

Title	Synthesis of Diarylselenides through Rh- Catalyzed Direct Diarylation of Elemental Selenium with Benzamides		
Author(s)	Xu-Xu, Qing Feng; Nishii, Yuji; Miura, Masahiro		
Citation	Journal of Organic Chemistry. 2022, 87(24), p. 16887–16894		
Version Type	АМ		
URL	https://hdl.handle.net/11094/92806		
rights	This document is the Accepted Manuscript version of a Published Work that appeared in final form in Journal of Organic Chemistry, © American Chemical Society after peer review and technical editing by the publisher. To access the final edited and published work see https://doi.org/10.1021/acs.joc.2c02131.		
Note			

Osaka University Knowledge Archive : OUKA

https://ir.library.osaka-u.ac.jp/

Osaka University

Synthesis of Diarylselenides through Rh-catalyzed Direct Diarylation of Elemental Selenium with Benzamides

Qing-Feng Xu-Xu,[†] Yuji Nishii,^{*,‡} Masahiro Miura^{*,†}

[†] Innovative Catalysis Science Division, Institute for Open and Transitionary Research Initiative (ICS-OTRI), Osaka University, Suita, Osaka 565-0871, Japan

[‡] Department of Applied Chemistry, Graduate School of Engineering, Osaka University, Suita, Osaka 565-0871, Japan



ABSTRACT: Diarylselenides are a representative class of molecules in organoselenium compounds. We herein report a Rh-catalyzed direct diarylation of selenium with benzamide derivatives. The use of elemental selenium as the Se source is intriguing in the view of atom economy, cost, stability, and handling. A series of diarylselenides with amide moieties were readily accessible through the directed C–H activation. The intermediacy of electrophilic Se(IV) species was indicated by the control experiments.

Selenium-containing compounds have gained diverse interests from chemical and pharmaceutical communities due to their versatile therapeutic activity and unique redox behavior.¹⁻⁴ Diaryselenide is a representative key structure for various bioactive compounds involving 5-LOX inhibitors, RAR agonists, cancer cell growth inhibitors, etc. (Scheme 1A).⁵⁻⁶ The redox activity and the nucleophilicity of Se atom can be modulated by changing aryl substituents.

Scheme 1. Synthetic Approaches to Diarylselenides



Conventional synthetic methods have relied on the use of reactive selenium species such as arylselenyl halides,7 diaryldiselenides,8 and aryl selenocyanates.9 These reagents are, however, usually prepared via cumbersome processes, difficult to handle, and of limited function group tolerance (Scheme 1B).^{10,11} Thus, development of methods approaching diarylselenides in an efficient and straightforward fashion is highly desired. Undoubtably, utilization of elemental selenium (or selenium powder) as the Se source is intriguing in view of atom economy, cost, stability, and handling.¹² Reported procedures for the arylation of elemental selenium utilize haloarenes,¹³ arvlboronic acids,14 or activated electron-rich arenes (electrophilic aromatic substitution)¹⁵ as the aryl sources, while, to the best of our knowledge, there is no report on diarylation through C-H bond cleavage of unactivated aromatic compounds. On the other hand, a first direct catalytic C-H selenium annulation was reported by Nishihara in 2017 with the aid of Ni catalyst and 8aminoquinoline bidentate directing group.^{8b} Our group also established a direct annulation of secondary amides with elemental selenium adopting Rh catalyst.16 As continuous interest in this field, we herein report a Rh-catalyzed diarylation of selenium with benzamide derivatives (Scheme 1C). For the C-H activation of N-arylbenzamides, the site-selectivity is usually not a trivial issue,¹⁷ whereas the developed protocol showed good chemoselectivity and regioselectivity. The intermediacy of electrophilic Se(IV) species was indicated by the control experiments.

At the outset, we examined the reaction of a secondary amide 1a with selenium powder under the conditions with a standard Cp*Rh catalyst (Table 1). The desired product 2a was obtained in 15% yield using [Cp*RhCl₂]₂ (2.0 mol %), AgSbF₆ (20 mol %), AgOAc (2.2 equiv), and AcOH (2 equiv) in PhCF₃ solvent at 80 °C (entry 1). The connectivity of the Se atom and the aniline fragments within 2a was confirmed by an X-ray crystallographic analysis (CCDC 2182413). The product yield was not considerably improved with other metal catalysts (entries 2-4). To our delight, an electron-deficient Cp^E ligated was effective for this reaction, giving 2a in 36% yield (entry 5). We then screened the solvent (see the Supporting Information for additional data). Several solvents showed higher productivity and HFIP (hexafluoro-isopropanol) was the most effective to afford 2a in 56% yield (entry 8). Replacement of AgOAc with AgTFA slightly improved the product yield (entry 9). The reaction could be conducted at 60 °C with comparable efficiency (entries 10-11). This reaction was successfully proceeded in 1.0 mmol scale (Scheme 2).

With the optimized conditions of Table 1 entry 11, the scope of secondary N-arylbenzamides was evaluated (Scheme 3). N-Phenylbenzamide (1b) reacted smoothly to afford the desired product 2b in 59% yield. For the aniline moiety, halogen (1c-1e) as well as alkyl (1f) substituents were compatible to give the corresponding selenides in moderate yields. For the benzoyl fragment, para-substituted (1g, 1h, 1j), 3,5-dimethyl (1l), thienyl (1i), and naphtyl (1k) substrates were successfully converted to the target products. Interestingly, an acetanilide 1m was also applicable to the present reaction albeit with lower efficiency. Adopting sterically demanding pivaloyl (1n) and 1adamantanecarbonyl (10) groups, the target compounds were obtained in relatively higher yields. The pivaloyl group could be removed under acidic conditions to afford the corresponding aniline 1n-1 (Scheme 4). Unfortunately, the reaction of metaand ortho-substituted benzamides resulted in rather complicated, and the product yield could not be determined (not shown).

Scheme 3. Scope of N-Arylbenzamides Substrates ^a



Table 1. Optimization of the Reaction Cond	itions ^a
	Q

20	$Cp^*Co(CO)I_2$	PhCF ₃	18
3	[Cp*IrCl ₂] ₂	PhCF ₃	8
4	[Ru(p-cymene)Cl ₂] ₂	PhCF ₃	13
5	[Cp ^E RhCl ₂] ₂	PhCF ₃	36
6	[Cp ^E RhCl ₂] ₂	CHCl ₃	50
7	[Cp ^E RhCl ₂] ₂	TFE	46
8	[Cp ^E RhCl ₂] ₂	HFIP	56
9 ^d	[Cp ^E RhCl ₂] ₂	HFIP	61 (59)
10 ^{d,f}	[Cp ^E RhCl ₂] ₂	HFIP	63 (64)
11 ^{d,f,g}	[Cp ^E RhCl ₂] ₂	HFIP	65 (68)

^{*a*} Reaction conditions: **1a** (0.2 mmol), Se (0.6 mmol), catalyst (2.0 mol%), AgSbF₆ (0.04 mmol), AgOAc (0.44 mmol), and AcOH (0.4 mmol) in solvent (0.4 mL) at 80 °C for 24 h. ^{*b*} Determined by NMR analysis. Isolated yield in parentheses. ^{*c*} With 4.0 mol% of catalyst. ^{*d*} AgTFA was used instead of AgOAc. ^{*e*} With 1.0 mol% of catalyst and 10 mol% of AgSbF₆. ^{*f*} 60 °C. ^{*g*} 20 h.

Scheme 2. Mmol-Scale Synthesis





^{*a*} Reaction conditions: **1** (0.2 mmol), Se (0.6 mmol), $[Cp^{E}RhCl_{2}]_{2}$ (2.0 mol %), AgTFA (0.44 mmol), and AcOH (0.4 mmol) in HFIP (0.4 mL) at 60 °C for 20 h.

Scheme 4. Protecting Group Removal



Next, we examined the direct selenation of *N*,*N*-disubstituted benzamides (Scheme 5). For these class of compounds, the standard Cp*Rh catalyst was considerably effective, and a model reaction of **3a** and selenium powder produced the target product **4a** in 70% yield using [Cp*Rh(MeCN)₃][SbF₆]₂ (10 mol%) as the catalyst and AgOAc (2.2 equiv) as the oxidant at 100 °C in PhCF₃ solvent (see the Supporting Information for the additional data). A series of *N*,*N*-dialkyl amides **3b–3d** were smoothly reacted to give the corresponding selenides **4b–4d** in moderate yields. This protocol was also applicable to an *N*-al-kyl-*N*-aryl (**3e**) and an *N*,*N*-diaryl (**3f**) amides. The positional selectivity of selenation for **4f** was confirmed by transamidation: reaction of **4f** with *p*-toluidine afforded **4f-1**. In addition, 2-phe-nylpyridine (**5**) was successfully coupled with selenium to give the selenide **6** in 42% yield.¹⁸

Scheme 5. Scope of N,N-Disubstituted Benzamides^a



^{*a*} Reaction conditions: **3** (0.2 mmol), Se (0.6 mmol) [Cp*Rh(MeCN)₃][SbF₆]₂ (10 mol %), and AgOAc (0.44 mmol) in PhCF₃ (0.2 mL) at 100 °C for 24 h.

To gain some insights into the reaction mechanism, we conducted several control experiments (Scheme 6). Deuterium incorporation to 1a under the standard conditions was examined adopting AcOH- d_4 as a deuterium source. After 2 h in the absence of selenium powder, the recovered 1a had a considerable 25% deuterium content at the ortho positions, indicating the reversibility of C-H bond activation step (Scheme 6A). In our previous report, SeO(OAc)₂ formed by treating selenium powder with AgOAc was assumed to be an active electrophile for the direct selenation. To test the intermediacy of this Se(IV) species, we prepared SeO(OAc)₂ and subsequently treated it with 1a under the catalytic conditions. As expected, the product 2a was obtained in 57% yield (Scheme 6B-1). In sharp contrast, no reaction took place when Ag salts were removed by filtration in prior to react with 1a (Scheme 6B-2). Addition of extra AgCl after filtration restored the catalytic activity to give 2a in 58% yield (Scheme 6B-3). These results support the indispensability of Ag(I) salts for the C-Se bond formation while the mechanistic detail is unclear at this stage. In the case AgTFA was used instead of AgOAc for the similar experiment, 2a was obtained in 57% yield along with the corresponding selenoxide 2a-O (6% yield). Additionally, conversion of this selenoxide to 2a under the reaction conditions was confirmed (Scheme 6C), which indicated that 2a-O was involved in the mechanism as an intermediate.16c

Scheme 6. Control Experiments



On the basis of these results and information from reported literatures, a possible catalytic cycle for the reaction of 1 with selenium is proposed as shown in Scheme 7. Thus, a catalytically active species, which is assumed to be $[Cp^{E}Rh(OAc)][SbF_{6}]$ (OAc may be OCOCF₃) formed through anion exchange in the reaction medium, induces the coordination-assisted C-H activation of 1 to form a six-membered metallacycle A. Meanwhile, $SeO(OAc)_2$ (OAc may be OCOCF₃) is given by the disproportionation of elemental selenium and coordinates to the complex A. The nucleophilic carbon atom within B substitutes the leaving group on the Se atom to give C. This step should be facilitated by Ag salts according to the Scheme 6B while the detail is still unclear. The liberated **D** coordinate to another rhodacycle species A to form an intermediate E. The nucleophilic carbon atom would replace the OAc leaving group on the Se atom, eventually affording the primary product selenoxide 2-O. This would be reduced to the desired diarylselenide 2 by low valent selenium species.

Scheme 7. A Proposed Reaction Mechanism



In conclusion, we have developed a Rh-catalyzed diarylation of elemental selenium with benzamide derivatives. A series of diarylselenides are readily accessible through the directinggroup-assisted C–H activation. The reaction shows high regioselectivity, and the mechanistic study suggests that Se(IV) species are involved in the reactions in contrast to well-established Se(-II) species as active intermediates in the reported literatures. This work represents a new reaction mode in chalcogen chemistry and related works are performing in our laboratories.

EXPERIMENTAL SECTION

Nuclear magnetic resonance spectra were measured at 400 MHz (¹H NMR), at 100 MHz (¹³C{¹H} NMR), at 376 MHz (¹⁹F{¹H} NMR), and at 76 MHz (⁷⁷Se{¹H} NMR) in 5 mm NMR tubes. ¹H NMR chemical shifts were reported in ppm relative to the resonance of TMS (δ 0.00) or the residual solvent signals at δ 2.50 for DMSO-*d*₆. ¹³C{¹H} NMR chemical shifts were reported in ppm relative to the residual solvent signals at δ 77.2 for CDCl₃ and at δ 39.5 for DMSO-*d*₆. ⁷⁷Se{¹H} NMR chemical

shifts were reported in ppm relative to the external standard diphenyldiselenide at δ 460.0. High resolution mass spectra (HRMS) were recorded by APCI-TOF. Preparative gel permeation chromatography (GPC) was conducted with two in-line YMC-GPC T2000 preparative columns. Thin layer chromatography analyses were performed on commercial glass plates bearing a 0.25 mm layer of Merck silica gel 60F_{254}. Silica gel column chromatography was performed using Wakosil® C-200 (64~210 μm).

All manipulations were performed under N₂ using standard Schlenk techniques unless otherwise noted. Amide compounds were prepared by standard condensation of the corresponding acid chlorides and amines similarly to the literature:¹⁹ all **1** and **3** are known in literatures.²⁰ [Cp^ERhCl₂]₂ was prepared according to the literature procedure.²¹ All other reagents were purchased from suppliers and used without further purification.

Rh-catalyzed Synthesis of DiaryIselenides 2a (Scheme 2). To a screw-top glass tube were added $[Cp^ERhCl_2]_2$ (17.0 mg, 0.02 mmol, 2.0 mol%), AgSbF₆ (68.7 mg, 0.2 mmol, 20 mol%), anilide **1a** (229 mg, 1.0 mmol), Se (237 mg, 3.0 mmol, 3.0 equiv), and AgTFA (480 mg, 2.2 mmol, 2.2 equiv). HFIP (2.0 mL) and AcOH (114 µL, 2.0 mmol, 2.0 equiv) were added via syringe. The suspension was stirred at 60 °C using aluminum heating blocks for 20 h under N₂. The resulting mixture was diluted with CHCl₃ and filtered through a pad of Celite eluting with CHCl₃ and EtOAc. The filtrate was concentrated in vacuo. The residue was purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 7/1/1) to give **2a** as white solid (163.2 mg, 61% yield).

General Procedure for the Rh-catalyzed Synthesis of Diarylselenides 2 (Scheme 3). To a screw-top glass tube were added $[Cp^ERhCl_2]_2$ (3.4 mg, 4.0×10^{-3} mmol, 2.0 mol%), AgSbF₆ (13.7 mg, 0.04 mmol, 20 mol%), anilide 1 (0.2 mmol), Se (47.4 mg, 0.6 mmol, 3.0 equiv), and AgTFA (97.2 mg, 0.44 mmol, 2.2 equiv). HFIP (0.4 mL, 0.5 M) and AcOH (23 µL, 0.4 mmol, 2.0 equiv) were added via syringe. The suspension was stirred at 60 °C using aluminum heating blocks for 20 h under N₂. The resulting mixture was diluted with CHCl₃ and filtered through a pad of Celite eluting with CHCl₃ and EtOAc. The filtrate was concentrated in vacuo. The residue was purified by silica gel column chromatography or preparative TLC and, if indicated, by GPC.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(4-fluoroben-

zamide) (2a) Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 7/1/1), white solid (36.6 mg, 68% yield), single crystals suitable for the X-ray crystallographic analysis were obtained by slow evaporation from the CHCl₃ solution, m.p. 193-194 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.28 (s, 2H), 8.06 (d, *J* = 8.0 Hz, 2H), 7.79-7.57 (m, 4H), 7.30-7.23 (m, 2H), 7.17 (dd, *J* = 8.4, 1.6 Hz, 2H), 7.14-7.03 (m, 4H), 2.20 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.1 (d, *J* = 251.5 Hz), 164.5, 136.1, 136.0, 135.0, 130.9, 130.7 (d, *J* = 3.1 Hz), 129.6 (d, *J* = 9.0 Hz), 123.2, 120.6, 116.0 (d, *J* = 22.0 Hz), 20.8; ¹⁹F{¹H} NMR (376 MHz, CDCl₃) δ -107.1; ⁷⁷Se{¹H} NMR (76 MHz, CDCl₃) δ 264.1; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₈H₂₃F₂N₂O₂Se 537.0887; Found 537.0879.

N,*N*'-(selenobis(2,1-phenylene))dibenzamide (2b). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 7/1/1), white solid (27.8 mg, 59% yield), m.p. 165-166 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.48 (s, 2H), 8.31 (dd, *J* = 8.0, 1.2 Hz, 2H), 7.78-7.70 (m, 4H), 7.57-7.49 (m, 2H), 7.49-7.36 (m, 8H), 7.04 (td, *J* = 7.6, 1.6 Hz, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.7, 138.6, 134.8, 134.5, 132.3, 130.2, 129.0, 127.2, 125.9, 122.9, 120.9; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₆H₂₁N₂O₂Se 473.0763; Found 473.0764.

N,N'-(selenobis(4-chloro-2,1-phenylene))dibenzamide (2c). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 3/1/1), white solid (31.7 mg, 59% yield), m.p. 220-221 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.34 (s, 2H), 8.25 (d, *J* = 8.8 Hz, 2H), 7.77-7.68 (m, 4H), 7.59-7.52 (m, 2H), 7.51-7.44 (m, 6H), 7.38 (dd, *J* = 8.4, 2.4 Hz, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 137.3, 134.2, 134.1, 132.5, 130.8, 130.6, 129.1, 127.1, 124.1, 121.6; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₆H₁₉Cl₂N₂O₂Se 540.9983; Found 540.9999.

N,*N*'-(selenobis(4-bromo-2,1-phenylene))dibenzamide (2d). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 6/1/1 to 3/1/1), white solid (26.0 mg, 41% yield), m.p. 231-233 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.34 (s, 2H), 8.21 (d, *J* = 8.8 Hz, 2H), 7.75-7.68 (m, 4H), 7.62 (d, *J* = 2.0 Hz, 2H), 7.60-7.43 (m, 8H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.6, 137.9, 137.0, 134.1, 133.6, 132.6, 129.2, 127.1, 124.4, 121.7, 118.3; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₆H₁₉Br₂N₂O₂Se 628.8973; Found 628.8965.

N,N'-(selenobis(4-fluoro-2,1-phenylene))dibenzamide (2e). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 6/1/1 to 4/1/1), white solid (14.1 mg, 28% yield), m.p. 198-199 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.27 (s, 2H), 8.15 (dd, *J* = 8.4, 5.2 Hz, 2H), 7.82-7.69 (m, 4H), 7.59-7.50 (m, 2H), 7.50-7.42 (m, 4H), 7.18-7.06 (m, 4H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 165.8, 159.7 (d, *J* = 249.6 Hz), 134.6 (d, *J* = 3.0 Hz), 134.1, 132.4, 129.1, 127.2, 125.1 (d, *J* = 7.8 Hz), 123.1 (d, *J* = 7.1 Hz), 121.0 (d, *J* = 24.0 Hz), 117.2 (d, *J* = 22.1 Hz); ¹⁹F {¹H} NMR (376 MHz, CDCl₃) δ -114.8; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₆H₁₉F₂N₂O₂Se 509.0574; Found 509.0558.

N,N'-(selenobis(4-butyl-2,1-phenylene))dibenzamide (2f). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 20/1/1 to 10/1/1), white solid (33.4 mg, 57% yield), m.p. 175-176 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.44 (s, 2H), 8.18 (d, *J* = 8.0 Hz, 2H), 7.76-7.68 (m, 4H), 7.56-7.48 (m, 2H), 7.47-7.37 (m, 4H), 7.28 (d, *J* = 2.0 Hz, 2H), 7.18 (dd, *J* = 8.4, 2.0 Hz, 2H), 2.42 (t, *J* = 7.6 Hz, 4H), 1.54-1.34 (m, 4H), 1.24 (h, *J* = 7.2 Hz, 4H), 0.85 (t, *J* = 7.2 Hz, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.5, 140.7, 136.4, 134.7, 134.6, 132.1, 130.2, 128.9, 127.2, 122.7, 120.4, 34.9, 33.4, 22.3, 14.0; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₃₄H₃₇N₂O₂Se 585.2015; Found 585.2015.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(4-chloroben-

zamide) (2g). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 7/1/1), white solid (30.8 mg, 54% yield), m.p. 251-252 °C; ¹H NMR (400 MHz, DMSO-*d*₆) δ

10.07 (s, 2H), 7.89 (d, J = 8.8 Hz, 4H), 7.57 (d, J = 8.8 Hz, 4H), 7.39 (d, J = 7.6 Hz, 2H), 7.25-7.07 (m, 4H), 2.19 (s, 6H); ¹³C {¹H} NMR (100 MHz, DMSO- d_6) δ 164.5, 136.5, 136.3, 135.5, 134.3, 132.8, 129.5, 129.1, 128.6, 128.1, 126.3, 20.4; HRMS (APCI) m/z: [M+H]⁺ Calcd for C₂₈H₂₃Cl₂N₂O₂Se 569.0296; Found 569.0283.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(4-methylben-

zamide) (2h). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 7/1/1), white solid (31.8 mg, 60% yield), m.p. 215-217 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.36 (s, 2H), 8.16 (d, *J* = 8.4 Hz, 2H), 7.60 (d, *J* = 8.4 Hz, 4H), 7.29 (d, *J* = 1.2 Hz, 2H), 7.21 (d, *J* = 7.6 Hz, 4H), 7.17 (dd, *J* = 8.4, 1.6 Hz, 2H), 2.39 (s, 6H), 2.19 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.5, 142.6, 136.4, 135.4, 135.2, 131.8, 130.9, 129.5, 127.1, 122.7, 120.3, 21.6, 20.8; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₃₀H₂₉N₂O₂Se 529.1389; Found 529.1402.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(thiophene-2-

carboxamide) (2i). Purified by preparative TLC (*n*-hexane/EtOAc = 3/1), white semisolid (26.8 mg, 52% yield); ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 2H), 8.05 (d, J = 8.4 Hz, 2H), 7.51 (dd, J = 4.8, 1.2 Hz, 2H), 7.47 (dd, J = 3.6, 1.2 Hz, 2H), 7.30 (d, J = 1.6 Hz, 2H), 7.16 (dd, J = 8.4, 1.6 Hz, 2H), 7.08 (dd, J = 4.8, 3.6 Hz, 2H), 2.21 (s, 6H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 160.0, 139.0, 135.84, 135.81, 135.2, 131.1, 130.9, 128.7, 128.0, 123.1, 120.6, 20.8; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₄H₂₁N₂O₂S₂Se 513.0204; Found 513.0184.

N,*N*'-(selenobis(2,1-phenylene))bis(4-methoxybenzamide)

(2j). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 5/1/1), white solid (21.2 mg, 40% yield), m.p. 191-193 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.44 (s, 2H), 8.31 (dd, *J* = 8.4, 1.2 Hz, 2H), 7.76-7.67 (m, 4H), 7.45 (dd, *J* = 8.0, 1.6 Hz, 2H), 7.42-7.35 (m, 2H), 7.03 (td, *J* = 7.6, 1.2 Hz, 2H), 6.95-6.88 (m, 4H), 3.85 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 165.2, 162.8, 138.8, 134.8, 130.2, 129.1, 126.7, 125.6, 122.7, 120.8, 114.2, 55.6; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₈H₂₅N₂O₄Se 533.0974; Found 533.0959.

N,N'-(selenobis(2,1-phenylene))bis(2-naphthamide) (2k). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 6/1/1 to 5/1/1), white solid (29.2 mg, 51% yield), m.p. 215-216 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.61 (s, 2H), 8.33 (d, *J* = 9.2 Hz, 2H), 8.19 (d, *J* = 1.2 Hz, 2H), 7.89-7.82 (m, 6H), 7.76 (dd, *J* = 8.8, 1.6 Hz, 2H), 7.60-7.49 (m, 6H), 7.46-7.40 (m, 2H), 7.05 (td, *J* = 7.6, 1.6 Hz, 2H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 165.8, 138.6, 135.0, 134.8, 132.6, 131.6, 130.2, 129.2, 128.9, 128.1, 127.89, 127.87, 127.0, 126.0, 123.5, 123.1, 121.2; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₃₄H₂₅N₂O₂Se 573.1076; Found 573.1076.

N,N'-(selenobis(2,1-phenylene))bis(3,5-dimethylbenzamide) (21). Purified by silica gel column chromatography (*n*-hexane/EtOAc = 6/1 to 5/1), white solid (31.2 mg, 59% yield), m.p. 179-180 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.38 (s, 2H), 8.30 (dd, J = 8.4, 1.2 Hz, 2H), 7.46 (dd, J = 7.6, 1.2 Hz, 2H), 7.43-7.37 (m, 2H), 7.29 (s, 4H), 7.14 (s, 2H), 7.04 (td, J = 7.6, 1.2 Hz, 2H), 2.33 (s, 12H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 166.1, 138.7, 138.6, 134.8, 134.6, 133.9, 130.1, 125.8, 124.9, 122.9, 120.8, 21.4; HRMS (APCI) m/z: $[M+H]^+$ Calcd for $C_{30}H_{29}N_2O_2Se$ 529.1389; Found 529.1381.

N,*N*'-(selenobis(4-methyl-2,1-phenylene))diacetamide (2m). Purified by preparative TLC (*n*-hexane/EtOAc = 1/1), colorless oil (7.0 mg, 23% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.82 (d, J = 8.0 Hz, 2H), 7.65 (s, 2H), 7.22 (s, 2H), 7.14 (d, J = 8.0 Hz, 2H), 2.26 (s, 6H), 2.05 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 168.9, 136.0, 135.8, 134.7, 130.4, 124.0, 121.8, 24.2, 20.9; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₁₈H₂₁N₂O₂Se 377.0763; Found 377.0767.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(2,2-dime-

thylpropanamide) (2n). Purified by preparative TLC (*n*-hexane/EtOAc = 3/1), colorless oil (22.4 mg, 49% yield); ¹H NMR (400 MHz, CDCl₃) δ 8.03 (d, J = 8.4 Hz, 2H), 7.89 (s, 2H), 7.14 (dd, J = 8.4, 1.6 Hz, 2H), 7.11 (d, J = 1.6 Hz, 2H), 2.23 (s, 6H), 1.24 (s, 18H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 176.8, 135.7, 135.4, 134.2, 130.4, 122.8, 120.5, 40.0, 27.6, 20.9; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₄H₃₃N₂O₂Se 461.1702; Found 461.1714.

N,N'-(selenobis(4-methyl-2,1-phenylene))bis(adamantane-

1-carboxamide) (20). Purified by preparative TLC (*n*-hexane/EtOAc = 5/1), colorless oil (22.9 mg, 37% yield); ¹H NMR (400 MHz, CDCl₃) δ 8.09 (d, J = 8.0 Hz, 2H), 7.84 (s, 2H), 7.20-7.07 (m, 4H), 2.23 (s, 6H), 2.05 (s, 6H), 1.88-1.83 (m, 12H), 1.80-1.62 (m, 12H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 176.2, 135.8, 135.1, 134.2, 130.4, 122.5, 120.1, 41.9, 39.3, 36.5, 28.2, 20.9; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₃₆H₄₅N₂O₂Se 617.2641; Found 617.2659.

Removal of the Protecting Group (Scheme 4). To a screw-top glass tube were added **2n** (54.6 mg, 0.12 mmol), 20 w/w% aqueous H₂SO₄ (3.5 mL), and MeOH (3.5 mL). The suspension was stirred under N₂ atmosphere at 70 °C for 4 d. The resulting mixture was poured into water and washed with CHCl₃. The aqueous phase was neutralized with NaHCO₃aq and extracted with CHCl₃ three times. The combined organic extracts was concentrated in vacuo and purified by preparative TLC (*n*-hexane/EtOAc = 3/1) to give **2n-1** as yellow oil (23.5 mg, 68% yield).

2,2'-selenobis(4-methylaniline) (2n-1). ¹H NMR (400 MHz, CDCl₃) δ 7.21 (d, J = 1.6 Hz, 2H), 6.92 (dd, J = 8.0, 1.6 Hz, 2H), 6.64 (d, J = 8.0 Hz, 2H), 3.60 (br, 4H), 2.18 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 145.0, 135.8, 130.5, 128.7, 115.6, 114.5, 20.4; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₁₄H₁₇N₂Se 293.0551; Found 293.0543.

General Procedure for the Rh-catalyzed Synthesis of Diarylselenides 4 (Scheme 4). To a screw-top glass tube were added [Cp*Rh(MeCN)₃][SbF₆]₂ (16.6 mg, 0.02 mmol, 10 mol%), amide 3 (0.2 mmol), Se (47.4 mg, 0.6 mmol, 3.0 equiv), and AgOAc (73.4 mg, 0.44 mmol, 2.2 equiv.). PhCF₃ (0.2 mL, 1.0 M) was added via syringe. The suspension was stirred at 100 °C using aluminum heating blocks for 24 h under N₂. The resulting mixture was diluted with CHCl₃ and filtered through a pad of Celite eluting with CHCl₃ and EtOAc. The filtrate was concentrated in vacuo. The residue was purified by silica gel column chromatography or preparative TLC and, if indicated, by GPC.

2,2'-selenobis(4-methoxy-*N***,N-diethylbenzamide) (4a).** Purified by silica gel column chromatography (*n*-hexane/EtOAc = 1/1 to 1/2), colorless oil (34.3 mg, 70% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.18 (d, *J* = 8.4 Hz, 2H), 7.01 (d, *J* = 2.8 Hz, 2H), 6.81 (dd, *J* = 8.4, 2.4 Hz, 2H), 3.72 (s, 6H), 3.63-3.08 (m, 8H), 1.36-0.97 (m, 12H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 169.8, 159.9, 133.0, 129.8, 127.6, 119.8, 113.7, 55.5, 43.2, 39.2, 14.4, 12.9; HRMS (APCI) *m*/*z*: [M+H]⁺ Calcd for C₂₄H₃₃N₂O₄Se 493.1600; Found 493.1603.

2,2'-selenobis(4-methoxy-*N*,*N***-dimethylbenzamide)** (4b). Purified by silica gel column chromatography (EtOAc to EtOAc/MeOH = 20/1), colorless oil (20.8 mg, 48% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.20 (d, *J* = 8.4 Hz, 2H), 7.01 (d, *J* = 2.4 Hz, 2H), 6.83 (dd, *J* = 8.4, 2.4 Hz, 2H), 3.73 (s, 6H), 3.16-2.78 (m, 12H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 170.4, 160.1, 132.5, 130.0, 128.3, 120.0, 113.8, 55.6, 39.1, 35.0; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₀H₂₅N₂O₄Se 437.0974; Found 437.0969.

(selenobis(4-methoxy-2,1-phenylene))bis(pyrrolidin-1-

ylmethanone) (4c). Purified by silica gel column chromatography (EtOAc/MeOH = 20/1 to 10/1), colorless oil (25.9 mg, 53% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.25 (d, J = 8.4 Hz, 2H), 7.01 (d, J = 2.4 Hz, 2H), 6.81 (dd, J = 8.4, 2.4 Hz, 2H), 3.72 (s, 6H), 3.59 (t, J = 6.8 Hz, 4H), 3.27 (t, J = 6.4 Hz, 4H), 1.99-1.79 (m, 8H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 168.6, 160.1, 133.5, 130.2, 128.1, 119.9, 113.6, 55.5, 49.0, 45.8, 26.2, 24.6; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₄H₂₉N₂O₄Se 489.1287; Found 489.1287.

2,2'-selenobis(*N*,*N*-diethylbenzamide) (4d). Purified by preparative TLC (*n*-hexane/EtOAc = 1/1), colorless oil (18.2 mg, 42% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.44-7.41 (m, 2H), 7.33-7.17 (m, 6H), 3.57 (q, *J* = 6.8 Hz, 4H), 3.19 (q, *J* = 7.2 Hz, 4H), 1.27 (t, *J* = 7.2 Hz, 6H), 1.07 (t, *J* = 7.2 Hz, 6H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 169.7, 140.6, 134.9, 129.6, 128.5, 127.7, 126.5, 43.1, 39.1, 14.3, 12.9; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₂H₂₉N₂O₂Se 433.1389; Found 433.1376.

2,2'-selenobis(*N*-ethyl-*N*-phenylbenzamide) (4e). Purified by silica gel column chromatography (*n*-hexane/EtOAc = 1/1), colorless oil (32.2 mg, 56% yield, 0.219 mmol scale); ¹H NMR (400 MHz, CDCl₃) δ 7.25-6.90 (m, 18H), 3.99 (s, 4H), 1.25 (t, J = 6.8 Hz, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 169.7, 142.2, 140.1, 133.8, 130.6, 129.3, 129.1, 128.3, 128.2, 127.2, 126.5, 44.8, 13.2; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₃₀H₂₉N₂O₂Se 529.1389; Found 529.1388.

2,2'-selenobis(*N*,*N*-bis(4-methoxyphenyl)benzamide) (4f). Purified by silica gel column chromatography (*n*-hexane/EtOAc/CHCl₃ = 3/1/1 to 3/3/2), white foam (30.5 mg, 41% yield); ¹H NMR (400 MHz, CDCl₃) δ 7.31 (dd, *J* = 7.6, 1.2 Hz, 2H), 7.28-7.05 (m, 10H), 7.01 (td, *J* = 7.2, 1.6 Hz, 2H), 6.95 (dd, *J* = 8.0, 1.2 Hz, 2H), 6.74 (br, 8H), 3.73 (s, 12H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 170.3, 158.0, 140.1, 136.1, 134.4, 130.6, 129.6, 128.7, 127.9, 126.9, 114.3, 55.5; HRMS (APCI) m/z: [M+H]⁺ Calcd for C₄₂H₃₇N₂O₆Se 745.1811; Found 745.1858.

bis(2-(pyridin-2-yl)phenyl)selane (6). Purified by silica gel column chromatography (*n*-hexane/EtOAc = 1/1), white foam (16.1 mg, 42% yield); ¹H NMR (400 MHz, CDCl₃) δ 8.63 (ddd, J = 4.8, 1.6, 0.8 Hz, 2H), 7.65 (td, J = 7.6, 1.6 Hz, 2H), 7.57-7.48 (m, 4H), 7.45 (dd, J = 8.0, 1.2 Hz, 2H), 7.31 (td, J = 7.6, 1.2 Hz, 2H), 7.24-7.14 (m, 4H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 159.0, 148.7, 142.6, 136.0, 135.1, 133.5, 129.8, 129.1, 127.3, 123.5, 122.0; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₂H₁₇N₂Se 389.0551; Found 389.0557.

Transamidation (Scheme 5). To a screw-top glass tube were added **4f** (43.1 mg, 0.058 mmol), *p*-tolidine (45.3 mg, 0.232 mmol, 4.0 equiv.), LiHMDS (58.2 mg, 0.348 mmol, 6.0 equiv.), THF (0.2 mL), and toluene (0.6 mL) in glove-box. The suspension was stirred at rt for 24 h. After quenched with saturated NH₄Claq, the resulting mixture was extracted with CHCl₃ and concentrated in vacuo. The obtained crude material was purified by preparative TLC (hexane/EA = 10/1) to give **4f-1** as colorless oil (7.7 mg, 27% yield).

2,2'-selenobis(N-(*p***-tolyl)benzamide) (4f-1).** ¹H NMR (400 MHz, CDCl₃) δ 7.80 (s, 2H), 7.67 – 7.59 (m, 2H), 7.54 – 7.48 (m, 2H), 7.40 – 7.28 (m, 4H), 7.23 (d, J = 8.4 Hz, 4H), 7.05 (d, J = 8.4 Hz, 4H), 2.31 (s, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.3, 138.1, 135.2, 135.0, 134.3, 131.5, 131.0, 129.5, 129.1, 128.2, 120.1, 21.1; HRMS (APCI) *m/z*: [M+H]⁺ Calcd for C₂₈H₂₅N₂O₂Se 501.1076; Found 501.1065.

N,N'-(seleninylbis(4-methyl-2,1-phenylene))bis(4-fluoro-

benzamide) (2a-O) To a CHCl₃ solution of 2a (26.8 mg, 0.05 mmol in 1.0 mL) was added dropwise H2O2 (17.0 mg, 0.15 mmol, 30% w/w in water). The mixture was stirred at room temperature 18 h. After complete conversion of 2a (monitored by TLC) the resulting mixture was diluted with CHCl3 and washed by saturated Na₂S₂O₃aq and brine. The combined organic extracts was dried over Na₂SO₄ and concentrated in vacuo to give the title compound as white solid (27.5 mg, >99% yield). m.p. 223 °C (decomposed); ¹H NMR (400 MHz, DMSO-*d*₆) δ 11.37 (s, 2H), 8.07-7.89 (m, 4H), 7.58 (d, J = 8.0 Hz, 2H), 7.49-7.39 (m, 6H), 7.20 (dd, J = 8.4, 1.6 Hz, 2H), 1.84 (s, 6H); ${}^{13}C{}^{1}H{}$ NMR (100 MHz, DMSO- d_6) δ 164.8 (d, J = 250.2 Hz), 163.9, 136.5 (d, J = 3.8 Hz), 134.3, 132.4, 130.7 (d, J = 9.3 Hz), 130.6, 130.1, 129.0, 124.6, 116.1 (d, J = 22.0 Hz), 20.2; ¹⁹F {¹H} NMR (376 MHz, DMSO-d₆) δ -108.0; HRMS (APCI) m/z: [M+H]⁺ Calcd for C₂₈H₂₃F₂N₂O₃Se 553.0836; Found 553.0836.

ASSOCIATED CONTENT

The Supporting Information is available free of charge athttps://pubs.acs.org.xxx

 ^{1}H , $^{13}C{^{1}H}$, $^{19}F{^{1}H}$, and $^{77}Se{^{1}H}$ NMR spectra for all products, detailed optimization studies, and X-ray crystallographic analysis (PDF)

AUTHOR INFORMATION

Corresponding Authors

Masahiro Miura - Innovative Catalysis Science Division, Institute for Open and Transdisciplinary Research Initiatives (ICS-OTRI), Osaka University, Suita, Osaka 565-0871, Japan; orcid.org/0000-0001-8288-6439;

Email: miura@chem.eng.osaka-u.ac.jp

Yuji Nishii - Department of Applied Chemistry, Graduate School of Engineering, Osaka University, Suita, Osaka 565-0871, Japan; orcid.org/0000-0002-6824-0639;

Email: y_nishii@chem.eng.osaka-u.ac.jp

Author

Qing-Feng Xu-Xu - Innovative Catalysis Science Division, Institute for Open and Transdisciplinary Research Initiatives (ICS-OTRI), Osaka University, Suita, Osaka 565-0871, Japan; orcid.org/0000-0003-4318-1828

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

Supporting Information.

This work was supported by the Japan Society for the Promotion of Science (JSPS) KAKENHI Grants JP 21K14627 (Grant-in-Aid for Young Scientists) to Y.N., JP 17H06092 (Grant-in-Aid for Specially Promoted Research) to M.M.

REFERENCES

(1) For selected reviews, see: (a) Mugesh, G.; du Mont, W.-W.; Sies, H. Chemistry of Biologically Important Synthetic Organoselenium Compounds. *Chem. Rev.* **2001**, *101*, 2125-2180. (b) Ruberte, A. C.; Sanmartin, C.; Aydillo, C.; Sharma, A. K.; Plano, D. Development and Therapeutic Potential of Selenazo Compounds. *J. Med. Chem.* **2020**, *63*, 1473-1489. (c) Santi, C.; Scimmi, C.; Sancineto, L. Ebselen and Analogues: Pharmacological Properties and Synthetic Strategies for Their Preparation. *Molecules* **2021**, *26*, 4230. (d) Hou, W.; Dong, H.; Zhang, X.; Wang, Y.; Su, L.; Xu, H. Selenium as an emerging versatile player in heterocycles and natural products modification. *Drug Discov. Today* **2022**, *27*, 2268.

(2) (a) Reich, H. J.; Hondal, R. J. Why Nature Chose Selenium. ACS Chem. Biol. **2016**, 11, 821-841. (b) Hou, W.; Xu, H. Incorporating Selenium into Heterocycles and Natural Products From Chemical Properties to Pharmacological Activities. J. Med. Chem. **2022**, 65, 4436-4456.

(3) For selected reviews on application in catalysis or organic synthesis, see: (a) Shao, L.; Li, Y.; Lu, J.; Jiang, X. Recent progress in selenium-catalyzed organic reactions. Org. Chem. Front. 2019, 6, 2999-3041. (b) Liao, L.; Zhao, X. Selenium-catalyzed Functionalization of Alkynes. Chem. Lett. 2021, 50, 1104-1113. (c) Liao, L.; Zhao, X. Modern Organoselenium Catalysis: Opportunities and Challenges. Synlett 2021, 32, 1262-1268. For selected examples recently, see: (d) Wang, W.; Zhu, H.; Liu, S.; Zhao, Z.; Zhang, L.; Hao, J.; Wang, Y. Chalcogen-Chalcogen Bonding Catalysis Enables Assembly of Discrete Molecules. J. Am. Chem. Soc. 2019, 141, 9175-9179. (e) Chisholm, T. S.; Clarke, R. J.; Kulkarni, S. S.; Payne, R. J. Peptide Ligation at High Dilution via Reductive Diselenide-Selenoester Ligation. J. Am. Chem. Soc. 2020, 142, 1090-1100. (f) Panigrahi, H. N. R.; Arora, P. S. Two-Component Redox Organocatalyst for Peptide Bond Formation. J. Am. Chem. Soc. 2022, 144, 3637-3643. (g) Yuan, X.; Wang, Y. Synthesis and Characterization of a Selenide Catalyst for the Activation of Alkenes through Se... n Bonding. Angew. Chem. Int. Ed. 2022, DOI: 10.1002/anie.202203671.

(4) For selected reviews on application in synthesis of functional molecules, see: (a) Manjare, S. T.; Kim, Y.; Churchill, D. G. Selenium-

and Tellurium-Containing Fluorescent Molecular Probes for the Detection of Biologically Important Analytes. *Acc. Chem. Res.* **2014**, *47*, 2985-2998. (b) Brutchey, R. L. Diorganyl Dichalcogenides as Useful Synthons for Colloidal Semiconductor Nanocrystals. *Acc. Chem. Res.* **2015**, *48*, 11, 2918-2926.

(5) (a) Engman, L.; Stern, D.; Frisell, H.; Vessman, K.; Berglund, M.; Ek, B.; Andersson, C.-M. Synthesis, antioxidant properties, biological activity and molecular modelling of a series of chalcogen analogues of the 5-lipoxygenase inhibitor DuP 654. Bioorg. Med. Chem. 1995, 3, 1255-1262. (b) Kumar, S.; Sharma, N.; Maurya, I. K.; Bhasin, A. K. K.; Wangoo, N.; Brandão, P.; Félix, V.; Bhasin, K. K.; Sharma, R. K. Facile synthesis, structural evaluation, antimicrobial activity and synergistic effects of novel imidazo[1,2-a]pyridine based organoselenium compounds. Eur. J. Med. Chem. 2016, 123, 916-924. (c) Casaril, A. M.; Domingues, M.; Fronza, M.; Vieira, B. Begnini, K.; Lenardão, E. J.; Seixas, F. K.; Collares, T.; Nogueira, C. W.; Savegnago, L. Antidepressant-like effect of a new selenium-containing compound is accompanied by a reduction of neuroinflammation and oxidative stress in lipopolysaccharide-challenged mice. J. Psychopharmacol. 2017, 31, 1263-1273. (d) Domingues, M.; Casaril, A. M.; Birmann, P. T.; Lourenço, D.; Vieira, B.; Begnini, K.; Lenardão, E. J.; Collares, T.; Seixas, F. K.; Savegnago, L. Front. Neurosci. 2018, 12, 486. (e) Rodrigues, J.; Saba, S.; Joussef, A. C.; Rafique, J.; Braga, A. L. KIO3-Catalyzed C(sp²)-H Bond Selenylation/Sulfenylation of (Hetero)arenes: Synthesis of Chalcogenated (Hetero)arenes and their Evaluation for Anti-Alzheimer Activity. Asian J. Org. Chem. 2018, 7, 1819-1824

(6) (a) Luo, J.; Zhu, Z.; Liu, Y.; Zhao, X. Diaryl Selenide Catalyzed Vicinal Trifluoromethylthioamination of Alkenes. Org. Lett. 2015, 17, 3620-3623. (b) Verma, A.; Jana, S.; Prasad, C. D.; Yadav, A.; Kumar, S. Organoselenium and DMAP co-catalysis: regioselective synthesis of medium-sized halolactones and bromooxepanes from unactivated alkenes. Chem. Commun. 2016, 52, 4179-4182. (c) Zhu, Z.; Luo, J.; Zhao, X. Combination of Lewis Basic Selenium Catalysis and Redox Selenium Chemistry: Synthesis of Trifluoromethylthiolated Tertiary Alcohols with Alkenes. Org. Lett. 2017, 19, 4940-4943. (d) Rode, K.; Palomba, M.; Ortgies, S.; Rieger, R.; Breder, A. Aerobic Allylation of Alcohols with Non-Activated Alkenes Enabled by Light-Driven Selenium-π-Acid Catalysis. Synthesis 2018, 50, 3875-3885. (e) Gieuw, M. H.; Leung, V. M. Y.; Ke, Z.; Yeung, Y. Y. Electrophilic Bromolactonization of Cyclopropyl Diesters Using Lewis Basic Chalcogenide Catalysts. Adv. Synth. Catal. 2018, 360, 4306-4311. (f) Nalbandian, C. J.; Brown, Z. E.; Alvarez, E.; Gustafson, J. L. Lewis Base/Bronsted Acid Dual-Catalytic C-H Sulfenylation of Aromatics. Org. Lett. 2018, 20, 3211-3214. (g) Wei, W.; Zhao, X. Organoselenium-Catalyzed Cross-Dehydrogenative Coupling of Alkenes and Azlactones. Org. Lett. 2022, 24, 1780-1785.

(7) (a) Freitas, C. S.; Barcellos, A. M.; Ricordi, V. G.; Pena, J. M.; Perin, G.; Jacob, R. G.; Lenardão, E. J.; Alves, D. Synthesis of diaryl selenides using electrophilic selenium species and nucleophilic boron reagents in ionic liquids. Green Chem. 2011, 13, 2931-2938. (b) Kundu, D.; Ahammed, S.; Ranu, B. C. Microwave-Assisted Reaction of Aryl Diazonium Fluoroborate and Diaryl Dichalcogenides in Dimethyl Carbonate: A General Procedure for the Synthesis of Unsymmetrical Diaryl Chalcogenides, Green Chem. 2012, 14, 2024-2030. (c) Yu, S.; Wan, B.; Li, X. Rh(III)-Catalyzed Selenylation of Arenes with Selenenyl Chlorides/ Diselenides via C-H Activation. Org. Lett. 2015, 17, 58-61. (d) Shu, S.; Fan, Z.; Yao, Q.; Zhang, A. Ru(II)-Catalyzed Direct C(sp²)-H Activation/Selenylation of Arenes with Selenyl Chlorides. J. Org. Chem. 2016, 81, 5263-5269. (e) He, M.; Gu, L.; Tan, Y.; Wang, Y.; Wang, Y.; Zhang, C.; Ma, W. Palladium-Catalyzed Distal C-H Selenylation of 2-Aryl Acetamides with Diselenides and Selenyl Chlorides. Adv. Synth. Catal. 2020, 362, 5708-5715.

(8) (a) Ricordi, V. G.; Freitas, C. S.; Perin, G.; Lenardão, E. J.; Jacob, R. G.; Savegnago, L.; Alves, D. Glycerol as a Recyclable Solvent for Copper-Catalyzed Cross-Coupling Reactions of Diaryl Diselenides

with Aryl Boronic Acids, Green Chem. 2012, 14, 1030-1034. (b) Iwasaki, M.; Tsuchiya, Y.; Nakajima, K.; Nishihara, Y. Chelate-Assisted Direct Selenation of Aryl C-H Bonds with Diselenides Catalyzed by Palladium. Org. Lett. 2014, 16, 4920-4923. (c) Qiu, R.; Reddy, V. P.; Iwasaki, T.; Kambe, N. The Palladium-Catalyzed Intermolecular C-H Chalcogenation of Arenes. J. Org. Chem. 2015, 80, 367-374. (d) Zhang, Q.-B.; Ban, Y.-L.; Yuan, P.-F.; Peng, S.-J.; Fang, J.-G.; Wu, L.-Z.; Liu, O. Visible-light-mediated aerobic selenation of (hetero)arenes with diselenides. Green Chem. 2017, 19, 5559-5563. (e) Gandeepan, P.; Koeller, J.; Ackermann, L. Expedient C-H Chalcogenation of Indolines and Indoles by Positional-Selective Copper Catalysis. ACS Catal. 2017, 7, 1030-1034. (f) Reddy, C. R.; Ranjan, Ravi.; Prajapti, S. K. Copper-Catalyzed Intramolecular Chalcogenoamination of Enynyl Azides: Synthesis of 5-Selenyl/Sulfenyl Nicotinates. Org. Lett. 2019, 21, 623-626. (g) Sun, L.; Wang, L.; Alhumade, H.; Yi, H.; Cai, H.; Lei, A. Electrochemical Radical Selenylation of Alkenes and Arenes via Se-Se Bond Activation. Org. Lett. 2021, 23, 7724-7729. (h) Kajiwara, R.; Takamatsu, K.; Hirano, K.; Miura, M. Copper-Mediated Regioselective C-H Sulfenylation and Selenation of Phenols with Phenanthroline Bidentate Auxiliary. Org. Lett. 2020, 22, 5915-5919. (i) Kumar, M.; Raziullah; Ahmad, A.; Dutta, H. S.; Khan, A. A.; Rastogi, A.; Kant, R.; Koley, D. Cu(II)-Catalyzed C-N, C-O, C-Cl, C-S, and C-Se Bond Formation via C(sp²)-H Activation Using 7-Azaindole as an Intrinsic Directing Group. J. Org. Chem. 2021, 86, 15185-15202.

(9) Guan, Y.; Townsend, S. D. Metal-Free Synthesis of Unsymmetrical Organoselenides and Selenoglycosides. *Org. Lett.* **2017**, *19*, 5252-5255.

(10) (a) Kumar, R. U.; Reddy, K. H. V.; Satish, G. Swapna, K.; Nageswar, Y.V.D. Metal free synthesis of diaryl selenides using SeO₂ as a selenium source. Tetrahedron Lett. 2016, 57, 4138-4141. (b) Cao, Y.; Liu, J.; Liu, F.; Jiang, L.; Yi, W. Copper-catalyzed direct and odorless selenylation with a sodium selenite-based reagent. Org. Chem. Front. 2019, 6, 825-829. (c) Kadu, R.; Batabyal, M.; Kadyan, H.; Kumar, S. An efficient copper-catalyzed synthesis of symmetrical bis(Narylbenzamide) selenides and their conversion to hypervalent spirodiazaselenuranes and hydroxy congeners. Dalton Trans. 2019, 48, 7249-7260. (d) Abenante, L.; Padilha, N. B.; Anghinoni, J. M.; Penteado, F.; Rosati, O.; Santi, C.; Silva, M. S.; Lenardão, E. J. Arylseleninic acid as a green, bench-stable selenylating agent: synthesis of selanylanilines and 3-selanylindoles. Org. Biomol. Chem. 2020, 18, 5210-5217. (e) Redon, S.; Remusat, V.; Vanelle, P. Green Synthesis of Diaryl Selenides from Arylboronic Acids and Arylseleninic Acids. Synlett 2022; 33, 483-487.

(11) For selected reviews, see: (a) Rampon, D. S.; Luz, E. Q.; Lima, D. B.; Balaguez, R. A.; Schneider, P. H.; Alves, D. Transition metal catalysed direct selanylation of arenes and heteroarenes. *Dalton Trans.* **2019**, *48*, 9851-9905. (b) Ma, W.; Kaplaneris, N.; Fang, X.; Gu, L.; Mei, R.; Ackermann, L. Chelation-assisted transition metal-catalysed C-H chalcogenylations. *Org. Chem. Front.* **2020**, *7*, 1022-1060. (c) Sonawane, A. D.; Sonawane, R. A.; Ninomiya, M.; Koketsu, M. Diorganyl diselenides: a powerful tool for the construction of selenium containing scaffolds. *Dalton Trans.* **2021**, *50*, 12764-12790. (d) Barcellos, A. M.; Sacramento, M.; da Costa, G. P.; Perin, G.; Lenardão, E. J.; Alves, D. Organoboron compounds as versatile reagents in the transition metal-catalyzed C-S, C-Se and C-Te bond formation. *Coord. Chem. Rev.* **2021**, *442*, 214012.

(12) For selected reviews, see (a) Ma, Y.-T.; Liu, M.-C.; Zhou, Y.-B.; Wu, H.-Y. Synthesis of Organoselenium Compounds with Elemental Selenium. *Adv. Synth. Catal.* **2021**, *363*, 5386-5406. (b) Guo, T.; Li, Z.; Bi, L.; Fan, L.; Zhang, P. Recent advances in organic synthesis applying elemental selenium. *Tetrahedron* **2022**, *112*, 132752.

(13) (a) Taniguchi, N. Mono- or Dichalcogenation of Aryl Iodide with Sulfur or Selenium by Copper Catalyst and Aluminum. *Synlett* **2005**, *11*, 1687-1690. (b) Luo, D.; Wu, G.; Yang, H.; Liu, M.; Gao, W.; Huang, X.; Chen, J.; Wu, H. Copper-Catalyzed Three-Component Re-

action for Regioselective Aryl- and Heteroarylselenation of Indoles using Selenium Powder. J. Org. Chem. **2016**, *81*, 4485-4493. (c) Gao, C.; Wu, G.; Min, L.; Liu, M.; Gao, W.; Ding, J.; Chen, J.; Huang, X.; Wu, H. Copper-Catalyzed Three-Component Coupling Reaction of Azoles, Se Powder, and Aryl Iodides, J. Org. Chem. **2017**, *82*, 250-255. (d) Sun, P.; Jiang, M.; Wei, W.; Min, Y.; Zhang, W.; Li, W.; Yang, D.; Wang, H. Copper-Catalyzed Selenylation of Imidazo[1,2-a]pyridines with Selenium Powder via a Radical Pathway, J. Org. Chem. **2017**, *82*, 2906-2913. (e) Yadav, D.; Dixit, A. K.; Raghothama, S.; Awasthi, S. K. Ni nanoparticle-confined covalent organic polymer directed diaryl-selenides synthesis. *Dalton Trans.* **2020**, 49, 12266-12272. (f) Cremer, C.; Eltester, M. A.; Bourakhouadar, H.; Atodiresei, I. L.; Patureau, F. W. Dehydrogenative C-H Phenochalcogenazination. Org. Lett. **2021**, *23*, 3243-3247.

(14) (a) Guo, T.; Wei, X.-N.; Zhu, Y.-L.; Chen, H.; Han, S.-L.; Ma, Y.-C. Copper-Catalyzed One-Pot Synthesis of Chalcogen-Benzothiazoles/Imidazo[1,2-*a*]pyridines with Sulfur/Selenium Powder and Aryl Boronic Acids. *Synlett* **2018**, *29*, 1530-1536. (b) Zhang, X.; Huang, X.-B.; Zhou, Y.-B.; Liu, M.-C.; Wu, H.-Y. Metal-Free Synthesis of Aryl Selenocyanates and Selenaheterocycles with Elemental Selenium. *Chem. Eur. J.* **2021**, *27*, 944-948.

(15) (a) Shibahara, F.; Kanai, T.; Yamaguchi, E.; Kamei, A.; Yamauchi, T.; Murai, T. Copper-Catalyzed C-H Bond Direct Chalcogenation of Aromatic Compounds Leading to Diaryl Sulfides, Selenides, and Diselenides by Using Elemental Sulfur and Selenium as Chalcogen Sources Under Oxidative Conditions. Chem. Asian J. 2014, 9, 237-244. (b) Kondo, K.; Matsumura, M.; Kanasaki, K.; Murata, Y.; Kakusawa, N.; Yasuike, S. Synthesis of 2-Aryl-3-(arylselanyl)imidazo[1,2-a]pyridines: Copper-Catalyzed One-Pot, Two-Step Se-Arylation of Selenium with Imidazopyridines and Triarylbismuthanes. Synthesis 2018, 50, 2200-2210. (c) Matsumura, M.; Takahashi, T.; Yamauchi, H.; Sakuma, S.; Hayashi, Y.; Hyodo, T.; Obata, T.; Yamaguchi, K.; Fujiwara, Y.; Yasuike, S. Synthesis and anticancer activity of bis(2arylimidazo[1,2-a]pyridin-3-yl) selenides and diselenides: the coppercatalyzed tandem C-H selenation of 2-arylimidazo[1,2-a]pyridine with selenium. Beilstein J. Org. Chem. 2020, 16, 1075-1083. (d) Guo, T.; Wei, X.-N.; Zhang, M.; Liu, Y.; Zhu, L.-M.; Zhao, Y.-H. Catalyst and additive-free oxidative dual C-H sulfenylation of imidazoheterocycles with elemental sulfur using DMSO as a solvent and an oxidant. Chem. Commun. 2020, 56, 5751-5754.

(16) (a) Moon, S.; Kato, M.; Nishii, Y.; Miura, M. Synthesis of Benzo[b]thiophenes through Rhodium-Catalyzed Three-Component Reaction using Elemental Sulfur. *Adv. Synth. Catal.* **2020**, *362*, 1669-1673. (b) Moon, S.; Nishii, Y.; Miura, M. Synthesis of Isothiazoles and Isoselenazoles through Rhodium-Catalyzed Oxidative Annulation with Elemental Sulfur and Selenium. *Org. Lett.* **2021**, *23*, 49-53. (c) Xu-Xu, Q.-F.; Nishii, Y.; Uetake, Y.; Sakurai, H.; Miura, M. Synthesis of Benzoisoselenazolones via Rh(III)-Catalyzed Direct Annulative Selenation Using Elemental Selenium. *Chem. Eur. J.* **2021**, *27*, 17952-17959.

(17) Wang, D.; Yu, X.; Xu, X.; Ge, B.; Wang, X.; Zhang, Y. Noncoordinating-Anion-Directed Reversal of Activation Site: Selective C-H Bond Activation of N-Aryl Rings. *Chem. Eur. J.* **2016**, *22*, 8663-8668. (18) (a) Hashimoto, Y.; Hirano, K.; Satoh, T.; Kakiuchi, F.; Miura, M. Ruthenium(II)-Catalyzed Regio- and Stereoselective Hydroarylation of Alkynes via Directed C-H Functionalization. *Org. Lett.* **2012**, *14*, 8, 2058-2061. (b) Hashimoto, Y.; Hirano, K.; Satoh, T.; Kakiuchi, F.; Miura, M. Regioselective C-H Bond Cleavage/Alkyne Insertion under Ruthenium Catalysis. *J. Org. Chem.* **2013**, *78*, 638-646.

(19) Zhang, N.; Li, B.; Zhong, H.; Huang, J. Synthesis of *N*-alkyl and *N*-aryl isoquinolones and derivatives *via* Pd-catalysed C–H activation and cyclization reactions. *Org. Biomol. Chem.* **2012**, *10*, 9429-9439.

(20) (a) Xiao, F.; Liu, Y; Tang, C; Deng, G.-J. Peroxide-Mediated Transition-Metal-Free Direct Amidation of Alcohols with Nitroarenes. Org. Lett. 2012, 14, 984-987. (b) Teo, Y.-C.; Yong, F.-F.; Ithnin, I. K.; Yio, S.-H. T.; Lin, Z. Efficient Manganese/Copper Bimetallic Catalyst for N-Arylation of Amides and Sulfonamides Under Mild Conditions in Water. Eur. J. Org. Chem. 2013, 2013, 515-524. (c) Liu, J.; Liu, Q.; Yi, H.; Qin, C.; Bai, R.; Qi, X.; Lan, Y.; Lei, A. Visible-Light-Mediated Decarboxylation/Oxidative Amidation of a-Keto Acids with Amines under Mild Reaction Conditions Using O2. Angew. Chem. Int. Ed. 2014, 53, 502-506. (d) Hong, G.; Wu, S.; Zhu, X.; Mao, D.; Wang, L. Peroxide-mediated direct synthesis of amides from aroyl surrogates. Tetrahedron 2016, 72, 436-441. (e) Mane, R. S.; Bhanage, B. M. Pd/C-Catalyzed Aminocarbonylation of Aryl Iodides via Oxidative C-N Bond Activation of Tertiary Amines to Tertiary Amides. J. Org. Chem. 2016, 81, 1223-1228. (f) Digwal, C. S.; Yadav, U.; Ramya, P. V. S.; Sana, S.; Swain, B.; Kamal, A. Vanadium-Catalyzed Oxidative C(CO)-C(CO) Bond Cleavage for C-N Bond Formation: One-Pot Domino Transformation of 1,2-Diketones and Amidines into Imides and Amides. J. Org. Chem. 2017, 82, 7332-7345. (g) Thurow, S.; Lenardo, E. J.; Just-Baringo, X.; Procter, D. J. Reduction of Selenoamides to Amines Using SmI₂-H₂O. Org. Lett. 2017, 19, 50-53. (h) De, S.; Yin, J.; Ma, D. Copper-Catalyzed Coupling Reaction of (Hetero)Aryl Chlorides and Amides. Org. Lett. 2017, 19, 4864-4867. (i) Mei, C.; Lu, W. Palladium(II)-Catalyzed Oxidative Homo- and Cross-Coupling of Aryl ortho-sp2 C-H Bonds of Anilides at Room Temperatur. J. Org. Chem. 2018, 83, 4812-4823. (j) Peng, J.-B.; Li, D.; Geng, H.-Q.; Wu, X.-F. Palladium-Catalyzed Amide Synthesis via Aminocarbonylation of Arylboronic Acids with Nitroarenes. Org. Lett. 2019, 21, 4878-4881. (k) Karthik, S.; Muthuvel, K.; Gandhi, T. Base-Promoted Amidation and Esterification of Imidazolium Salts via Acyl C-C bond Cleavage: Access to Aromatic Amides and Esters. J. Org. Chem. 2019, 84, 738-751. (1) Ling, L.; Chen, C.; Luo, M.; Zeng, X. Chromium-Catalyzed Activation of Acyl C-O Bonds with Magnesium for Amidation of Esters with Nitroarenes. Org. Lett. 2019, 21, 1912-1916. (m) Ye, P.; Shao, Y.; Ye, X.; Zhang, F.; Li, R.; Sun, J.; Xu, B.; Chen, J. Homoleptic Bis(trimethylsilyl)amides of Yttrium Complexes Catalyzed Hydroboration Reduction of Amides to Amines. Org. Lett. 2020, 22, 1306-1310. (n) Kolekar, Y. A.; Bhanage, B. M. Pd-Catalyzed Oxidative Aminocarbonylation of Arylboronic Acids with Unreactive Tertiary Amines via C-N Bond Activation. J. Org. Chem. 2021, 86, 14028-14035.

(21) Shibata, Y.; Tanaka, K. Catalytic [2+2+1] Cross-Cyclotrimerization of Silylacetylenes and Two Alkynyl Esters To Produce Substituted Silylfulvenes. *Angew. Chem., Int. Ed.* **2011**, *50*, 10917-10921.