37</sup> but  
29 also the color centers at the SiO<sub>2</sub>/SiC interface.

30 Figures 3(a) and 3(b) show the typical PL spectra of the Ox. 1200°C and Ox.+CO<sub>2</sub> 1200°C  
31 samples measured at room temperature. Find all the measured spectra for these samples in the  
32 supplementary information. Note that the two sharp peaks located above 2.2 eV, regardless of the

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1 sample conditions, are the transverse and longitudinal optic Raman signals of 4H-SiC crystal<sup>23,41,42</sup>.  
2 The luminescence of color centers appears as broad signals in the energy range of 1.5 – 2.2 eV. For  
3 the as-oxidized sample (Fig. 3(a)), the shape, intensity, and peak position of PL spectra strongly vary  
4 among the investigated defects, in agreement with previous reports<sup>23,24</sup>. In contrast, the color centers  
5 in the CO<sub>2</sub>-annealed sample (Fig. 3(b)) more frequently exhibit similar PL signals, as represented by  
6 the defects #2–#4. This suggests that CO<sub>2</sub> annealing is effective in reducing the variation in the optical  
7 signals. This reduction could be due to a decrease in the types of color centers and/or to the reduction  
8 in the charge/strain in the local environment around the defect. Then, the single photon properties of  
9 the defects are evaluated through HBT measurements, where typical results are shown in Figs. 3(c)  
10 and 3(d). Note that the second-order correlation function,  $g^{(2)}(\tau)$  is calculated without background  
11 subtraction in this study. While anti-bunching behavior is observed for the investigated defects, the  
12  $g^{(2)}(0)$  values do not reach below 0.5 for several of the defects (e.g., #4) in the Ox. 1200°C sample  
13 (Fig. 3(c)). This means that the color centers in the as-oxidized sample are not isolated SPEs. In fact,  
14 we find that only 12 out of the investigated 18 diffraction limited confocal spots exhibited SPE  
15 characteristics for the Ox. 1200°C sample. The high density of color centers (Fig. 2(a)) may contribute  
16 to the background signal superimposed on the defect signal of interest, thereby degrading the SPE  
17 characteristics. In contrast, the  $g^{(2)}(0)$  values are well below 0.5 for the Ox.+CO<sub>2</sub> 1200°C sample (Fig.  
18 3(d)). We confirm that 11 out of the investigated 11 defects in the Ox.+CO<sub>2</sub> 1200°C sample behave  
19 as SPEs. While the CO<sub>2</sub> anneal leaves defects with relatively low PL intensities (Fig. 3(c)), the count  
20 rates are still very high ( $> 5.5 \times 10^4$  counts/sec) after annealing at 1200°C. Therefore, CO<sub>2</sub> annealing  
21 is effective not only in reducing the number of color centers but also in forming well-isolated and  
22 bright SPEs at the SiO<sub>2</sub>/SiC interface.

23 Finally, we discuss the relationship between the optically and electrically active defects. Figure  
24 4 shows typical capacitance–voltage ( $C$ – $V$ ) characteristics for the Ox. 1200°C and Ox.+CO<sub>2</sub> 1200°C  
25 samples measured at 1 MHz. The gate voltage is swept forward (depletion to accumulation) and then  
26 backward (accumulation to depletion). The Ox. 1200°C sample exhibits a large positive flatband  
27 voltage ( $V_{FB}$ ) shift as well as a significant  $C$ – $V$  hysteresis. These positive  $V_{FB}$  shift and  $C$ – $V$  hysteresis  
28 likely correspond to the effective fixed charge (mostly related to deep acceptor-type interface states)  
29 and the near-interface oxide traps, respectively. The plausible candidates for the interface states and  
30 near-interface oxide traps are carbon-related defects<sup>27–31</sup> and oxygen vacancies<sup>34,35</sup>, respectively. In  
31 the Ox.+CO<sub>2</sub> 1200°C sample, both the  $V_{FB}$  shift and the hysteresis are clearly suppressed. Hence, CO<sub>2</sub>  
32 annealing is effective in passivating the interface defects and near-interface oxide traps, in accordance

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1 with a previous study<sup>37</sup>. In this study, two types of electrical traps were estimated from the  $C$ - $V$   
2 characteristics, i.e., deep interface traps and near-interface oxide traps. The former is evaluated from  
3 the  $V_{FB}$  shift, while the latter is evaluated from the  $C$ - $V$  hysteresis. Here, we reasonably assume that  
4 the positive  $V_{FB}$  shift observed in SiC MOS structures is mainly caused by interface traps with deep  
5 trap levels. The areal density of charges corresponding to the  $V_{FB}$  shift ( $N_{eff}$ ) and  $C$ - $V$  hysteresis ( $N_{hys}$ )  
6 for each sample condition is shown together with the color center density in Fig. 5. We find that the  
7 color center density is more closely related to  $N_{eff}$  than  $N_{hys}$ . This indicates that the color centers  
8 correlate with the deep interface defects rather than the near-interface oxide traps. This does not  
9 contradict with the previous study which stated that the color centers are located at the SiC-side of  
10 the interface<sup>22</sup>. On the other hand, it is notable that the density of color centers is five orders of  
11 magnitude smaller than that of electrically active defects. This means that only a part of the  
12 electrically active defects is detected as color centers. As mentioned earlier, a possible cause for the  
13 observed color centers is the oxidation induced defects consist of carbon and/or oxygen atoms. They  
14 likely induce deep trap levels at the SiO<sub>2</sub>/SiC interface which could be reduced either by high-  
15 temperature oxidation or CO<sub>2</sub> annealing.

16 In summary, we investigated the impact of oxidation and annealing treatment on the density and  
17 optical properties of color centers at the SiO<sub>2</sub>/SiC interface. Although the color center density remains  
18 in the same order ( $10^7$  cm<sup>-2</sup>) even when changing the oxidation temperature of SiC within the range  
19 of 1200°C and 1500°C, we observe a reduction in density by two orders of magnitude through post-  
20 oxidation CO<sub>2</sub> annealing at 1200°C. We find that one-third of the investigated color centers in the as-  
21 oxidized sample at 1200°C indicate  $g^2(0)$  values higher than 0.5 and do not exhibit SPE characteristics.  
22 In contrast, all 11 investigated defects behave as SPEs in the sample annealed in CO<sub>2</sub> at 1200°C.  
23 Therefore, CO<sub>2</sub> annealing not only reduces the density of color centers but also forms well-isolated  
24 SPEs at the SiO<sub>2</sub>/SiC interface. Furthermore, we emphasize that the color center is more closely  
25 associated with the effective fixed charge rather than the near-interface oxide traps, suggesting that  
26 the origin of color centers lies in interface defects with deep trap levels.

27

#### 28 **Supplementary Material**

29 All of the measured PL spectra of Ox. 1200°C and Ox.+CO<sub>2</sub> 1200°C samples can be found in the  
30 supplementary information.

31

#### 32 **Acknowledgments**

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2

3 **AUTHOR DECLARATIONS**

4 **Conflict of Interest**

5 The authors have no conflicts to disclose.

6

7 **DATA AVAILABILITY**

8 The data that support the findings of this study are available from the corresponding authors upon  
9 reasonable request.



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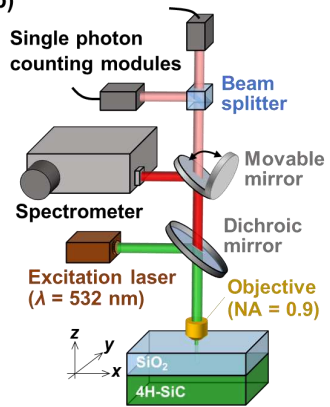
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1 FIGURES

(a)

- ***n*-type 4H-SiC(0001)**  
( $N_D: \sim 1.0 \times 10^{16} \text{ cm}^{-3}$ )
- **Wet cleaning & Removal of surface oxide**
- **Oxidation & Annealing**
  - Dry oxidation @1200–1500°C
  - Dry oxidation @1200°C + CO<sub>2</sub> annealing @1050–1200°C
- **Al deposition**
- **Optical & Electrical measurements**

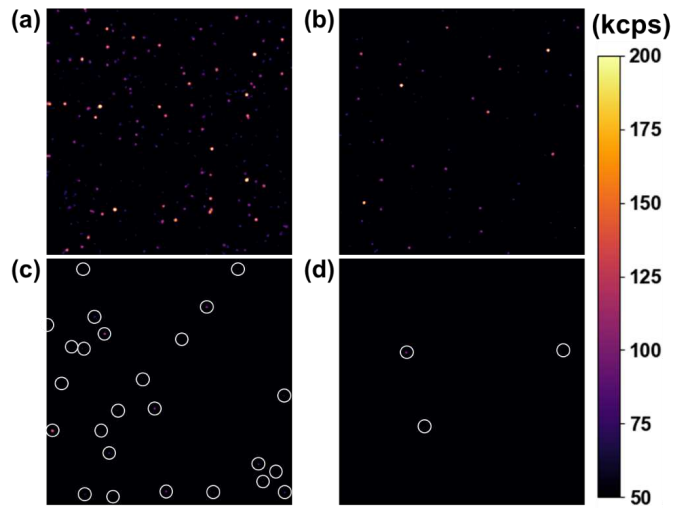
(b)



2 Fig. 1. (a) Flow of sample preparation and measurements and (b) schematic view of our in-house-  
3 built confocal microscope setup.

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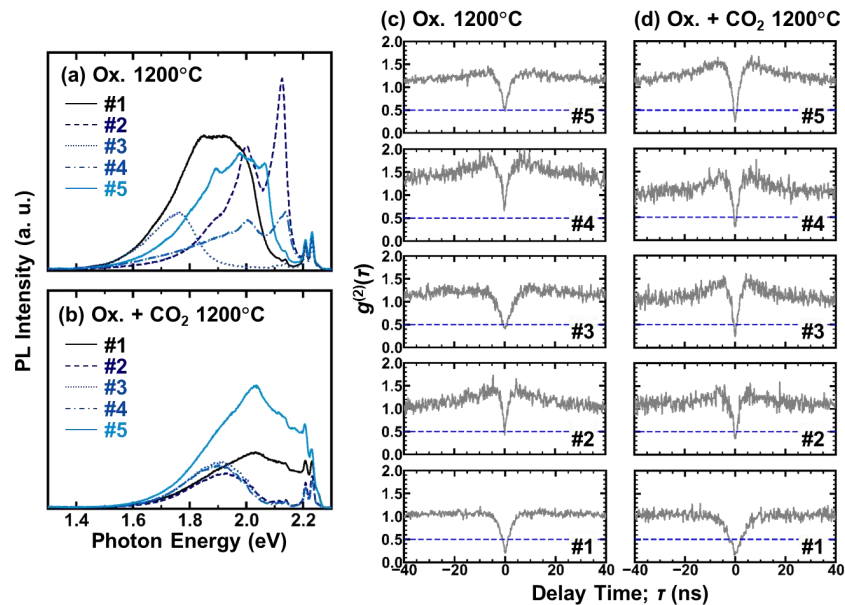
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1 Fig. 2. Confocal PL mapping images of SiO<sub>2</sub>/SiC samples (area: 30×30 μm<sup>2</sup>): (a) Ox. 1200°C, (b) Ox.  
 2 1500°C, (c) Ox.+CO<sub>2</sub> 1050°C, and (d) Ox.+CO<sub>2</sub> 1200°C. Excitation laser power is set to 0.5 mW. For  
 3 (c) and (d), the color centers with a count rate higher than 5.0×10<sup>4</sup> counts/sec are circled as guides.

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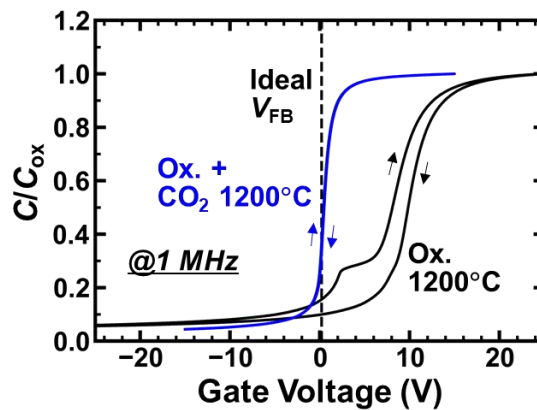
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1 Fig. 3. (a)(b) PL spectra for five different color centers observed in (a) Ox. 1200°C and (b) Ox.+CO<sub>2</sub>  
 2 1200°C. (c)(d) Results of HBT measurements without background subtraction for the defects in (c)  
 3 Ox. 1200°C and (d) Ox.+CO<sub>2</sub> 1200°C. Blue dashed lines correspond to the  $g^{(2)}(\tau)$  value of 0.5. The  
 4 sample number #1–#5 in (c) and (d) corresponds to the defect with same number in (a) and (b),  
 5 respectively.

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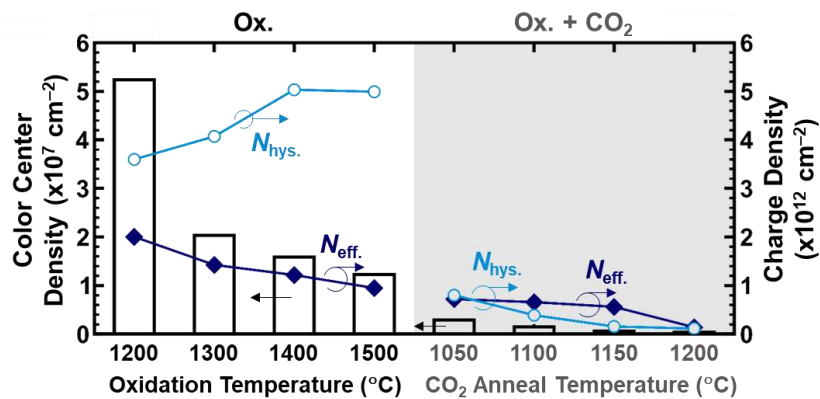
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1 Fig. 4. Bidirectional 1-MHz  $C-V$  characteristics of SiC MOS capacitors: Ox. 1200°C and Ox.+CO<sub>2</sub>  
 2 1200°C. Dashed line indicates ideal  $V_{FB}$  position.

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1 Fig. 5. Density of optically and electrically active defects shown as a function of sample conditions.  
 2 The density of color centers with a count rate higher than  $5.0 \times 10^4$  counts/sec is estimated through  
 3 image processing. The areal density of charges corresponding to the  $V_{\text{FB}}$  shift ( $N_{\text{eff.}}$ ) and  $C-V$   
 4 hysteresis ( $N_{\text{hys.}}$ ) are separately indicated.