

| Title        | Catalytic Asymmetric Construction of CF <sub>3</sub> -Substituted Chiral sp <sup>3</sup> Carbon Centers |
|--------------|---|
| Author(s)    | Hirano, Koji  |
| Citation     | Synthesis (Germany). 2022, 54(17), p. 3708-3718   |
| Version Type | АМ  |
| URL          | https://hdl.handle.net/11094/92743  |
| rights       | © 2022 Georg Thieme Verlag. All rights reserved.  |
| Note         |   |

# The University of Osaka Institutional Knowledge Archive : OUKA

https://ir.library.osaka-u.ac.jp/

The University of Osaka

# Catalytic Asymmetric Construction of CF<sub>3</sub>-Substituted Chiral sp<sup>3</sup> Carbon Centers

#### Koji Hirano\*

Department of Applied Chemistry, Graduate School of Engineering, Osaka University, Suita, Osaka 565-0871, Japan

k\_hirano@chem.eng.osaka-u.ac.jp

Click here to insert a dedication

Abstract Due to the unique steric and electronic nature of fluorine atom, organofluorine compounds have received significant attention in the fields of pharmaceuticals and agrochemicals. In particular, the CF3 group is frequently found in biologically active compounds. However, compared to aryl- and alkenyl-CF3-containing molecules, the construction of sp3 carbonbased alkyl-CF<sub>3</sub>-containing molecules, particularly via catalytic enantioselective synthesis, remains a considerable challenge in spite of their high potential in medicinal applications. This short review focuses on recent advances in this research area, and the reported strategies are categorized according to reaction types and starting substrates. In addition, chiral catalysts, substrate scope, and reaction mechanisms are briefly summarized.

- Introduction
- 2 Stereoselective Introduction of a CF<sub>3</sub> Group 2.1 Nucleophilic Addition to Carbonyls and Imines
- 2.2 Electrophilic Substitution at the  $\alpha$  Position of Carbonyls
- 2.3 Allylic Nucleophilic Substitution
- Stereoselective Functionalization of CF<sub>3</sub>-Substituted Molecules
- 3.1 Electrophilic Substitution of α-CF<sub>3</sub> Carbonyls
- 3.2 Substitution of α-Halo CF<sub>3</sub> Compounds
- 3.3 Addition-type Reactions with CF3-Substituted Alkenes
- Conclusion and Outlook

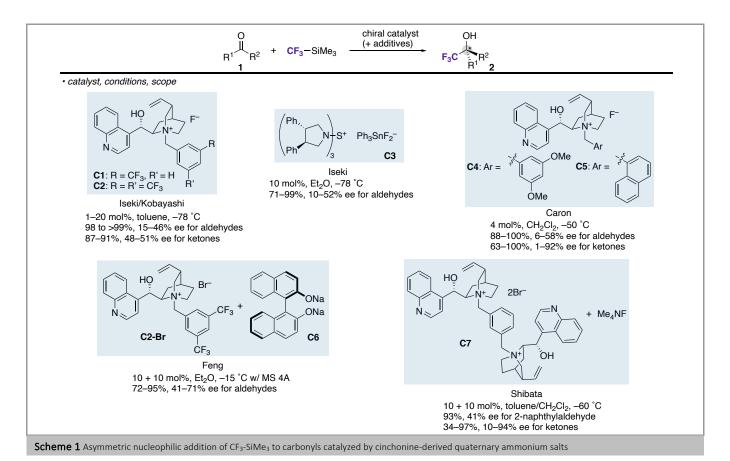
Key words addition, asymmetric catalysis, fluorine, substitution, trifluoromethyl group

# 1 Introduction

Owing to the unique steric and electronic characteristics of the fluorine atom, its introduction into parent organic molecules

often improves their biological activities. organofluorine compounds have received significant attention in the design of new drug candidates and agrochemicals.1 In particular, the trifluoromethyl group (CF3) is the most frequently occurring fluorinated organic group in biologically active compounds.2 Thus, the development of efficient synthetic protocols for the preparation of CF3-substitued molecules has been a long-standing research subject within the synthetic community. The synthesis of sp2 carbon-based arylalkenyl-CF3-containing molecules has witnessed remarkable progress as a result of the efforts of many synthetic chemists.<sup>3</sup> On the other hand, the construction of sp<sup>3</sup> carbon-CF<sub>3</sub> bonds, particularly via catalytic asymmetric process, remains underdeveloped, despite the fact that such motifs are found in several bioactive molecules, as exemplified by Alpelisib,4 Lansoprazole,5 Efavirenz,6 Befloxatone,7 and Odanacatib8 (Figure 1). In this short review, recent advances on the catalytic asymmetric construction of CF3-substituted chiral sp<sup>3</sup> carbon centers are described. The reported strategies are divided into two sections, namely, (1) stereoselective introduction of a CF<sub>3</sub> group, and (2) stereoselective functionalization of CF3-substituted molecules. In each section, the developed protocols are categorized according to the starting substrates employed. Additionally, chiral catalysts, substrate scope, and reaction mechanisms are briefly summarized.

Figure 1 Examples of CF<sub>3</sub>-substituted sp<sup>3</sup> carbon centers in bioactive molecules



### 2 Stereoselective Introduction of a CF<sub>3</sub> Group

# 2.1 Nucleophilic Addition to Carbonyls and Imines

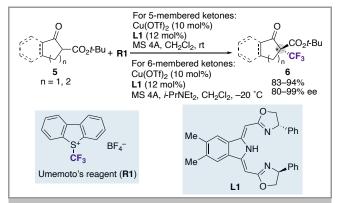
Since the reports of the preparation and synthetic application of CF<sub>3</sub>-SiMe<sub>3</sub> as a CF<sub>3</sub> anion by Rupport<sup>9a</sup> and Prakash,<sup>9b</sup> the fluoride-mediated nucleophilic addition reaction of CF3-SiMe3 to electrophilic carbonyls and imines has been widely studied for the synthesis of  $\alpha$ -CF3 alcohols and amines. However, the catalytic asymmetric equivalent is still somewhat challenging. A breakthrough was reported by Iseki and Kobayashi, where the cinchonine-derived N-benzylated quaternary ammonium fluorides C1 and C2 served as chiral catalysts to deliver the optically active  $\alpha$ -CF<sub>3</sub> alcohols 2 from aldehydes/ketones 1 The same research group subsequently developed the related chiral triaminosulfonium fluorostannate C3.11 The observed enantiomeric excess (ee) was generally low to moderate, but these seminal works clearly demonstrated the high potential of chiral ammonium and sulfonium fluoride catalysts in the asymmetric nucleophilic addition reaction with CF3-SiMe3. Caron extensively investigated the substituent effect on the nitrogen of the cinchoninium salt, leading to the development of the more active ammonium fluoride catalysts C4 and C5.12 reported the interesting synergy effect of chiral BINOL sodium salt C6 with the cinchonine-based catalyst C2-Br.13 In 2007, Shibata elegantly designed the linked cinchoninium bromide and accomplished the highly enantioselective trifluoromethylation of ketones by using a combination of C7 and Me<sub>4</sub>NF.<sup>14</sup>

Cinchonine-based quaternary ammonium catalysts were also applicable to the asymmetric trifluoromethylation of imines. Shibata reported the first successful example of the catalytic asymmetric addition of  $CF_3$ -SiMe $_3$  to the azomethine imines  $\bf 3$  in a toluene/ $CH_2Cl_2$  mixed solvent system with  $\bf C2$ - $\bf Br$  or its  $\it t$ -Bu-analogue  $\bf C8$  as the catalyst and KOH additive as an additive (Scheme 2, conditions A). The same reaction was possible in the non-toxic, environmentally benign fluorous solvent, Solkane®  $\bf 365mfc$  (1,1,1,3,3-pentafluorobutane), by using catalyst  $\bf C9$  possessing an  $\it ortho$ -iodo substituent (conditions B).  $\bf 15b$ 

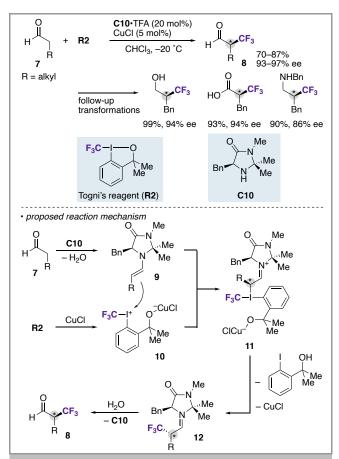
**Scheme 2** Asymmetric nucleophilic addition of CF<sub>3</sub>-SiMe<sub>3</sub> to imines catalyzed by cinchonine-derived quaternary ammonium salts

# 2.2 Electrophilic Substitution at the $\alpha$ Position of Carbonyls

The development of electrophilic trifluoromethylation agents of the type  $CF_{3}$ , such as  $Umemoto^{16}$  and  $Togni^{17}$  reagents, has allowed the enolizable carbonyl compounds to be adopted in trifluoromethylation reactions at the  $\alpha$  position. The catalytic asymmetric  $\alpha$ -trifluoromethylation of relatively acidic  $\beta$ -ketoesters  $\mathbf{5}$  with Umemoto's reagent  $(\mathbf{R1})$  was achieved in the presence of a  $Cu(OTf)_2$  catalyst and the N,N,N-pincer-type chiral ligand  $\mathbf{L1}$  to construct the  $CF_3$ -substituted quaternary carbon center (Scheme 3). Although the substrate was limited to a cyclic system, the five- and six-membered  $\alpha$ - $CF_3$  ketones  $\mathbf{6}$  were obtained in good yields and high enantioselectivities.



Scheme 3 Asymmetric electrophilic trifluoromethylation of  $\beta$ -ketoesters catalyzed by a Cu/N,N,N-pincer ligand complex



**Scheme 4** Asymmetric electrophilic trifluoromethylation of aldehydes catalyzed by a CuCl/chiral imidazolidinone dual system

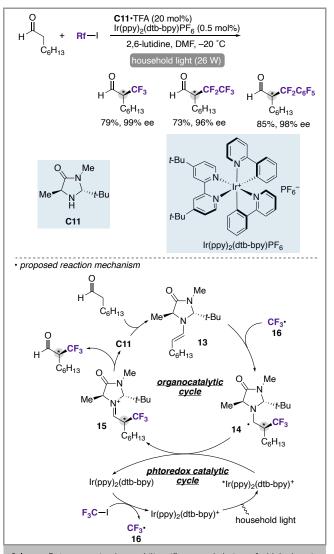
The readily available simple aliphatic aldehydes 7 also underwent asymmetric  $\alpha$ -trifluoromethylation with Togni's reagent (R2) under CuCl/MacMillan chiral imidazolidinone C10 dual catalyst system (Scheme 4).<sup>19</sup> The obtained  $\alpha$ -CF<sub>3</sub> chiral aldehyde 8 could be converted to the corresponding alcohol, carboxylic acid, and amine only with negligible to minor erosion of enantiomeric excess. The proposed reaction mechanism includes: (1) the formation of enamine intermediate 9 by dehydration of the aldehyde 7 with C10, (2) enantioselectivity-determining electrophilic attack of CuCl-activated, ring-opening hypervalent iodine reagent 10, (3) C-CF<sub>3</sub> forming reductive elimination from 11 to deliver 12, and finally (4) hydrolysis to give 8 along with regeneration of the starting imidazolidinone C10.

A similar enantioselective  $\alpha$ -trifluoromethylation of aldehyde with CF<sub>3</sub>-I was also possible by the combination of chiral imidazolidinone C11 and the photoredox catalyst Ir(ppy)<sub>2</sub>(dtb-bpy)PF<sub>6</sub> (Scheme 5).<sup>20</sup> A feature of this protocol is its successful application to more general fluoroalkylation with easily available and sometimes commercial fluoroalkyl iodides, which is a significant synthetic advantage compared to the aforementioned electrophilic trifluoromethylations with Umemoto and Togni reagents (see Schemes 3 and 4). The reaction is considered to proceed via the merger of a C11-based organocatalytic cycle and an Ir-promoted photoredox catalytic cycle. The photo-irradiated excited \*Ir(ppy)<sub>2</sub>(dtb-bpy)+ catalyst initially accepts single electron transfer from a sacrificial quantity of the in situ generated enamine intermediate 13 to form reduced Ir(ppy)<sub>2</sub>(dtb-bpy). Its

strongly reducing nature enables the one-electron reduction of CF<sub>3</sub>-I to generate the CF<sub>3</sub> radical **16** and Ir(ppy)<sub>2</sub>(dtb-bpy)<sup>+</sup> in the ground state. The electrophilic radical addition of **16** to the enamine **13** occurs in an enantioselective manner to give the stereodefined  $\alpha$ -amino alkyl radical **14**. Subsequent one-electron oxidation with the excited \*Ir(ppy)<sub>2</sub>(dtb-bpy)<sup>+</sup> species is followed by hydrolysis to afford the enantioenriched  $\alpha$ -CF<sub>3</sub> aldehyde along with regeneration of the imidazolidinone catalyst **C11**. The concurrently formed Ir(ppy)<sub>2</sub>(dtb-bpy) enters the second single-electron transfer with CF<sub>3</sub>-I to complete the photoredox catalytic cycle.

### 2.3 Allylic Nucleophilic Substitution

palladium-catalyzed allylic substitution proceeding via a  $\pi$ -allyl palladium species, the so-called Tsuji-Trost reaction, has been widely studied and applied to the synthesis of complex natural products and bioactive molecules.<sup>21</sup> Now, various allylic electrophiles and external nucleophiles can be successfully employed in this type of reaction. However, there was no successful use of a CF3 nucleophile until Trost reported the asymmetric allylic substitution of allylic fluorides 17 with CF3-SiMe3 under Pd/originally developed diamidophosphite bidentate ligand L2 catalysis in 2019 (Scheme 6).22 The key to success is the fluoride leaving group as well as the specific chiral ligand: other common leaving groups, such as carbonate, acetate, and chloride, did not provide the trifluoromethylated product 18 at all. Only six-membered cyclic substrates were applicable, but this protocol can construct a CF<sub>3</sub>-substituted chiral sp<sup>3</sup> carbon center without any proximal directing heteroatoms. The reaction is considered to proceed via a common outer-sphere mechanism, $^{21}$  involving an electrophilic  $\pi$ -allyl Pd intermediate and an externa CF3 anion nucleophile.



**Scheme 5** Asymmetric electrophilic trifluoromethylation of aldehydes via merger of organocatalytic and photoredox catalytic cycles

$$\begin{array}{c} \text{CPPd}(\eta^3\text{-}\text{C}_3\text{H}_5) \text{ (6 mol\%)} \\ \text{17} \\ \text{Y = CH}_2, \text{ NBoc} \\ \text{R = H, Ar, alkynyl, alkenyl} \\ \end{array} \begin{array}{c} \text{L2 (7 mol\%)} \\ \text{MTBE, 50 °C} \\ \text{43-95\%} \\ \text{60-94\% ee} \\ \end{array}$$

 $\begin{tabular}{ll} Scheme 6 & Pd-catalyzed asymmetric allylic trifluoromethylation of allylic fluorides with CF_3-SiMe_3 \\ \end{tabular}$ 

While somewhat specific, Shibata<sup>23</sup> and Jiang<sup>24</sup> independently reported the enantioselective allylic trifluoromethylation of Morita-Baylis-Hillman (MBH)-type allyl electrophiles **19** with CF<sub>3</sub>-SiMe<sub>3</sub> by using (DHQD)<sub>2</sub>PHAL as the nucleophilic chiral organocatalyst (Scheme 7). The racemic MBH adduct **19** initially undergoes an S<sub>N</sub>2' reaction with (DHQD)<sub>2</sub>PHAL and is converted to the activated allylic intermediate **20**. A subsequent face-selective second S<sub>N</sub>2' reaction with CF<sub>3</sub>-SiMe<sub>3</sub> furnishes the net S<sub>N</sub>2-type product **21** in enantioenriched form.

$$\begin{array}{c} \text{OBoc} \\ \text{RO}_2\text{C} \\ \text{Ar} + \text{CF}_3 - \text{SiMe}_3 \\ \text{19} \\ \text{R} = \text{Me, } \text{t-Bu} \\ \end{array} \begin{array}{c} \text{Shibata's conditions:} \\ \text{(DHQD)}_2\text{PHAL (10 mol\%)} \\ \text{THF, 60 °C} \\ \text{or} \\ \text{Jiang's conditions:} \\ \text{(DHQD)}_2\text{PHAL (15 mol\%)} \\ \text{KF, DME/MeCN, rt} \\ \end{array} \begin{array}{c} \text{RO}_2\text{C} \\ \text{Ar} \\ \text{37-86\%} \\ \text{48-94\% ee} \\ \end{array}$$

Scheme 7 Organocatalytic asymmetric allylic trifluoromethylation of MBH adducts

# 3 Stereoselective Functionalization of CF<sub>3</sub>-Substituted Molecules

Since CF<sub>3</sub>-carbonyls, -imines, and their derivatives are readily prepared, enantioselective reductions and addition-type reactions toward optically active  $\alpha$ -CF<sub>3</sub> alcohols and amines have been actively studied by many synthetic chemists. In addition, comprehensive review articles covering these studies have already appeared,<sup>25</sup> hence this short review thus does not discuss the stereoselective functionalization of CF<sub>3</sub>-carbonyls and imines.

# 3.1 Electrophilic Substitution of α-CF<sub>3</sub> Carbonyls

The preparation of  $\alpha$ -CF $_3$  carbonyl compounds is also relatively easy, but their application in the metal-catalyzed asymmetric electrophilic substitution at the  $\alpha$  position is generally difficult because of the propensity for metal fluoride (MF) elimination from the metal enolate intermediate via an intramolecular metal-fluorine interaction (Scheme 8).

Scheme 8  $\alpha$ -CF3 metal enolate: metal fluoride elimination vs electrophilic substitution

In this context, Kumagai and Shibasaki elegantly designed the  $\alpha$ -CF $_3$  amide **22** that bears a 7-azaindoline directing group and successfully performed a Cu-catalyzed asymmetric Mannichtype reaction with high diastereo- and enantioselectivity (Scheme 9).<sup>27</sup> The key to success was coordination of the 7-azaindoline nitrogen to the Cu center, to suppress the possible

metal fluoride elimination, which was supported by X-ray crystallographic analysis of the Cu/22 complex. Subsequently, the same research group extended this strategy to asymmetric  $\alpha$ -allylation with the allyl carbonate 25 by utilizing a Cu/Pd dual catalyst system (Scheme 10).<sup>28</sup> Both the  $\alpha$ -CF<sub>3</sub> chiral Cu enolate nucleophile and achiral  $\pi$ -allyl Pd electrophile are generated in a catalytic manner and successfully underwent cross-coupling to form the enantioenriched product 26.

Scheme 9 The Cu-catalyzed, 7-azaindoline-directed asymmetric Mannichtype reaction of an  $\alpha$ -CF<sub>3</sub> amide with imines

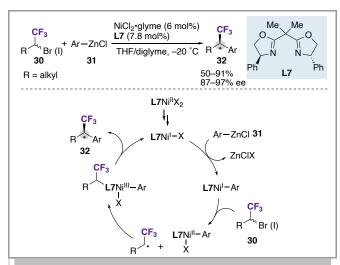
The aforementioned work in Schemes 9 and 10 effectively promoted electrophilic substitution at the  $\alpha$  position of the  $\alpha$ -CF<sub>3</sub> carbonyl over the undesired fluoride elimination, but the tedious attachment and detachment of the 7-azaindoine auxiliary is unavoidable. In 2021, Wang and Xue reported the Pd-catalyzed asymmetric  $\alpha$ -allylation of simple  $\alpha$ -CF3 cyclic ketones 27 with MBH adducts 28 without any directing groups (Scheme 11).29 Moreover, the ligand-controlled regiodivergent allylation was also achieved: the phox-type ligand L5 selectively afforded the less hindered liner-type product 29, whereas the branched and more congested regioisomer 29' was formed predominantly in the presence of phosphoramidite ligand L6. The observed divergent regioselectivity can be explained by the ligand-dependent outer- and inner-sphere mechanism switch.30 Namely, the allyl fragment of the L5ligated  $\pi$ -allyl Pd species undergoes the relatively common outer-sphere substitution reaction<sup>21</sup> with the in-situ generated enolate from 27 at the more sterically accessible terminal position to deliver the linear-type product 29. On the other hand, in the case of the **L6**-coordinated  $\pi$ -allyl Pd intermediate, ligand exchange with the parent 27 on Pd is followed by a [3,3'] sigmatropic reductive elimination31 to afford the branched isomer 29'.

# 3.2 Substitution of α-Halo CF<sub>3</sub> Compounds

The metal-catalyzed enantioconvergent C–C cross-coupling reaction of racemic  $\alpha$ -halo CF $_3$  compounds with organometallic reagents is a powerful method to generate a chiral CF $_3$ -substituted sp $_3$  carbon center without the need of any proximal heteroatoms. Fu reported the seminal work on the Ni/box-

Scheme 10 The Cu/Pd-co-catalyzed, 7-azaindoline-directed asymmetric  $\alpha$ -allylation of an  $\alpha$ -CF<sub>3</sub> amide with allyl carbonates

**Scheme 11** Pd-catalyzed, ligand-controlled regiodivergent asymmetric  $\alpha$ -allylation of  $\alpha$ -CF<sub>3</sub> ketones

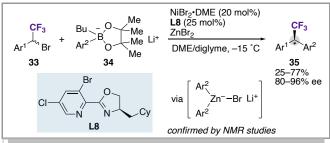


Scheme 12 Ni-catalyzed enantioconvergent cross-coupling of  $\alpha$ -halo CF $_3$  compounds with arylzinc reagents

type ligand L7-catalyzed enantioconvergent cross-coupling of racemic  $\alpha$ -Br and -I CF<sub>3</sub> compounds **30** with arylzinc reagents **31** (Scheme 12).<sup>32</sup> A radical-involved, organometallic Ni(I)/(III) redox mechanism was originally proposed, which was later supported experimentally and theoretically by the authors themselves and other research groups.<sup>33</sup> The ArNi(I) species generated by transmetalation with **31** undergoes single-electron transfer (SET) to **30** to form ArNi(II) and an alkyl

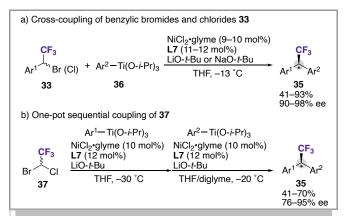
radical intermediate, where the stereochemical information of starting  ${\bf 30}$  is lost. Subsequent radical rebound to ArNi(II) is followed by reductive elimination from ArNi(III) to deliver the target product  ${\bf 32}$  with regeneration of Ni(I)X to complete the catalytic cycle. The enantioselectivity can be determined in the radical rebound or reductive elimination step.  $^{34}$ 

Shen subsequently developed the related Ni/pyridine-oxazoline ligand L8-catalyzed enantioconvergent cross-coupling of racemic  $CF_3$  benzyl bromides 33 with arylborates 34, generated from the parent arylboronates and BuLi (Scheme 13). Notably,  $ZnBr_2$  was a crucial additive, which dramatically accelerated the transmetalation process via formation of the corresponding arylzincate species. The successful use of benzylic halides 33 is complementary to the reaction with the alkyl halides 30 described in Scheme 12.



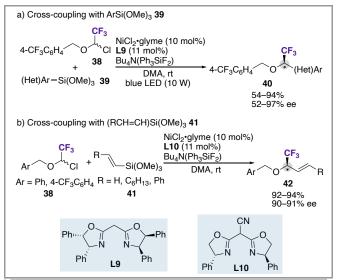
Scheme 13 Ni-catalyzed enantioconvergent cross-coupling of  $\alpha$ -halo CF3 compounds with arylboron reagents

Aryltitanates **36** could also be employed in a similar Nicatalyzed enantioconvergent cross-coupling reaction of compounds **33** (Scheme 14a).<sup>36</sup> A feature of the titanate-based process is the applicability of less reactive but stable and cheap benzylic chlorides as well as bromides. The successful conversion of the benzylic chloride allowed 1-bromo-1-chloro-2,2,2-trifluoroethane (**37**), a readily available anesthetic, to be adopted in one-pot sequential cross-coupling reaction with two different aryltitanates, directly giving products **35** in good yields (Scheme 14b).



Scheme 14 Ni-catalyzed enantioconvergent cross-coupling of  $\alpha$ -halo CF3 compounds with aryltitanate reagents

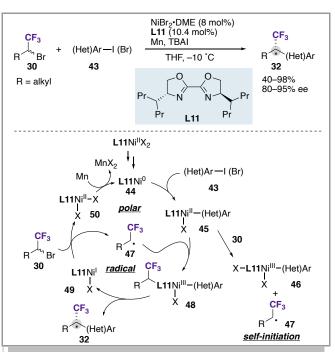
While viable substrates were limited to the somewhat specialized ether derivatives **38**, Gandelman reported the Hiyama-type cross-coupling of (Het)ArSi(OMe)<sub>3</sub> **39** with the assistance of blue LED irradiation (Scheme 15a).<sup>37</sup> The reaction itself proceeded with the same enantioselectivity even under dark conditions, but the reaction time was prolonged. The synthetic advantage of the Si-based Hiyama reaction is the accommodation of not only (Het)ArSi(OMe)<sub>3</sub> **39** but also (RCH=CH)Si(OMe)<sub>3</sub> **41** (Scheme 15b).



**Scheme 15** Ni-catalyzed enantioconvergent cross-coupling of  $\alpha$ -halo CF<sub>3</sub> ethers with organosilicon reagents

These cross-coupling reactions are quite useful for the convergent synthesis of chiral CF<sub>3</sub>-containing molecules, but one

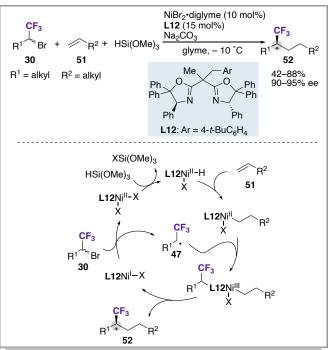
disadvantage is the prior preparation of organometallic reagents, which is sometimes difficult and problematic. Recently, Wang developed the Ni-catalyzed enantioselective reductive cross-coupling reaction (Scheme 16).38 Without any preactivation,  $\alpha\text{-halo}$   $\text{CF}_3$  compounds  $\boldsymbol{30}$  and (hetero)aryl iodides/bromides 43 were directly cross-coupled in the presence of metallic manganese (Mn) as the terminal reductant, delivering the enantioenriched products 32. advantages of the ready availability of various aryl halides, complicated fluorinated molecules can also be easily accessed. The authors did not mention a detailed mechanism, but the Ni(0)/(I)/(II)/(III)-involved, polar-radical-combined mechanism is believed to operate, which was originally developed by Weix in the related reductive cross-coupling of aryl halides and nonfluorinated alkyl halides.<sup>39</sup> The starting point is the in situ generated Ni(0) species 44, which undergoes a two-electron redox oxidative addition with 43. The formed ArNi(II) species 45 initially reacts with 30 to generate ArNi(III) 46 and alkyl radical 47; this step is a so-called self-initiation process. The radical 47 then combined with ArNi(II) 45 to afford ArNi(III) species 48. Reductive elimination furnishes 32 and Ni(I) 49, and the latter then enters into a single-electron reduction with 30, giving the radical 47 and Ni(II) 50. The catalytic cycle is closed by the two-electron reduction of 50 with Mn.



Scheme 16 Ni-catalyzed enantioconvergent reductive cross-coupling of  $\alpha$ -halo CF $_3$  compounds and aryl halides

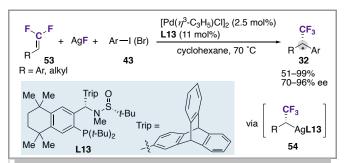
Another approach to avoid the stoichiometric preparation of organometallic reagents is via the NiH-catalyzed asymmetric hydroalkylation of terminal alkene **51** with compounds **30** (Scheme 17).<sup>40</sup> A catalytically generated alkylnickel species from alkene **51** and HSi(OMe)<sub>3</sub> undergoes an enantioselective cross-coupling reaction with **30** to form products **52** with high ee values. The proposed reaction mechanism involves SET-promoted alkyl radical intermediates and Ni(I)/(II)/(III) organometallic species, which is similar to the aforementioned pathways in Schemes 12 and 16. It should be noted that the

asymmetric Ni catalysis can control the point chirality at the CF<sub>3</sub>-substituted sp<sup>3</sup> carbon center with two different simple alkyl substituents, which is difficult by other means.



Scheme 17 Ni-catalyzed enantioselective hydroalkylation of terminal alkenes with  $\alpha$ -halo CF $_3$  compounds and a hydrosilane

While somewhat differently categorized, the related umpolung-type strategy using the  $\alpha$ -CF<sub>3</sub> alkylsilver **54**, generated in situ from *gem*-difluoroalkenes **53** and AgF, was reported by Li and Zhang (Scheme 18).<sup>41</sup> In the presence of [Pd( $\eta^3$ -C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub> cas the atalyst and the originally developed chiral sulfinamide phosphine ligand **L13**, the aryl halides **43** were converted into the desired products **32** with high enantioselectivities.

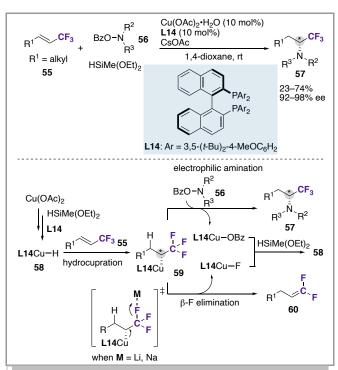


**Scheme 18** Pd-catalyzed asymmetric three-component-coupling of *gem*-difluoroalkenes, AgF, and aryl halides

# 3.3 Addition-type Reactions with CF<sub>3</sub>-Substituted Alkenes

CF<sub>3</sub>-substitted alkenes are potentially useful starting platforms for the preparation of chiral CF<sub>3</sub>-containing molecules, because the stereoselective functionalization of alkene moiety enables the rapid construction of CF<sub>3</sub>-substituted chiral sp<sup>3</sup> carbon centers. However, almost all the reported procedures relied on the presence of additional electron-withdrawing groups such as carbonyl and aromatic rings, which increase the reactivity and

feasibility of the stereocontrolled process.<sup>42</sup> The first successful example using simple CF3-substituted aliphatic alkenes 55 was reported by Hirano and Miura (Scheme 19), where the CsOAcassisted, CuH/L14-catalyzed asymmetric hydroamination with hydroxylamines 56 and HSiMe(OEt)<sub>2</sub> was in operation.<sup>43</sup> The optically active  $\alpha\text{-CF}_3$  amines 57 were obtained with high enantiomeric excess values. The reaction proceeds via (1) regio- and enantioselective hydrocupration of 55 with the L14CuH species 58 obtained from Cu(OAc)2 and HSiMe(OEt)2, (2) electrophilic amination of the stereodefined  $\alpha$ -CF<sub>3</sub> alkylcopper 59 with 56 in a stereoretentive manner to give enantioenriched 57, and (3) regeneration of L14CuH 58 by  $\sigma$ bond metathesis of concurrently formed L14CuOBz with HSiMe(OEt)<sub>2</sub>. The conceivable byproduct is the gemdifluoroalkene 60, obtained by  $\beta$ -fluoride elimination from 59, which is similar to the metal fluoride elimination from the  $\alpha\text{-CF}_3$ metal enolate mentioned in Scheme 8. Actually, when fluorophilic alkaline metal bases such as Li and Na were employed instead of CsOAc, the undesired  $\beta$ -fluoride elimination predominantly occurred via intermolecular metal (M) to F interaction.<sup>44</sup> Thus, the lower fluorophilicity of the Cs base  $^{45}$  is the key to success for functionalization at the  $\alpha$ position relative to CF<sub>3</sub>.



**Scheme 19** Cu-catalyzed, CsOAc-assisted asymmetric hydroamination of CF<sub>3</sub>-substituted aliphatic alkenes with hydroxylamines and a hydrosilane

The same authors subsequently developed a related Cucatalyzed asymmetric hydroallylation with allylic chlorides  $\bf 61$  instead of the hydroxylamine  $\bf 56$  (Scheme 20).  $^{46}$  In this case, the use of 18-crown-6 was crucial for the disruption of Cs-F interaction  $^{47}$  to suppress the undesired  $\beta$ -F elimination. This is one of the few successful examples of catalytic asymmetric construction of non-allylic and non-benzylic CF3-substituted sp³ chiral carbon centers.

In addition to the CuH species, CuBpin also undergoes the enantioselective addition reaction with simple CF<sub>3</sub>-substituted

aliphatic alkenes. Hoveyda reported the Cu-catalyzed enantioselective deuterioboration with pinB–Bpin and  $D_2O$  in the presence of the originally developed chiral NHC ligand **L16** (Scheme 21),<sup>48</sup> albeit with only one successful example.

$$R^{1} = \text{alkyl} \\ 55 \\ R^{2} = \text{H, Ar, alkyl} \\ \frac{\text{Me}}{\text{Si} = \text{O}_{n}} \\ \text{PMHS} \\ R^{2} = \text{H, Ar, alkyl} \\ \frac{\text{Me}}{\text{Si} = \text{O}_{n}} \\ \text{He} \\ \text{PMHS} \\ \text{Via} \\ \begin{bmatrix} \text{Cu(OAc)}_{2} (10 \text{ mol%}) \\ \text{L15 (10 mol%}) \\ \text{CsOPiv/18-crown-6} \\ \text{THF, rt} \\ \\ \frac{\text{62 } \text{R}^{2}}{\text{48-93\%}} \\ \frac{48-93\%}{94-99\%} \text{ ee} \\ \\ \text{Via} \\ \begin{bmatrix} \text{He} \\ \text{R}^{1} \\ \text{CF} \\ \text{F} \\ \text{OOO} \\ \text{O$$

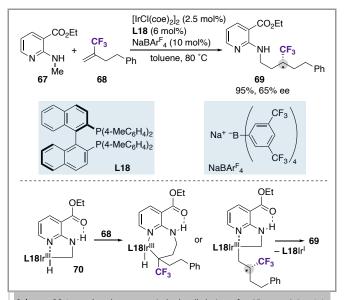
Scheme 20 Cu-catalyzed, 18-crown-6-Aasisted asymmetric hydroallylation of CF<sub>3</sub>-substituted aliphatic alkenes with allyl chlorides and a hydrosilane

 $\label{eq:continuous} \textbf{Scheme 21} \ \, \text{Cu-catalyzed asymmetric deuterioboration of CF}_3\text{-substituted} \\ \text{aliphatic alkenes with a diboron and } D_2O \\$ 

Recently, Huang and Zhang reported the Cu/L17-catalyzed enantioselective protoboration of CF3-substituted 1,3-dienes with pinB-Bpin and EtOH.49 Electronically activated, arylsubstituted dienes were mainly employed, but successful application to the aliphatic substrate 63 was also shown (Scheme 22). After oxidative follow-up treatment of initially formed 64 with NaBO3•4H2O, the corresponding alcohol 65 bearing a chiral CF3-substituted sp3 carbon center was obtained with good enantioselectivity. The proposed reaction mechanism is as follows: the in-situ generated L17CuBpin species adds to 63 regioselectively at the more sterically accessible terminal position to generate the allyl Cu intermediate 66. The observed protoborated product is produced by subsequent enantioselective S<sub>N</sub>2'-type protonation with EtOH. Also in this catalytic process, the fluoride elimination reaction potentially occurs, but DFT calculations suggest that it is an energetically higher and kinetically less favored pathway.

Scheme 22 Cu-catalyzed asymmetric protoboration of a CF<sub>3</sub>-substituted aliphatic 1,3-diene with with a diboron and EtOH

Another successful application of unactivated CF3-substituted aliphatic alkenes is in the Ir-catalyzed asymmetric C-H addition reaction of 2-(methylamino)pyridine derivative 67 (Scheme 23), as reported by Nishimura.50 Under Ir/L18 asymmetric catalysis, 1,1-disubstituted CF3 substrate 68 was converted to the enantioenriched amine 69 possessing a chiral CF3-substituted sp³ carbon center. A plausible reaction mechanism includes the pyridine-directed, oxidative addition of the N-Me C-H to Ir(I), in which the ester moiety at the C3 position of the pyridine ring plays an important role in the regulation of the substrate configuration (70). The CF3 alkene 68 then undergoes insertion into C-Ir or Ir-H bond of 70, and final reductive elimination affords the asymmetric hydroalkylation product 69. Although the enantioselectivity was still moderate, this protocol receives more attention because metalated and halogenated starting substrates are not needed.



**Scheme 23** Ir-catalyzed asymmetric hydroalkylation of a CF<sub>3</sub>-containing 1,1-disubstituted aliphatic alkene with a 2-(aminomethyl)pyridine derivative

# 4 Conclusion and Outlook

Recent development on chiral metal catalysts, organocatalysts, and trifluoromethylation reagents, as well as the rational design of reactions and substrates, has enabled significant progress on the catalytic asymmetric construction of  $CF_3$ -substituted chiral  $SF_3$  carbon centers. Given the prevalence of  $SF_3$  groups in recent

marketed drugs, the aforementioned advance are expected to find wide application in more concise and stereoselective synthesis of chiral CF3 drug molecules. However, almost all the reported protocols still require prefunctionalization, such as halogenation, metalation, introduction of reactive functional groups, or preinstallation of CF<sub>3</sub>, of the starting substrates. On the other hand, the direct sp3 C-H trifluoromethylation has recently been described by several research groups.<sup>51</sup> Although the reaction conditions and site-selectivity still need to be improved, the direct substitution of simple and unactivated aliphatic C-H bonds with a CF3 group is possible without any preactivation steps, and thus requires additional attention from a synthetic point of view. The further development of mild reaction conditions and highly site- and enantioselective catalysts is expected to enable the asymmetric sp<sup>3</sup> C-H trifluoromethylation of simple substrates and late-stage asymmetric introduction of CF<sub>3</sub>, which can open the door to the discovery of new drug candidates based on CF3-substituted chiral sp3 carbon centers.

### **Funding Information**

This work was supported by the Japan Society for the Promotion of Science (JSPS), KAKENHI (JP 22H02077) [Grant-in-Aid for Scientific Research(B)] and the Japan Science and Technology Agency (JST), FOREST Program (JPMJFR 211X).

# **Conflict of Interest**

The author declares no conflict of interest.

#### References

- (1) (a) Swallow, S. In Fluorine in Pharmaceutical and Medicinal Chemistry: From Biophysical Aspects to Clinical Applications; Gouverneur, V. Müller, K., Eds.; Imperial College Press: London, 2012; p 141. (b) Müller, K.; Faeh, C.; Diederich, F. Science 2007, 317, 1881. (c) Gillis, E. P.; Eastman, K. J.; Hill, M. D.; Donnelly, D. J.; Meanwell, N. A. J. Med. Chem. 2015, 58, 8315. (d) Ogawa, Y.; Tokunaga, E.; Kobayashi, O.; Hirai, K.; Shibata, N. iScience 2020, 23, 101467.
- (2) (a) Wishart, D. S.; Feunang, Y. D.; Guo, A. C.; Lo, E. J.; Marcu, A.; Grant, J. R.; Sajed, T.; Johnson, D.; Li, C.; Sayeeda, Z.; Assempour, N.; Iynkkaran, I.; Liu, Y.; Maciejewski, A.; Gale, N.; Wilson, A.; Chin, L.; Cummings, R.; Le, D.; Pon, A.; Knox, C.; Wilson, M. Nucleic Acids Res. 2018, 46, D1074. (b) Meanwell, N. A. J. Med. Chem. 2018, 61, 5822.
- (3) (a) Yang, X.; Wu, T.; Phipps, R. J.; Toste, F. D. Chem. Rev. 2015, 115, 826. (b) Alonso, C.; de Marigorta, E. M.; Rubiales, G.; Palacios, F. Chem. Rev. 2015, 115, 1847.
- (4) André, F.; Ciruelos, E.; Rubovszky, G.; Campone, M.; Loibl, S.; Rugo, H. S.; Iwata, H.; Conte, P.; Mayer, I. A.; Kaufman, B.; Yamashita, T.; Lu, Y.-S.; Inoue, K.; Takahashi, M.; Pápai, Z.; Longin, A.-S.; Mills, D.; Wilke, C.; Hirawat, S.; Juric, D. N. Engl. J. Med. 2019, 380, 1929.
- (5) Takagi, T.; Naito, Y.; Okada, H.; Ishii, T.; Mizushima, K.; Akagiri, S.; Adachi, S.; Hanada, O.; Kokura, S.; Ichikawa, H.; Itoh, K.; Yamamoto, M.; Matsui, H.; Yoshikawa, T. J. Pharmacol. Exp. Ther. 2009, 331, 255.
- (6) Young, S. D.; Britcher, S. F.; Tran, L. O.; Payne, L. S.; Lumma, W. C.; Lyle, T. A.; Huff, J. R.; Anderson, P. S.; Olsen, D. B.; Carroll, S. S.; Pettibone, D. J.; O'Brien, J. A.; Ball, R. G.; Balani, S. K.; Lin, J. H.; Chen, I.-W.; Schleif, W. A.; Sardana, V. V.; Long, W. J.; Byrnes, V. W.; Emini, E. A. Antimicrob. Agents Chemother. 1995, 39, 2602.
- (7) Curet. O.; Damoiswau, G.; Aubin, N.; Sontag, N.; Jarreau, F. X. J. Pharmacol. Exp. Ther. 1996, 277, 253.
- (8) Masarachia, P. J.; Pennypacher, B. L.; Pickarski, M.; Scott, K. R.; Wesolowski, G. A.; Smith, S. Y.; Samadfam, R.; Goetzmann, J. E.;

- Scott, B. B.; Kimmel, D. B.; Duong, L. T. *J. Bone Miner. Res.* **2012**, *27*, 509.
- (9) (a) Rupport, I.; Schlich, K.; Volbach, W. Tetrahedron Lett. 1984, 25, 2195. (b) Parakash, G. K. S.; Krishnamurti, R.; Olah, G. A. J. Am. Chem. Soc. 1989, 111, 393.
- (10) Iseki, K.; Nagai, T.; Kobayashi, Y. Tetrahedron Lett. 1994, 35, 3137.
- (11) Kuroki, Y.; Iseki, K. Tetrahedron Lett. 1999, 40, 8231.
- (12) Caron, S.; Do, N. M.; Aroin, P.; Larivée, A. Synthesis 2003, 1693.
- (13) Zhao, H.; Qin, B.; Liu, X.; Feng, X. Tetrahedron 2007, 63, 6822.
- (14) Mizuta, S.; Shibata, N.; Akiti, S.; Fujimoto, H.; Nakamura, S.; Toru, T. *Org. Lett.* **2007**, *9*, 3703.
- (15) (a) Kawai, H.; Kusuda, A.; Nakamura, S.; Shiro, M.; Shibata, N. Angew. Chem. Int. Ed. 2009, 48, 6324. (b) Okusu, S.; Kawai, H.; Xu, X.-H.; Tokunaga, E.; Shibata, N. J. Fluorine Chem. 2012, 143, 216.
- (16) Umemoto, T.; Ishihara, S. J. Am. Chem. Soc. 1993, 115, 2156.
- (17) Kieltsch, I.; Eisenberger, P.; Togni, A. Angew. Chem. Int. Ed. 2007, 46, 754.
- (18) Deng, Q.-H.; Wadepohl, H.; Gade, L. H. J. Am. Chem. Soc. 2012, 134, 10769.
- (19) Allen, A. E.; MacMillan, D. W. C. J. Am. Chem. Soc. 2010, 132, 4986.
- (20) Nagib, D. A.; Scott, M. E.; MacMillan, D. W. C. J. Am. Chem. Soc. 2009, 131, 10875.
- (21) (a) Tsuji, J.; Minami, I. Acc. Chem. Res. 1987, 20, 140. (b) Trost, B. M.; Crawley, M. L. Chem. Rev. 2003, 103, 2921. (c) Trost, B. M.; Machacek, M. R.; Aponick, A. Acc. Chem. Res. 2006, 39, 747. (d) Trost, B. M. Tetrahedron 2015, 71, 5708. (e) Tsuji, J. Tetrahedron 2015, 71, 6330.
- (22) Trost, B. M.; Gholami, H.; Zell, D. J. Am. Chem. Soc. 2019, 141, 11446.
- (23) Furukawa, T.; Nishimine, T.; Tokunaga, E.; Hasegawa, K.; Shiro, M.; Shibata, N. Org. Lett. 2011, 13, 3972.
- (24) Li, Y.; Liang, F.; Xu, Y.-c.; Wang, Q.-R.; Jiang, L. Org. Lett. 2011, 13, 6082.
- (25) (a) Nie, J.; Guo, H.-C.; Cahard, D.; Ma, J.-A. Chem. Rev. 2011, 111, 455. (b) Onyeagusi, C. I.; Malcolmson, S. J. ACS Catal. 2020, 10, 12507.
- (26) Itoh, Y.; Yamanaka, M.; Mikami, K. J. Am. Chem. Soc. 2004, 126, 13174.
- (27) Yin, L.; Brewitz, L.; Kumagai, N.; Shibasaki, M. J. Am. Chem. Soc. 2014, 136, 17958.
- (28) Saito, A.; Kumagai, N.; Shibasaki, M. Angew. Chem. Int. Ed. 2017, 56, 5551
- (29) Zhu, Y.; Ni, Y.; Lu, C.; Wang, X.; Wang, Y.; Xue, X.-S.; Pan, Y. Org. Lett. 2021, 23, 2443.
- (30) Bai, D.-C.; Yu, F.-L.; Wang, W.-Y.; Chen, D.; Li, H.; Liu, Q.-R.; Ding, C.-H.; Chen, B.; Hou, X.-L. Nat. Commun. 2016, 7, 11806.
- (31) (a) Méndez, M.; Cuerva, J. M.; Gómez-Bengoa, E.; Cárdenas, D. J.; Echavarren, A. M. Chem. Eur. J. 2002, 8, 3620. (b) Zhang, P.; Brozek, L. A.; Morken, J. P. J. Am. Chem. Soc. 2010, 132, 10686.
- (32) Liang, Y.; Fu, G. C. J. Am. Chem. Soc. 2015, 137, 9523.
- (33) (a) Jones, G. D.; Martin, J. L.; McFarland, C.; Allen, O. R.; Hall, R. E.; Haley, A. D.; Brandon, R. J.; Konovalova, T.; Desrochers, P. J.; Pulay, P.; Vicic, D. A. J. Am. Chem. Soc. 2006, 128, 13175. (b) Lin, X.; Phillips, D. L. J. Org. Chem. 2008, 73, 3680. (c) Schley, N. D.; Fu, G. C. J. Am. Chem. Soc. 2014, 136, 16588.
- (34) Gutierrez, O.; Tellis, J. C.; Primer, D. N.; Molander, G. A.; Kozlowski, M. C. J. Am. Chem. Soc. 2015, 137, 4896.
- (35) Huang, W.; Hu, M.; Wan, X.; Shen, Q. Nat. Commun. 2019, 10, 2963.
- (36) Varenikov, A.; Shapiro, E.; Gandelman, M. Org. Lett. 2020, 22, 9386.
- (37) Gandelman, M.; Varenikov, A. Nat. Commun. 2018, 9, 3566.
- (38) Min, Y.; Sheng, J.; Yu, J.-L.; Ni, S.-X.; Ma, G.; Gong, H.; Wang, X.-S. Angew. Chem. Int. Ed. 2021, 60, 9947.
- (39) Biswas, S.; Weix, D. J. J. Am. Chem. Soc. 2013, 135, 16192.
- (40) Guo, W.; Cheng, L.; Ma, G.; Tong, W. Wu, F. Org. Lett. 2022, 24, 1796.
- (41) Lin, T.-Y.; Pan, Z.; Tu, Y.; Zhu, S.; Wu, H.-H.; Liu, Y.; Li, Z.; Zhang, J. Angew. Chem. Int. Ed. 2020, 59, 22957.

- (42) For examples, see ref. 25 and: (a) Dong, K.; Li, Y.; Wang, Z.; Ding, K. Angew. Chem. Int. Ed. 2013, 52, 14191. (b) Massolo, E.; Benaglia, M.; Orlandi, M.; Rossi, S.; Celentano, G. Chem. Eur. J. 2015, 21, 3589. (c) Zhu, Y.; Li, X.; Chen, Q.; Su, J.; Jia F.; Qiu, S.; Ma, M.; Sun, Q.; Yan, W.; Wang, K.; Wang, R. Org. Lett. 2015, 17, 3826. (d) Zhu, Y.; Dong, Z.; Cheng, X.; Zhong, X.; Liu, X.; Shen, Z.; Yang, P.; Li, Y.; Wang, H.; Yan, W.; Wang, K.; Wang, R. Org. Lett. 2016, 18, 3546.
  (e) Holmes, M.; Nguyen, K. D.; Schwartz, L. A.; Luong, T.; Krische, M. J. J. Am. Chem. Soc. 2017, 139, 8114. (f) Pan, X.; Wang, Z.; Kan, L.; Mao, Y.; Zhu, Y.; Liu, L. Chem. Sci. 2020, 11, 2414.
- (43) Takata, T.; Hirano, K.; Miura, M. Org. Lett. 2019, 21, 4284.
- (44) (a) Corberán, R.; Mszar, N. W.; Hoveyda, A. H. Angew. Chem. Int. Ed. 2011, 50, 7079. (b) Kojima, R.; Akiyama, S.; Ito, H. Angew. Chem. Int. Ed. 2018, 57, 7196. (c) Sakaguchi, H.; Uetake, Y.; Ohashi, M.; Niwa, T.; Ogoshi, S.; Hosoya, T. J. Am. Chem. Soc. 2017, 139, 12855. (d) Sakaguchi, H.; Ohashi, M.; Ogoshi, S. Angew. Chem. Int. Ed. 2018, 57, 328. (e) Kojima, Y.; Takata, T.; Hirano, K.; Miura, M.

- Chem. Lett. **2020**, 49, 637. (f) For a related review, see: Wang, M.; Shi, Z. Chem. Lett. **2021**, 50, 553.
- (45) Mikami, K.; Itoh, Y.; Yamanaka, M. Chem. Rev. 2004, 104, 1.
- (46) Kojima, Y.; Miura, M.; Hirano, K. ACS Catal. 2021, 11, 11663.
- (47) Goff, C. M.; Matchette, M. A.; Shabestary, N.; Khazaeli, S. Polyhedron 1996, 15, 3897.
- (48) Paioti, P. H. S.; del Pozo, J.; Mikus, M. S.; Lee, J.; Koh, M. J.; Romiti, F.; Torker, S.; Hoveyda, A. H. J. Am. Chem. Soc. 2019, 141, 19917.
- (49) Wu, J.; Wu, H.; Li, X.; Liu, X.; Zhao, Q.; Huang, G.; Zhang, C. Angew. Chem. Int. Ed. 2021, 60, 20376.
- (50) Yamaguchi, D.; Nakamura, I.; Nishimura, T. Chem. Commun. 2021, 57, 11787.
- (51) (a) Guo, S.; AbuSalim, D. I.; Cook, S. P. J. Am. Chem. Soc. 2018, 140, 12378. (b) Sarver, P. J.; Bacauanu, V.; Schults, D. M.; DiRocco, D. A.; Lam, Y.-h.; Shere, E. C.; MacMillan, D. W. C. Nat. Chem. 2020, 12, 459. (c) He, J.; Nguyen, T. N.; Guo, S.; Cook, S. P. Org. Lett. 2021, 23, 702

#### **Biosketches**



Koji Hirano studied chemistry at Kyoto University where he obtained his B. Eng. (2003), M. Eng. (2005), and Dr. Eng. (2008) degrees under the supervision of Professor Koichiro Oshima. He subsequently worked as a postdoctoral fellow with Professor Tamio Hayashi at Kyoto University from April to September 2008. He then joined the research group of Professor Masahiro Miura at Osaka University as an assistant professor in October 2008, and was promoted to an associate professor in April 2015 and a full professor in May 2022. His current research interests are organic synthesis and organometallic chemistry, particularly the development of new synthetic strategies promoted by transition-metal catalysts and uniquely reactive heteroatom species.